

International Comparison EURAMET.QM-K3.2019 – Automotive exhaust gases

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Field

Amount of substance

Subject

Comparison of amount fractions carbon monoxide, carbon dioxide, oxygen and propane in nitrogen (track A – core competences)

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1 Introduction

This key comparison is one of a series of key comparisons in the gas analysis area assessing core competences (*track A key comparisons*). Such competences include, among others, the capabilities to prepare Primary Standard gas Mixtures (PSMs) [1], perform the necessary purity analysis on the materials used in the gas mixture preparation, the verification of the composition of newly prepared PSMs against existing ones, and the capability of calibrating a gas mixture.

For this key comparison, a mixture containing carbon monoxide, carbon dioxide, oxygen, and propane in nitrogen has been chosen which is used for calibration of exhaust automotive gas analysers. The key comparison design follows that of the key comparisons CCQM-K3 [2] and CCQM-K111 [3].

2 Design and organisation of the key comparison

2.1 Participants

Table 1 lists the participants in this key comparison.

Table 1: List of participants

Acronym	Country	Institute
BAM	DE	Bundesanstalt für Materialforschung und -prüfung, Berlin, Germany
CEM	ES	Centro Español de Metrología, Madrid, Spain
LNE	FR	Laboratoire national de métrologie et d'essais, Paris, France
BFKH	HU	Government Office of the Capital City Budapest, BFKH Budapest, Hungary
GUM	PL	Główny Urząd Miar, Warszawa, Poland
SMU	SK	Slovak Institute of Metrology, Bratislava, Slovak Republic
METAS	CH	METAS - Federal Institute of Metrology, Bern - Wabern, Switzerland
UME	TR	TUBITAK National Metrology Institute (UME), Gebze Kocaeli, Turkey
CMI	CZ	Český metrologický institute, Praha, Czech Republic
VSL	NL	Van Swinden Laboratorium, Delft, The Netherlands

2.2 Measurement standards

A set of mixtures was prepared gravimetrically by VSL for the CCQM-K3.2019 key comparison [4]. The same set of standards was used for this EURAMET key comparison. The set of standards was verified before shipment to the participants and after their return.

For the preparation, carbon dioxide grade 5.5 was used from Air Products. Oxygen grade 6.0 was used from Linde. Carbon monoxide grade 4.7 was used from Air Products. Propane grade 3.5 was used from Air Products and nitrogen from Air Products, grade 6.0 BIP+. Carbon dioxide

and oxygen were transferred as pure gas. Carbon monoxide was transferred from a premixture obtained by one gravimetric dilution step from the pure gas. Propane was transferred from a premixture obtained by two subsequent gravimetric dilution steps. The mixtures were verified against a set of VSL PSMs. All used pure gases were subjected to a purity analysis in accordance with ISO 19229 [5] prior to use for preparation of the gas mixtures.

The filling pressure in the cylinders was approximately 11.5 MPa. Aluminium cylinders of 5 dm³ water volume from Luxfer UK with an Aculife IV treatment were used. The mixture composition and its associated uncertainty was calculated in accordance with ISO 6142-1 [1]. The amount fractions as calculated from gravimetry and purity verification of the parent gases were used as key comparison reference values (KCRVs). Each individual cylinder had its own reference values and associated expanded uncertainties. The expanded uncertainties included a contribution from the verification of the gas mixtures.

The nominal ranges of amount fractions of the targeted components in the mixtures are given in Table 2.

Table 2 Nominal composition of mixtures, given in amount fractions

Component	Amount fraction x
Carbon monoxide	0.5 cmol mol ⁻¹ – 2 cmol mol ⁻¹
Carbon dioxide	2 cmol mol ⁻¹ – 5 cmol mol ⁻¹
Oxygen	1 cmol mol ⁻¹ – 4 cmol mol ⁻¹
Propane	100 µmol mol ⁻¹ – 300 µmol mol ⁻¹
Nitrogen	Balance

2.3 Measurement protocol

The measurement protocol requested each laboratory to perform at least 3 measurements, with independent calibrations. The replicates, leading to a submittable measurement value, were to be carried out under repeatability conditions. The protocol informed the participants about the nominal amount fraction ranges. The laboratories were also requested to submit a description of their method and a full description of their uncertainty evaluation used for evaluating the uncertainty of their result.

2.4 Schedule

The schedule of this key comparison was as follows (Table 3).

Table 3: Key comparison schedule

Date	Event
October 2018	Agreement of protocol
October 2018	Registration of participants
January/February 2021	Re-verification of mixture compositions
May 2021	Dispatch of mixtures
August 2021	Reports and cylinders arrived back at VSL
September 2021	Re-verification of the mixtures
January 2022	Draft A report available

2.5 Measurement equation

The key comparison reference values are based on the weighing data, the molar masses of the components and the purity verification of the parent gases. All mixtures underwent verification prior to shipping them to the participants. After return of the cylinders, they have been verified once more to reconfirm the stability of the mixtures.

In the preparation, the following four groups of uncertainty contributions have been considered:

1. gravimetric preparation (weighing process) ($x_{i,\text{grav}}$)
2. purity of the parent gases ($\Delta x_{i,\text{purity}}$)
3. stability of the gas mixture ($\Delta x_{i,\text{stab}}$)
4. correction due to partial recovery of a component ($\Delta x_{i,\text{nr}}$)

Previous experience has indicated that there are no stability issues and no correction is needed for the partial recovery of a component. These terms are zero, and so are their associated standard uncertainties.

The amount fraction $x_{i,\text{prep}}$ of a particular component in mixture i , as it appears during use of the cylinder, can now be expressed as

$$x_{i,\text{prep}} = x_{i,\text{grav}} + \Delta x_{i,\text{purity}}. \quad (1)$$

The equation for calculating the associated standard uncertainty reads as

$$u^2(x_{i,\text{prep}}) = u^2(x_{i,\text{grav}}) + u^2(\Delta x_{i,\text{purity}}). \quad (2)$$

The validity of the mixtures has been demonstrated by verifying the composition as calculated from the preparation data with that obtained from analytical chemical measurement. In order to have a positive demonstration of the preparation data (including uncertainty), the following condition should be met [1]

$$|x_{i,\text{prep}} - x_{i,\text{ver}}| \leq 2 \sqrt{u_{i,\text{prep}}^2 + u_{i,\text{ver}}^2}. \quad (3)$$

The factor 2 is a coverage factor (normal distribution, 95 % level of confidence). The assumption must be made that both preparation and verification are unbiased. Such bias has never been observed. The uncertainty associated with the verification highly depends on the experimental design followed. In this particular key comparison, an approach has been chosen which is consistent with CCQM-K3 [2] and takes advantage of the work done in the gravimetry study CCQM-P41 [6].

The verification experiments have demonstrated that within the uncertainty of these measurements, the gravimetric values of the key comparison mixtures agreed with older measurement standards.

The expression for the standard uncertainty of the key comparison reference value is

$$u^2(x_{i,\text{ref}}) = u^2(x_{i,\text{prep}}) + u^2(x_{i,\text{ver}}) \quad (4)$$

The values for $u_{i,\text{ver}}$ are given in the tables containing the results of this key comparison.

The relative difference from verification is defined as

$$e_{ij} = \frac{x_{\text{anal},ij} - x_{\text{prep},ij}}{x_{\text{prep},ij}} \quad (5)$$

and expressed as percentage. $x_{\text{prep},ij}$ denotes the amount fraction as calculated from static gravimetric preparation for mixture i and verification j , and $x_{\text{anal},ij}$ the amount fraction obtained from the analytical verification. The results of the verification of the mixtures are summarised in Tables 4–7.

The uncertainty of the differences is dominated by the precision of the measurements, both within a run as well as between runs. Hence, the differences have been treated as uncorrelated. For each of the components and each of the mixtures, a difference has been calculated from the verification data before shipment and after return of the mixtures. The differences e_{ij} have been modelled using a random effects model

$$e_{ij} = \mu_i + B_i + \varepsilon_{ij} \quad (6)$$

where μ_i denotes the mean relative difference for mixture i , B_i a reproducibility effect in the verification measurements and ε_{ij} a random measurement error term. Both B_i and ε_{ij} are assumed to be normally distributed with zero mean and variances τ_i^2 and σ_i^2 , respectively.

The mean difference was formed by using restricted maximum likelihood estimation (REML) to the observed differences. Under the assumptions of the random effects model, the log likelihood function takes the form [7]

$$\log L(\mu, \tau^2) = -\frac{1}{2} \sum_j \log \left(2\pi(\sigma_j^2 + \tau^2) \right) - \frac{1}{2} \sum_j \log \frac{(e_j - \mu)^2}{(\sigma_j^2 + \tau^2)} \quad (7)$$

where the index i has been dropped. To obtain the estimates for the model parameters, the log likelihood function is maximised. The restricted maximum likelihood estimator is used, as this estimator is generally preferred over the maximum likelihood estimator in case of small data sets (thus, a small number of degrees of freedom).

The calculations have been performed using the metafor package [8] and R [9]. The standard uncertainty of the difference was calculated using the procedures of ISO 6143 [10] (in the case of a multipoint calibration) and ISO 12963 [11] (in the case of a single point calibration), followed by applying the law of propagation of uncertainty from the GUM (Guide to the expression of uncertainty in measurement) [12] to obtain the standard uncertainty associated with the relative difference.

Table 4 shows the verification results of the mixtures for carbon dioxide. The values for the excess standard deviation (τ) vary appreciably which is partly due to the low number of degrees of freedom on which they are based. The within-group standard uncertainty (σ) is very similar for each of the mixtures. All mixtures save D340045 and 8451 E pass on the basis of this calculation the verification criterion of ISO 6143 (see equation (3)). For the verification uncertainty in this key comparison, the pooled standard uncertainty of all verifications is used, which is 0.032 %. With this standard uncertainty, also D340045 and 8451 E meets the criterion of ISO 6143.

Table 5 summarises the verification results of the mixtures for carbon monoxide. All mixtures save 8449 E pass, based on the same calculations as used for carbon dioxide, the verification criterion of ISO 6143. As in the case of carbon dioxide, the values of the between-group standard deviation τ are appreciably different. For the verification uncertainty, the pooled standard uncertainty of all verifications is used, which is 0.038 %. With this standard uncertainty, all but one of the mixtures pass the criterion of ISO 6143. For mixture 8449 E, the standard uncertainty is set at 0.050 %. The deviation seen in the verification is larger than for the other mixtures. The larger verification uncertainty has no adverse effect on the evaluation of the result of the NMI receiving this mixture.

Table 4: Verification results for carbon dioxide, expressed as relative differences with respect to the amount fractions as calculated from preparation (%)

Mixture	e_1	$u(e_1)$	e_2	$u(e_2)$	e_3	$u(e_3)$	e_4	$u(e_4)$	e_5	$u(e_5)$	τ	σ	\bar{e}	$u(\bar{e})$
ML 6817	0.00	0.03	0.00	0.05	0.02	0.01	-0.04	0.02	-0.01	0.02	0.02	0.03	-0.004	0.013
8500 E	-0.06	0.03	-0.10	0.04	0.00	0.01	0.08	0.01	0.00	0.01	0.06	0.02	-0.010	0.029
MR 8468	0.09	0.02	-0.06	0.06	-0.08	0.01	-0.03	0.01	0.02	0.01	0.07	0.03	-0.009	0.031
D751979	0.03	0.03	0.07	0.05	-0.07	0.02	-0.02	0.02	-0.03	0.02	0.04	0.03	-0.016	0.020
ML 6812	-0.03	0.04	-0.18	0.06	0.01	0.01	0.00	0.03	0.16	0.03	0.11	0.04	-0.001	0.051
D340045	0.05	0.03	0.10	0.04	0.09	0.02	-0.01	0.02	0.05	0.02	0.04	0.03	0.052	0.020
8449 E	-0.05	0.04	-0.05	0.04	0.04	0.02	-0.02	0.01	-0.18	0.01	0.08	0.03	-0.053	0.038
8451 E	-0.07	0.04	0.03	0.05	0.05	0.01	0.02	0.01	0.04	0.01	0.02	0.03	0.031	0.011
8457 E	0.11	0.03	-0.10	0.05	-0.04	0.01	-0.04	0.03	-0.15	0.03	0.09	0.03	-0.043	0.044

Table 5: Verification results for carbon monoxide, expressed as relative differences with respect to the amount fractions as calculated from preparation (%)

Mixture	e_1	$u(e_1)$	e_2	$u(e_2)$	e_3	$u(e_3)$	e_4	$u(e_4)$	τ	σ	\bar{e}	$u(\bar{e})$
ML 6817	0.02	0.02	0.00	0.03	0.01	0.01	0.01	0.02	0.00	0.02	0.011	0.008
8500 E	-0.12	0.02	-0.16	0.02	0.03	0.01	0.01	0.02	0.09	0.02	-0.059	0.047
MR 8468	0.01	0.02	-0.07	0.03	-0.04	0.01	-0.10	0.03	0.04	0.02	-0.046	0.023
D751979	-0.03	0.02	0.01	0.02	0.01	0.01	0.16	0.02	0.08	0.02	0.038	0.042
ML 6812	-0.02	0.03	-0.18	0.02	0.01	0.03	0.05	0.04	0.10	0.03	-0.037	0.051
D340045	-0.11	0.02	-0.07	0.02	0.02	0.01	0.01	0.02	0.06	0.02	-0.037	0.032
8449 E	-0.12	0.02	-0.11	0.03	0.03	0.02	-0.14	0.02	0.08	0.02	-0.084	0.040
8451 E	-0.10	0.02	-0.06	0.02	-0.05	0.01	0.07	0.02	0.07	0.02	-0.034	0.037
8457 E	-0.02	0.03	-0.15	0.02	0.01	0.02	-0.23	0.03	0.11	0.02	-0.097	0.056

Table 6: Verification results for propane, expressed as relative differences with respect to the amount fractions as calculated from preparation (%)

Mixture	e_1	$u(e_1)$	e_2	$u(e_2)$	e_3	$u(e_3)$	τ	σ	\bar{e}	$u(\bar{e})$
ML 6817	-0.06	0.03	0.06	0.02	0.00	0.01	0.05	0.02	0.004	0.033
8500 E	-0.01	0.02	0.02	0.02	-0.01	0.02	0.00	0.02	-0.003	0.011
MR 8468	-0.04	0.02	-0.04	0.02	0.00	0.02	0.01	0.02	-0.026	0.013
D751979	0.01	0.02	-0.02	0.02	-0.05	0.02	0.02	0.02	-0.020	0.017
ML 6812	-0.07	0.02	0.14	0.02	0.00	0.02	0.11	0.02	0.024	0.062
D340045	-0.04	0.02	0.01	0.02	0.04	0.02	0.03	0.02	0.005	0.023
8449 E	-0.03	0.03	0.02	0.02	0.03	0.02	0.02	0.02	0.010	0.018
8451 E	-0.08	0.02	0.03	0.02	0.03	0.01	0.06	0.02	-0.006	0.036
8457 E	-0.05	0.02	-0.02	0.02	-0.03	0.01	0.00	0.02	-0.031	0.010

Table 7: Verification results for oxygen, expressed as relative differences with respect to the amount fractions as calculated from preparation (%)

Mixture	e_1	$u(e_1)$	e_2	$u(e_2)$	e_3	$u(e_3)$	τ	σ	\bar{e}	$u(\bar{e})$
ML 6817	0.01	0.02	-0.04	0.02	0.09	0.02	0.06	0.02	0.020	0.038
8500 E	-0.03	0.02	-0.02	0.03	-0.27	0.02	0.14	0.02	-0.107	0.082
MR 8468	-0.05	0.03	-0.01	0.02	0.04	0.03	0.04	0.03	-0.008	0.025
D751979	-0.04	0.02	-0.02	0.02	0.13	0.02	0.09	0.02	0.024	0.054
ML 6812	0.04	0.03	-0.02	0.02	0.14	0.03	0.08	0.03	0.051	0.047
D340045	-0.01	0.02	0.02	0.02	-0.05	0.02	0.03	0.02	-0.014	0.020
8449 E	0.03	0.02	0.05	0.02	-0.04	0.02	0.04	0.02	0.014	0.027
8451 E	0.05	0.02	0.00	0.02	-0.04	0.02	0.04	0.02	0.002	0.025
8457 E	0.03	0.02	-0.07	0.02	-0.05	0.02	0.05	0.02	-0.032	0.029

Table 6 summarises the verification results of the mixtures for propane. All mixtures save 8457 E pass, based on the same calculations as used for carbon dioxide, the verification criterion of ISO 6143. As in the case of carbon dioxide, the values of the between-group standard deviation τ are appreciably different. For the verification uncertainty, the pooled standard uncertainty of all verifications is used, which is 0.028 %. With this standard uncertainty, all mixtures pass the criterion of ISO 6143.

Table 7 summarises the verification results of the mixtures for oxygen. All mixtures pass, based on the same calculations as used for carbon dioxide, the verification criterion of ISO 6143. As in the case of carbon dioxide, the values of the between-group standard deviation τ are appreciably different. For the verification uncertainty, the pooled standard uncertainty of all verifications is used, which is 0.041 %. With this standard uncertainty, all mixtures pass the criterion of ISO 6143.

2.6 Measurement methods

The measurement methods used by the participants are described in the measurement reports annexed to this report. A summary of the calibration methods, dates of measurement and reporting, and the way in which metrological traceability is established is given in Table 8.

Table 8: Summary of calibration methods and metrological traceability

Laboratory	Measurement period	Calibration	Traceability	Matrix standards	Measurement technique
BAM	1 July to 12 August 2021	two-point calibration with bracketing (TPC) protocol	Own standards (ISO 6142)	Nitrogen	GC with TCD and methanizer/FID
CEM	16 July to 17 August 2021	ISO 6143	Own standards (ISO 6142)	Nitrogen	micro-GC-TCD
LNE	16 August to 24 August 2021	Single-point calibration	Own standards (ISO 6142)	Nitrogen	PDHID detector (O ₂) and a FID with methanizer (CO, CO ₂ and C ₃ H ₈)
BFKH	19 July to 26 August 2021	Single-point calibration	Own standards	Nitrogen	GC-FID and GC-TCD
GUM	28 June to 4 August 2021	ISO 6143	Own standards (ISO 6142)	Nitrogen	GC-FID and GC-TCD
SMU	28 June to 27 July 2021	ISO 6143	Own standards (ISO 6142)	Nitrogen	GC-FID (C ₃ H ₈) and GC-TCD (CO, CO ₂ & O ₂)
METAS	2 June to 6 September 2021	ISO 6143	Own standards and NPL & VSL standards	Nitrogen	GC-FID with methanizer (CO, CO ₂ & C ₃ H ₈) and paramagnetic measurement system (O ₂)
UME	2 July to 9 September 2021	ISO 6143	Own standards (ISO 6142)	Nitrogen	GC-FID (CO, CO ₂ and C ₃ H ₈) and GC-TCD (O ₂)
CMI	5 July to 16 August 2021	Four-point linear model	Own standards (ISO 6142)	Nitrogen	GC with 2x TCD and FID

2.7 Degrees of equivalence

A unilateral degree of equivalence in key comparisons is defined as

$$\Delta x_i = d_i = x_i - x_{i,\text{KCRV}}, \quad (8)$$

and the uncertainty of the difference d_i at 95 % level of confidence. Here, $x_{i,\text{ref}}$ denotes the key comparison reference value, and x_i the result of laboratory i .¹ Appreciating the special conditions in gas analysis, it can be expressed as

$$\Delta x_i = d_i = x_i - x_{i,\text{ref}}. \quad (9)$$

The standard uncertainty of d_i can be expressed as

$$u^2(d_i) = u^2(x_i) + u^2(x_{i,\text{prep}}) + u^2(\Delta x_{i,\text{ver}}), \quad (10)$$

assuming that the aggregated error terms are uncorrelated. As discussed, the combined standard uncertainty of the reference value comprises that from preparation and that from verification for the mixture involved.

3 Results

In this section, the results of the key comparison are summarised. In the tables, the following data are presented

x_{prep}	amount fraction, from preparation (cmol/mol)
u_{prep}	standard uncertainty of x_{prep} (cmol/mol)
u_{ver}	standard uncertainty from verification (cmol/mol)
u_{ref}	standard uncertainty of reference value (cmol/mol)
x_{lab}	result of laboratory (cmol/mol)
U_{lab}	stated uncertainty of laboratory, at 95 % level of confidence (cmol/mol)
k_{lab}	stated coverage factor
d_i	difference between laboratory result and reference value (cmol/mol)
k	assigned coverage factor for degree of equivalence
$U(d_i)$	Expanded uncertainty of difference d_i , at 95 % level of confidence ² (cmol/mol)

Table 9: Results for the amount fraction carbon dioxide

Laboratory	Cylinder	x_{prep} (cmol/mol)	u_{prep} (cmol/mol)	u_{ver} (cmol/mol)	u_{ref} (cmol/mol)	x_{lab} (cmol/mol)	U_{lab} (cmol/mol)	k_{lab}	d_i (cmol/mol)	k	$U(d_i)$ (cmol/mol)
BAM	ML 6817	1.9951	0.0004	0.0006	0.0008	1.9963	0.0060	2	0.0012	2	0.0062
CEM	8500 E	1.9998	0.0004	0.0006	0.0008	2.016	0.013	2	0.0162	2	0.0131
LNE	8457 E	2.0138	0.0004	0.0006	0.0008	2.007	0.017	2	-0.0068	2	0.0171
BFKH	8451 E	1.9846	0.0004	0.0006	0.0008	1.9831	0.0018	2	-0.0015	2	0.0023
GUM	D340045	1.9965	0.0004	0.0006	0.0008	1.9947	0.0070	2	-0.0018	2	0.0072
SMU	8449 E	2.0025	0.0004	0.0006	0.0008	2.0027	0.0060	2	0.0002	2	0.0062
METAS	MR 8468	2.0018	0.0004	0.0006	0.0008	2.005	0.007	2	0.0032	2	0.0072
UME	D751979	1.9731	0.0004	0.0006	0.0007	1.9752	0.0038	2	0.0021	2	0.0041
CMI	ML 6812	1.9993	0.0004	0.0006	0.0008	2.0002	0.0051	2	0.0009	2	0.0053

¹ Each laboratory receives one cylinder, so that the same index can be used for both a laboratory and a cylinder.

² As defined in the MRA [8], a degree of equivalence is given by d and $U(d)$.

Table 9 shows the results for the amount fraction carbon dioxide. All results but one (CEM) are consistent with the key comparison reference value. The degrees of equivalence are plotted in Figure 1.

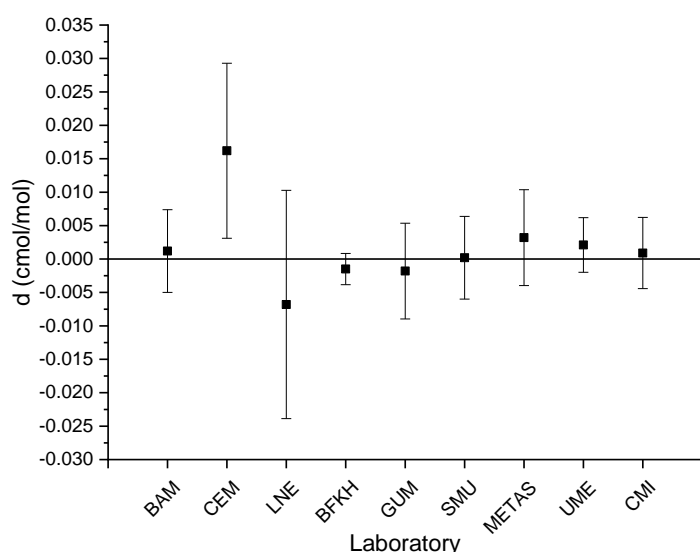


Figure 1: Degrees of equivalence for the amount fraction carbon dioxide

Table 10 shows the results for the amount fraction carbon monoxide. All results, save that of METAS are consistent with the key comparison reference value. The degrees of equivalence are plotted in Figure 2.

Table 10: Results for the amount fraction carbon monoxide

Laboratory	Cylinder	x_{prep} (cmol/mol)	u_{prep} (cmol/mol)	u_{ver} (cmol/mol)	u_{ref} (cmol/mol)	x_{lab} (cmol/mol)	U_{lab} (cmol/mol)	k_{lab}	d_i (cmol/mol)	k	$U(d_i)$ (cmol/mol)
BAM	ML 6817	1.0135	0.0002	0.0004	0.0004	1.0145	0.0030	2	0.0010	2	0.0031
CEM	8500 E	0.9939	0.0002	0.0004	0.0004	0.9962	0.0046	2	0.0023	2	0.0047
LNE	8457 E	1.0162	0.0002	0.0004	0.0004	1.0130	0.0080	2	-0.0032	2	0.0080
BFKH	8451 E	1.0026	0.0002	0.0004	0.0004	1.0021	0.0017	2	-0.0005	2	0.0019
GUM	D340045	1.0010	0.0002	0.0004	0.0004	1.0032	0.0040	2	0.0022	2	0.0041
SMU	8449 E	1.0007	0.0002	0.0004	0.0004	1.0008	0.0030	2	0.0001	2	0.0031
METAS	MR 8468	1.0033	0.0002	0.0005	0.0005	1.0087	0.0040	2	0.0054	2	0.0041
UME	D751979	1.0100	0.0002	0.0004	0.0004	1.0118	0.0020	2	0.0018	2	0.0022
CMI	ML 6812	1.0052	0.0002	0.0004	0.0004	1.0024	0.0052	2	-0.0028	2	0.0053

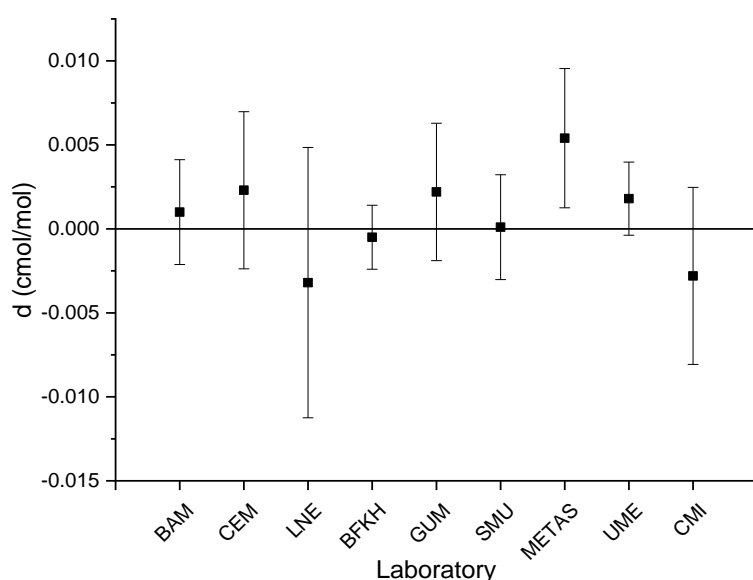


Figure 2: Degrees of equivalence for the amount fraction carbon monoxide

In Table 11 the results for the amount fraction propane are shown. All results, save those from BFKH and UME are consistent with the key comparison reference value. The degrees of equivalence are plotted in Figure 3.

Table 11: Results for the amount fraction propane

Laboratory	Cylinder	x_{prep} (cmol/mol)	u_{prep} (cmol/mol)	u_{ver} (cmol/mol)	u_{ref} (cmol/mol)	x_{lab} (cmol/mol)	u_{lab} (cmol/mol)	k_{lab}	d_i (cmol/mol)	k	$U(d_i)$ (cmol/mol)
BAM	ML 6817	0.019981	0.000005	0.000007	0.000009	0.019976	0.000080	2	-0.000005	2	0.000082
CEM	8500 E	0.020116	0.000005	0.000007	0.000009	0.02001	0.00029	2	-0.000106	2	0.000291
LNE	8457 E	0.020327	0.000005	0.000007	0.000009	0.02027	0.00013	2	-0.000057	2	0.000131
BFKH	8451 E	0.020017	0.000005	0.000007	0.000009	0.020052	0.000028	2	0.000035	2	0.000033
GUM	D340045	0.019986	0.000005	0.000007	0.000009	0.019968	0.000078	2	-0.000018	2	0.000080
SMU	8449 E	0.019868	0.000005	0.000007	0.000009	0.019906	0.000061	2	0.000038	2	0.000063
METAS	MR 8468	0.019954	0.000005	0.000007	0.000009	0.01995	0.00008	2	-0.000004	2	0.000082
UME	D751979	0.019665	0.000005	0.000007	0.000009	0.01973	0.00004	2	0.000065	2	0.000044
CMI	ML 6812	0.020163	0.000005	0.000007	0.000009	0.02015	0.00044	2	-0.000013	2	0.000440

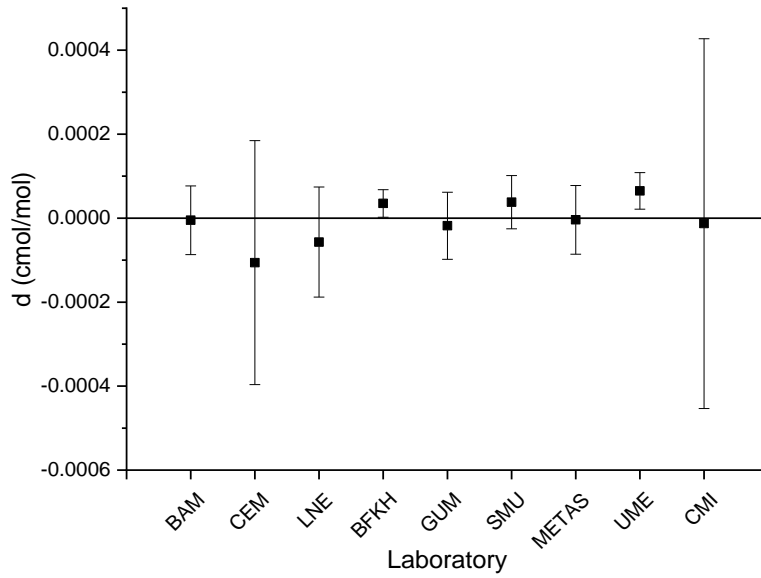


Figure 3: Degrees of equivalence for the amount fraction propane

Table 12 shows the results for the amount fraction oxygen. The result from BFKH is discrepant with respect to the key comparison reference value. The degrees of equivalence are plotted in Figure 4.

Table 12: Results for the amount fraction oxygen

Laboratory	Cylinder	x_{prep} (cmol/mol)	u_{prep} (cmol/mol)	u_{ver} (cmol/mol)	u_{ref} (cmol/mol)	x_{lab} (cmol/mol)	U_{lab} (cmol/mol)	k_{lab}	d_i (cmol/mol)	k	$U(d_i)$ (cmol/mol)
BAM	ML 6817	2.9698	0.0006	0.0012	0.0014	2.9733	0.0089	2	0.0035	2	0.0093
CEM	8500 E	3.0137	0.0006	0.0012	0.0014	3.028	0.021	2	0.0143	2	0.0212
LNE	8457 E	2.8924	0.0006	0.0012	0.0013	2.918	0.027	2	0.0256	2	0.0271
BFKH	8451 E	3.0080	0.0006	0.0012	0.0014	3.0031	0.0028	2	-0.0049	2	0.0039
GUM	D340045	3.0016	0.0006	0.0012	0.0014	3.0037	0.0098	2	0.0021	2	0.0102
SMU	8449 E	3.0116	0.0006	0.0012	0.0014	3.011	0.013	2	-0.0006	2	0.0133
METAS	MR 8468	2.9833	0.0006	0.0012	0.0014	2.972	0.012	2	-0.0113	2	0.0123
UME	D751979	3.0649	0.0006	0.0013	0.0014	3.0641	0.0038	2	-0.0008	2	0.0047
CMI	ML 6812	2.9606	0.0006	0.0012	0.0014	2.9619	0.0099	2	0.0013	2	0.0103

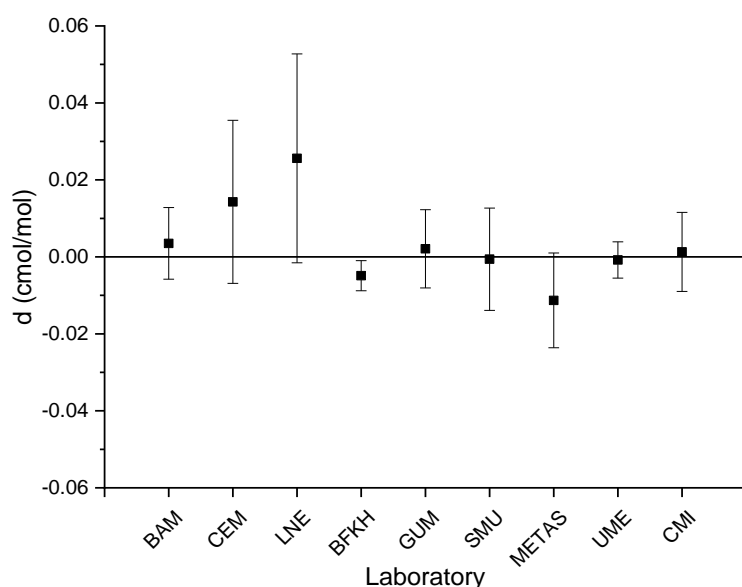


Figure 4: Degrees of equivalence for the amount fraction oxygen

4 Supported CMC claims

The results of this key comparison can be used to support CMC claims in two different ways, as detailed in the report of CCQM-K3.2019 [4].

5 Discussion and conclusions

The results in this key comparison show generally good agreement. Most results agree within 0.3 % relative of the key comparison reference value. The deviations and the stated expanded uncertainties are generally larger than for the corresponding binary mixtures in previous key comparisons (e.g., CCQM-K120 Carbon dioxide, CCQM-K111 Propane, CCQM-K51 Carbon monoxide).

References

- [1] International Organization for Standardization, ISO 6142-1:2015 Gas analysis — Preparation of calibration gas mixtures — Part 1: Gravimetric method for Class I mixtures, 1st edition
- [2] Van der Veen A.M.H, De Leer E.W.B., Perrochet J.-F., Wang Lin Zhen, Heine H.-J., Knopf D., Richter W., Barbe J., Marschal A., Vargha G., Deák E., Takahashi C., Kim J.S., Kim Y.D., Kim B.M., Kustikov Y.A., Khatskevitch E.A., Pankratov V.V., Popova T.A., Konopelko L., Musil S., Holland P., Milton M.J.T., Miller W.R., Guenther F.R., International Comparison CCQM-K3, Final Report, 2000
- [3] Van der Veen A.M.H., Van der Hout J.W., Ziel P.R., Oudwater R.J., Fioravante A.L., Augusto C.R., Brum M.C., Uehara S., Akima D., Bae H.K., Kang N., Woo J.C., Liaskos C.E., Roderick G.C., Brewer P.H., Brown A.S., Bartlett S., Downey M.L., Konopelko L.A., Kolobova A.V., Pankov A.A., Orshanskaya A.A., Efremova O.V., “International Comparison CCQM-K111 – Propane in nitrogen”, Final Report, Metrologia Techn. Suppl. 54 (2017), 08009
- [4] Van der Veen A.M.H., Zalewska E., Kobayashi M., Akima D., Uehara S., Fioravante A.L., Augusto C.R., Ribeiro C.C., Silva V., Dias F., Botas A., Costa C., Lee J.S., Lee J.B., Lim J., Bae H., Kang N., Cecelski C.E., Harris K.J., Miller Jr. W.R., Carney J., Tshilongo J., Ntsasa N.G., Jozela M.I., Leshabane N., Marebane P.M., Worton D.R., Mussell Webber E.B., Moreno S., Brewer

P.J., Konopelko L.A., Kolobova A.V., Meshkov A.V., Efremova O.V., “International Comparison CCQM-K3.2019 Automotive exhaust gases”, Final Report, Metrologia Techn. Suppl., to be published

- [5] International Organization for Standardization, ISO 19229:2019 Gas analysis — Gas analysis — Purity analysis and the treatment of purity data, 2nd edition
- [6] Van der Veen A.M.H., Brinkmann F.N.C., Arnautovic M., Besley L., Heine H.-J., Lopez Esteban T., Sega M., Kato K., Kim J.S., Perez Castorena A., Rakowska A., Milton M.J.T., Guenther F.R., Francy R., Dlugokencky E., “International comparison CCQM-P41 Greenhouse gases. 2. Direct comparison of primary standard gas mixtures”, Metrologia 44 (2007), Techn. Suppl. 08003
- [7] Brockwell S.E., Gordon I.R., “A comparison of statistical methods for meta-analysis”, Statistics in Medicine 20 (2001), pp. 825-840
- [8] Viechtbauer, W., Conducting meta-analyses in R with the metafor package. Journal of Statistical Software, 36(3) (2010), 1–48
- [9] R Core Team (2020). R: A language and environment for statistical computing. R Foundation for Statistical Computing, Vienna, Austria. URL <https://www.R-project.org/>
- [10] International Organization for Standardization, ISO 6143:2001 Gas analysis — Comparison methods for determining and checking the composition of calibration gas mixtures, 2nd edition
- [11] International Organization for Standardization, ISO 12963:2017 Gas analysis — Comparison methods for the determination of the composition of gas mixtures based on one- and two-point calibration
- [12] International Organization for Standardization, ISO/IEC Guide 98-3:2008 Uncertainty of measurement — Part 3: Guide to the expression of uncertainty in measurement (GUM:1995), 1st edition

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Annex A Measurement report of BAM

A.1 Background

EURAMET.QM-K3.2019 is an international interlaboratory comparison for automotive gases organized by VSL. The study covers an O₂ containing automotive gas. This report provides the results of BAM as a participant.

A.2 Choice of method

For the determination of the components the method of choice was GC with TCD or FID detection and optionally methanizer/FID for the determination of CO and CO₂.

A.3 Sample: labeling, packing, pre-information

The sample was provided by VSL in a 5-L aluminum cylinder with cylinder number ML 6817 equipped with a valve which has DIN 477 thread No. 1 (hydrogen). The initial pressure upon receipt was not measured, the final pressure after finishing the measurement campaign was 6.4 MPa.

A.4 Sample pretreatment

No heating or rolling was applied. The sample was stored at indoor laboratory conditions.

A.5 Devices used and flushing

A DIN 477 No. 1 (hydrogen)-VRC ¼" fitting was adapted to the sample cylinder. A reduction valve, a needle valve for dosing, and a closing valve with an outlet to Swagelok 1/16" capillaries were attached. For the two simultaneously employed calibration gases a similar assembly was used but with a DIN 477 No. 14 (test gas) thread. The assembly for both the sample cylinder and the calibration gases were frequently connected and disconnected to the cylinder valve when it was mounted to the two GC analyzers used in this measuring campaign.

A freshly installed assembly was evacuated down to a pressure of approximately 10⁻³ mbar and then filled with gas from the cylinder prior to the actual analysis. The evacuating/flushing procedure was repeated five times before the analysis was started.

A.6 Measurement instruments & settings

A specially designed Siemens Maxum II process gas analyzer was used applying "method 1" (an established standard method to certify gas samples) which executes a sequence of 44 consecutive injections. The instrument is equipped with TCD detectors and has six different separation channels that are adapted for assigned analytes. All installed columns are custom-fit packed columns. The oven was set to 60 °C and operated in isothermal mode with helium as carrier gas for five of the six channels. Injection volumes vary between 30 and 500 µL depending on the channel. Details on Maxum operation can be found in the BAM SOP "GAS-StAA-027".

Channel 1 is for helium and hydrogen with N₂ as a carrier gas and two columns of type Hayesep (guard) and Molsieve (separation) columns.

Channel 2 is for methane with two Hayesep (guard and separation) columns.

Channel 3 is for methane, carbon dioxide, ethene, and ethane with two Hayesep (guard and separation) columns.

Channel 4 is for propene and propane with a Sorbitolchrom (guard), a TCEP-Carboblack (guard), and a TCEP-Unibeads (separation) column.

Channel 5 is for oxygen, nitrogen, and carbon monoxide with Shincarbon (guard) and Molsieve (separation) columns.

Channel 6 is for isobutane, n-butane, neopentane, isopentane, n-pentane, and n-hexane with two Wax-20M-Carboblack (guard and separation) columns.

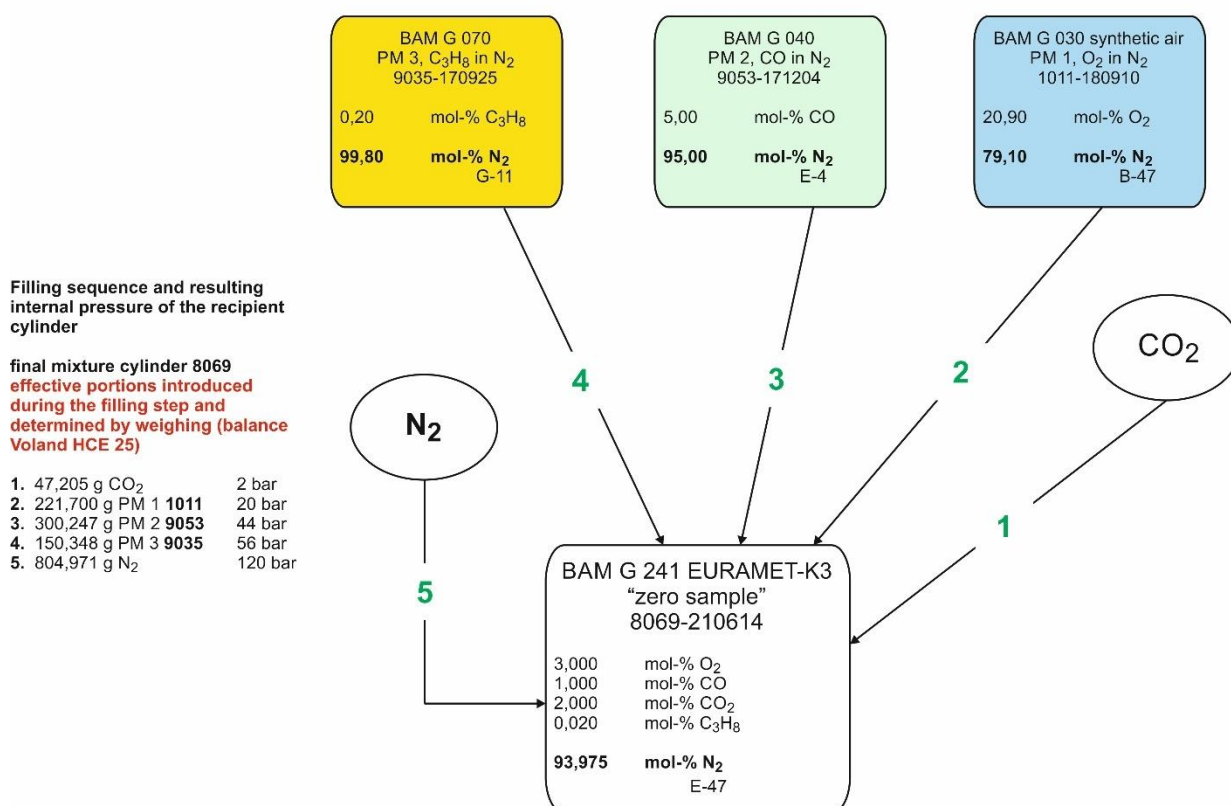
Additionally, a Thermo Scientific Trace 1310 analytical GC was used. It is equipped with two channels, namely a channel for permanent gases with backflush on a Restek HayesepQ and separation on a Restek Molsieve micro-packed column with TCD detection and channel for hydrocarbons with a Restek Rtx-1® capillary column for hydrocarbons and a methanizer/FID detector. A valve control circuit allows for building methods with a switch protocol.

Oxygen was determined on the channel for permanent gases. CO₂ and CO were visible there as well but directed on the hydrocarbon channel with a methanizer and measured together with the propane. The method operated in an isothermal mode at 80 °C.

A.7 Calibration

Three calibration gases were prepared independently from high-purity gases by a gravimetric method according to DIN EN ISO 6142-1:2015 as shown in the following general scheme:

filling scheme BAM gas G 241 EURAMET-K3 “zero sample”



Filling steps and resulting pressure in the recipient cylinder (calculated amounts and pressures to create the targeted composition; the pressures are calculated for room temperature):

1, 2, 3, ... indicates a filling step with pure gases or gas mixtures, respectively.

Gas cylinders (with individual component masses > 30 g) were weighed on a balance of type Voland HCE 25 with an uncertainty of $U(m) = 15$ mg (Type B).

Premixture 1 (PM 1; cylinder no. 1011)

Premixture 2 (PM 2; cylinder no. 9053)

388,439 g O₂
 1287,088 g N₂
 Total: 1675,527 g

81,191 g CO
 1542,096 g N₂
 Total: 1623,286 g

Premixture 3a (4 mol-% propane in N₂,
 cylinder no. 8005, not shown in the scheme)
 90,684 g C₃H₈
 1381,160 g N₂
 Total: 1471,844 g

Premixture 3 (PM 3; cylinder no. 9035)
 80,178 g PM 3a
 1490,110 g N₂
 Total: 1570,288 g

The composition (from gravimetry) of the calibration gases was:

Cylinder no.		Oxygen	Carbon-monoxide	Carbon-dioxide	Propane
8036-210614	$x_{\text{zert}} / \%$	2.854457	0.950082	1.898046	0.019002
	$u(x_{\text{zert}}) / \%$	0.000114	0.000095	0.000313	0.000005
	$u_{\text{rel}}(x_{\text{zert}})$	4.0E-05	1.0E-04	1.7E-04	2.7E-04
	$U_{\text{rel}}(x_{\text{zert}})$	8.0E-05	2.0E-04	3.3E-04	5.4E-04
8092-210614	$x_{\text{zert}} / \%$	3.149334	1.049895	2.098520	0.021012
	$u(x_{\text{zert}}) / \%$	0.000126	0.000094	0.000315	0.000006
	$u_{\text{rel}}(x_{\text{zert}})$	4.0E-05	9.0E-05	1.5E-04	2.9E-04
	$U_{\text{rel}}(x_{\text{zert}})$	8.0E-05	1.8E-04	3.0E-04	5.7E-04
8069-210614	$x_{\text{Hersteller}} / \%$	2.998175	1.000759	2.002137	0.020014
	$u(x_{\text{Hersteller}}) / \%$	0.000120	0.000095	0.000320	0.000005
	$u_{\text{rel}}(x_{\text{Hersteller}})$	4.0E-05	9.5E-05	1.6E-04	2.7E-04
	$U_{\text{rel}}(x_{\text{Hersteller}})$	8.0E-05	1.9E-04	3.2E-04	5.4E-04

In a measurement campaign prior to the investigation of the sample to validate the prepared mixtures, the calibration gases (two “bracket gases” 8036-210614 and 8092-210614 and a “zero” sample 8069-210614 that resembles the targeted composition) were found to be consistent [for details see file VAL-G241].

The purity analysis of the initial pure gases was based on the information provided by the supplier or on the results of determination of impurities in pure gases using an analysis method developed at BAM. As the analyte concentrations in the automotive gas are rather high, blanks of the initial gases were not significantly contributing to values and uncertainty.

A.8 Measurement outline

For a measurement sequence, the two calibration gases (C1 = 8036 & C2 = 8092) and the sample (S = 8069 or ML 6817) were connected to the GC. Using a stream selector valve, each calibration gas was sampled three times, the sample gas five times to the GC in the following consecutive order: C1–S–C2–S–C1–S–C2–S–C1–S–C2. When the stream is flowing through the sampling loop, four injections were made from each stream before the stream selector valve switched to the next stream in the sequence. The first injection of each stream was discarded, only the last three were used for data evaluation. The complete sequence runs over a period of four hours with the Maxum GC and of 22 hours with the Thermo GC, respectively. Using the Maxum GC, one measurement campaign consisting of three sequences was conducted. Using the Thermo GC, two measurement campaigns were conducted. Each campaign was

executed on different days and the assembly was flushed and mounted (dismounted) to the analyzer at the start.

The calibration follows the two-point calibration with bracketing (TPC) protocol as basically described in ISO 12963. The practical implementation in this study is briefly described and exemplarily illustrated below with numbers for an arbitrary measurement of propane {Code783_EURAMET_K3_Thermo_5, Ausw_MAX_1, 4. Kurvenwert}. All other data (stored in EXCEL spreadsheets) are also available upon request.

From the three recorded measurements for each gas, the mean value (MW) and standard deviation (Stdev) (understood as uncertainty of type A) and relative standard deviation (urel) are calculated for the observed signal intensities. In this case, the signal refers to the integrated peak area of propane.

3,2	MW	2.1958
	Stdev	0.0018
	urel	8.2E-04
2,4	MW	2.0870
	Stdev	0.0011
	urel	5.1E-04
1,3	MW	1.9880
	Stdev	0.0011
	urel	5.7E-04

Together with the gas composition (from gravimetry) with uncertainties taken as type B of the calibration gases, i.e.,

C1 8036- 210614	$x_{\text{zert}} / \%$	0.019002
	$u(x_{\text{zert}}) / \%$	0.000005
	$u_{\text{rel}}(x_{\text{zert}})$	2.7E-04
C2 8092- 210614	$x_{\text{zert}} / \%$	0.021012
	$u(x_{\text{zert}}) / \%$	0.000006
	$u_{\text{rel}}(x_{\text{zert}})$	2.9E-04

a bracketing calculation is performed to obtain the amount-of-substance fraction (in our evaluation worksheet denoted as “Kurvenwert”) for the sample S, the corresponding uncertainty is calculated according to the general rules of uncertainty propagation. The obtained values are:

m	106.7
b	0.0
Kurvenwert / %	0.019947
$u(\text{Kurvenwert}) / \%$	0.000014
$u_{\text{rel}}(\text{Kurvenwert})$	7.1E-04
(X3-X1)	0.0020
(Y3-Y1)	0
(Y3-Y2)	0
(Y2-Y1)	0
$(dx_2/dY_2)^2 \cdot u(Y_2)^2$	1.30E-10
	64%

$(dX2/dY3)^2 \cdot u(Y3)^2$	4.54E-11
	23%
$(dX2/dY1)^2 \cdot u(Y1)^2$	1.13E-11
	6%
$(dX2/dX1)^2 \cdot u(X1)^2$	7.39E-12
	4%
$(dX2/dX3)^2 \cdot u(X3)^2$	7.93E-12
	4%
sum	2.02E-10

As can be seen from these numbers, the uncertainty is dominated from the measurement statistics of the three gases (i.e., 64 % + 23 % + 6 %), the proportion of the three values vary about randomly between the five different sets in a measurement campaign. The uncertainty from the calibration gases has an influence of < 25 %, often < 5 % on the combined uncertainty of the amount of substance fraction.

From an evaluation of the measurement sequence, in total five bracketing combinations are possible. (Correlation from using C2,1; C2,2; C1,2 and C1,3 twice in the calculation is not considered.) The values and corresponding uncertainties obtained are:

Kurvenwert 1 / %	0.020017
$u(\text{Kurvenwert})$ / %	0.000018
Kurvenwert 2 / %	0.019947
$u(\text{Kurvenwert})$ / %	0.000014
Kurvenwert 3 / %	0.019972
$u(\text{Kurvenwert})$ / %	0.000011
Kurvenwert 4 / %	0.019960
$u(\text{Kurvenwert})$ / %	0.000015
Kurvenwert 5 / %	0.019983
$u(\text{Kurvenwert})$ / %	0.000015

The values from the measurement sequence are sufficiently compatible, i.e., they agree within their uncertainty. The five individual results (amount-of-substance fractions of propane) are pooled to:

Kurvenwert / %	0.019976
$u(\text{Kurvenwert})$ / %	0.000015
$u_{\text{rel}}(\text{Kurvenwert})$	7.4E-04

Since a measurement campaign consists of usually three full sequences (A1, A2, A3), the results (amount-of-substance fractions of propane) for each sequence are:

x_{A1} / %	0.019976
$u(x_{A1})$ / %	0.000015
$u_{\text{rel}}(x_{A1})$	7.4E-04

x_{A2} / %	0.019991
$u(x_{A2})$ / %	0.000019
$u_{\text{rel}}(x_{A2})$	9.4E-04

x_{A3} / %	0.019991
$u(x_{A3})$ / %	0.000015

$u_{\text{rel}}(X_{A3})$ 7.4E-04

All values from the measurement sequences are compatible, i.e., they agree within their uncertainty. The three individual results are pooled to the final result of a measurement campaign:

$X_{\text{BAM_MW}} / \%$	0.019986
$u(X_{\text{BAM_MW}}) / \%$	0.000016
$u_{\text{rel}}(X_{\text{BAM_MW}})$	8.1E-04
$U_{\text{rel}}(X_{\text{BAM_MW}})$	1.6E-03
$u^2(X_{\text{BAM_MW}})$	2.629E-10

Note that square root of n is not used in any of these calculations.

A.9 Considered sources of uncertainty

The results and uncertainties given here include the uncertainty of the composition of the calibration gases, the uncertainty from the measurement statistics (i.e., consecutive portions of three injections), the uncertainty propagation for the calibration approach, the bias within a measurement campaign over the period of time, and finally from combining the different individual measurement campaigns. Main source of uncertainty is the imprecision of the TCD and FID used.

A.10 Raw data:

2021-07-01 (for more details see \Code783_Euramet_K3)

2021-08-06 (for more details see \Code783_Euramet_K3\Thermo_4a)

2021-08-12 (for more details see \Code783_Euramet_K3\Thermo_5)

A.11 Results from the measurement campaigns

From the direct measurement the results of all compounds (note that the results were not normalized) are:

	Oxygen	Carbon monoxide	Carbon dioxide	Propane
Maxum				
$x_{\text{zert}} / \%$	2.9713	1.0134	1.9925	0.0200
$u(x_{\text{zert}}) / \%$	0.0008	0.0024	0.0025	0.0004
$u_{\text{rel}}(x_{\text{zert}})$	2.6E-04	2.4E-03	1.3E-03	1.8E-02
$U_{\text{rel}}(x_{\text{zert}})$	5.2E-04	4.8E-03	2.5E-03	3.6E-02
Thermo4a				
$x_{\text{zert}} / \%$	2.9718	1.0140	1.9953	0.019966
$u(x_{\text{zert}}) / \%$	0.0028	0.0007	0.0013	0.000015
$u_{\text{rel}}(x_{\text{zert}})$	9.3E-04	6.8E-04	6.4E-04	7.6E-04
$U_{\text{rel}}(x_{\text{zert}})$	1.9E-03	1.4E-03	1.3E-03	1.5E-03
Thermo5				
$x_{\text{zert}} / \%$	2.9748	1.0149	1.9972	0.019986
$u(x_{\text{zert}}) / \%$	0.0023	0.0007	0.0012	0.000016

$u_{\text{rel}}(x_{\text{zert}})$	7.9E-04	6.9E-04	6.2E-04	8.1E-04
$U_{\text{rel}}(x_{\text{zert}})$	1.6E-03	1.4E-03	1.2E-03	1.6E-03

The results from the two campaigns with the Thermo analyzer are consistent ($k < 1.1$) and pooled to:

Thermo all	Oxygen	Carbon monoxide	Carbon dioxide	Propane
$x_{\text{zert}} / \%$	2.973	1.0145	1.9963	0.019976
$u(x_{\text{zert}}) / \%$	0.003	0.0007	0.0013	0.000016
$u_{\text{rel}}(x_{\text{zert}})$	8.6E-04	6.9E-04	6.3E-04	7.8E-04
$U_{\text{rel}}(x_{\text{zert}})$	1.7E-03	1.4E-03	1.3E-03	1.6E-03

The values from Thermo and the Maxum campaign are also consistent ($k < 1.3$).

A.12 Consolidated results

consolidated	Oxygen	Carbon monoxide	Carbon dioxide	Propane
$x_{\text{zert}} / \%$	2.9733	1.0145	1.9963	0.019976
$u(x_{\text{zert}}) / \%$	0.0045	0.0015	0.0030	0.000040
$u_{\text{rel}}(x_{\text{zert}})$	1.5E-03	1.5E-03	1.5E-03	2.0E-03
$U_{\text{rel}}(x_{\text{zert}})$	3.0E-03	3.0E-03	3.0E-03	4.0E-03

Given is in the last row the expanded relative measurement uncertainty $U = u_c \cdot k$ with $k = 2$ according to the ISO/BIPM Guide to the Expression of Uncertainty in Measurement.

A.13 Responsibility

The calibration gases have been prepared by the filling team consisting of Claudia Boissière and Peer Wichmann under the supervision of Dr. Dirk Tuma. The measurements using the Maxum II process GC have been performed by Jeannette Pelchen, measurements using the Thermo analytical GC have been performed by Dr. Dirk Tuma. Strategy, reporting, and calculations have been performed by Dr. Heinrich Kipphardt. Advice was given by Dr. Michael Maiwald. The overall technical responsibility for the measurement result is with Dr. Heinrich Kipphardt.

Berlin, 2021-09-03

Heinrich Kipphardt FB 1.4

A.14 Appendix: Purity Tables with typical examples

Material	Nitrogen	99.99991	0.00003	3.3E-07
gas quality	6.0			
supplier	Linde			
batch	276311-21144639			
	Impurities	0.00010	0.00003	3.4E-01
	Sum	100.00000		

Index	compound	M / g/mol	$u(M)$ / g/mol	vq	x / %	$u(x)$ / %	$u_r(x)$
1	Nitrogen	28.01	0.00		99.99991	0.00003	3.3E-07
2	Argon	39.95	0.00		0.00005	0.00003	5.8E-01
3	Oxygen	16.00	0.00		0.00000	0.00000	5.8E-01
4	Carbon dioxide	44.01	0.01		0.00001	0.00000	5.8E-01
5	Carbon monoxide	28.01	0.01		0.00000	0.00000	5.8E-01
6	Methane	16.04	0.01		0.00000	0.00000	5.8E-01
17	Hydrogen	1.01	0.00		0.00000	0.00000	5.8E-01
31	Moisture	18.01	0.00		0.00003	0.00001	5.8E-01
36	UNKNOWN	80.00	20.00		0.00001	0.00000	5.8E-01
37	2-Butine	54.09	0.01				

Material	Oxygen				99.99989	0.00004	3.6E-07
gas quality	6.0						
supplier	Linde						
batch	276311-20326434						
	Impurities				0.00012	0.00004	3.1E-01
	Sum				100.00000		

Index	compound	M / g/mol	$u(M)$ / g/mol	vq	x / %	$u(x)$ / %	$u_r(x)$
1	Nitrogen	28.01	0.00		0.00003	0.00001	5.8E-01
2	Argon	39.95	0.00		0.00005	0.00003	5.8E-01
3	Oxygen	16.00	0.00		99.99989	0.00004	3.6E-07
4	Carbondioxide	44.01	0.01		0.00001	0.00000	5.8E-01
5	Carbon monoxide	28.01	0.01		0.00001	0.00000	5.8E-01
9	Propane	44.10	0.01		0.00001	0.00000	5.8E-01
31	Moisture	18.01	0.00		0.00003	0.00001	5.8E-01

Material	Carbon dioxide				99.99971	0.00009	8.9E-07
gas quality	5.5						
supplier	Air Liquide						
batch							
	Impurities				0.00029	0.00009	3.1E-01
	Sum				100.00000		

Index	compound	M / g/mol	$u(M)$ / g/mol	vq	x / %	$u(x)$ / %	$u_r(x)$
1	Nitrogen	28.01	0.00		0.00010	0.00006	5.8E-01
3	Oxygen	16.00	0.00		0.00005	0.00003	5.8E-01
4	Carbon dioxide	44.01	0.01		99.99971	0.00009	8.9E-07
5	Carbon monoxide	28.01	0.01		0.00003	0.00002	5.8E-01
22	NO	30.01	0.10		0.00001	0.00000	5.8E-01
31	Moisture	18.01	0.00		0.00010	0.00006	5.8E-01
36	unknown	80.00	20.00		0.00001	0.00000	5.8E-01

Material	Carbon monoxide	99.99929	0.00021	2.1E-06
gas quality	4.7			
supplier	Linde			
batch	276311-20642382 (24125)			
	Impurities	0.00071	0.00021	2.9E-01
	Sum	100.00000		

Index	compound	M / g/mol	$u(M)$ / g/mol	vq	x / %	$u(x)$ / %	$u_r(x)$
1	Nitrogen	28.01	0.00		0.00020	0.00012	5.8E-01
2	Argon	39.95	0.00		0.00010	0.00006	5.8E-01
3	Oxygen	16.00	0.00		0.00001	0.00001	5.8E-01
5	Carbon monoxide	28.01	0.01		99.99929	0.00021	2.1E-06
9	Propane	44.10	0.01		0.00010	0.00006	5.8E-01
17	Hydrogen	1.01	0.00		0.00005	0.00003	5.8E-01
31	Moisture	18.01	0.00		0.00025	0.00014	5.8E-01

Material	Propane	99.97175	0.01198	1.2E-04
gas quality	3.5			
supplier	Gerling			
batch	25E			
	Impurities	0.02825	0.01198	4.2E-01
	Sum	100.00000		

Index	compound	M / g/mol	$u(M)$ / g/mol	vq	x / %	$u(x)$ / %	$u_r(x)$
1	Nitrogen	28.01	0.00		0.00200	0.00116	5.8E-01
3	Oxygen	16.00	0.00		0.00050	0.00029	5.8E-01
4	Carbon dioxide	44.01	0.01		0.00050	0.00029	5.8E-01
9	Propane	44.10	0.01		99.97175	0.01198	1.2E-04
10	Propene	42.08	0.01		0.00500	0.00289	5.8E-01

Annex B Measurement report of CEM

B.1 Results

Cylinder number: 8500E

Measurement #1

Component	Date (dd/mm/yy)	Result (cmol/mol)	Standard deviation (% relative)	number of replicates
Oxygen	21/07/2021	3.0356	0.09	6
Carbon dioxide	05/08/2021	2.0272	0.18	6
Propane	16/07/2021	0.020225	0.36	6
Carbon monoxide	13/08/2021	0.9927	0.12	6

Measurement #2

Component	Date (dd/mm/yy)	Result (cmol/mol)	Standard deviation (% relative)	number of replicates
Oxygen	06/08/2021	3.0080	0.11	6
Carbon dioxide	11/08/2021	2.0073	0.13	6
Propane	20/07/2021	0.019897	0.49	6
Carbon monoxide	16/08/2021	0.9964	0.11	6

Measurement #3

Component	Date (dd/mm/yy)	Result (cmol/mol)	Standard deviation (% relative)	number of replicates
Oxygen	09/08/2021	3.0401	0.07	6
Carbon dioxide	12/08/2021	2.0136	0.16	6
Propane	22/07/2021	0.01990	0.54	6
Carbon monoxide	17/08/2021	0.9960	0.12	6

Results

Component	Result (cmol/mol)	Expanded uncertainty (cmol/mol)	Coverage factor
Oxygen	3.028	0.021	2
Carbon dioxide	2.016	0.013	2
Propane	0.02001	0.00029	2
Carbon monoxide	0.9962	0.0046	2

B.2 Calibration standards

Method of preparation: Two sets of primary standard gas mixtures (PSMs) were prepared using the gravimetric method according to ISO 6142-1:2015 [1]. All PSMs were prepared in nitrogen balance. The standard uncertainty relies on the uncertainty of the gravimetric preparation method and the purity data of the parent gases. Occasionally, other available PSMs were used in some of the analysis.

Weighing data:

- A first batch of six PSMs of propane (nominally amount fraction of 30 $\mu\text{mol/mol}$ to 350 $\mu\text{mol/mol}$) and oxygen (nominally amount fraction of 0.45 cmol/mol to 12 cmol/mol) in nitrogen was prepared by first adding pre-mixtures of propane in nitrogen (nominally amount fraction of 900 $\mu\text{mol/mol}$ to 4 cmol/mol) and then nitrogen and oxygen from pure sources.

Table 1: PSMs of propane and oxygen in nitrogen balance.

Cylinder	Component	Assigned value, x_i (cmol/mol)	Expanded uncertainty, U_i (cmol/mol)	Coverage factor k
MRP100177	Oxygen	0.4551	0.0019	2
	Propane	0.003018	0.000025	2
MRP100171	Oxygen	9.5049	0.0016	2
	Propane	0.005035	0.000074	2
MRP548527	Oxygen	4.8045	0.0018	2
	Propane	0.01003	0.00011	2
MRP100170	Oxygen	9.6320	0.0016	2
	Propane	0.02019	0.00014	2
MRP601155	Oxygen	9.5342	0.0024	2
	Propane	0.03151	0.00021	2
MRP392354	Oxygen	12.2114	0.0020	2
	Propane	0.03499	0.00023	2

- A second batch of three PSMs of carbon dioxide (nominally amount fraction of 2 cmol/mol to 5 cmol/mol), carbon monoxide (nominally amount fraction of 0.5 cmol/mol to 2 cmol/mol), and oxygen (nominally amount fraction of 2.8 cmol/mol to 3.2 cmol/mol) in nitrogen balance was prepared. Pre-mixtures of carbon monoxide in nitrogen (nominally amount fraction of 5 cmol/mol and 6 cmol/mol) were added in two of the three mixtures. Carbon dioxide, carbon monoxide, oxygen and nitrogen were then added from pure sources.

Table 2: PSMs of carbon dioxide, carbon monoxide and oxygen in nitrogen balance.

Cylinder	Component	Assigned value, x_i (cmol/mol)	Expanded uncertainty, U_i (cmol/mol)	Coverage factor k
MRP502452	Oxygen	2.8184	0.0036	2
	Carbon monoxide	0.5083	0.0012	2
	Carbon dioxide	1.9869	0.0013	2
MRP302441	Oxygen	3.0355	0.0035	2
	Carbon monoxide	1.2286	0.0012	2
	Carbon dioxide	3.2970	0.0026	2
MRP502447	Oxygen	3.2458	0.0051	2
	Carbon monoxide	2.7699	0.0047	2
	Carbon dioxide	4.9065	0.0026	2

- PSMs occasionally used: carbon dioxide (nominally amount fraction of 5 cmol/mol and 17 cmol/mol), and carbon monoxide (nominally amount fraction of 0.5 cmol/mol), in nitrogen balance.

Table 3: PSMs of carbon dioxide, carbon monoxide in nitrogen balance.

Cylinder	Component	Assigned value, x_i (cmol/mol)	Expanded uncertainty, U_i (cmol/mol)	Coverage factor k
MRP602434	Carbon dioxide	5.0023	0,0011	2
MRP166321	Carbon dioxide	17.0015	0.0011	2
MRP306300	Carbon monoxide	0.5129	0.0010	2

Purity tables (composition) of the parent gases:

Nitrogen 6.0 (LINDE)			Carbon dioxide N48 (AIR LIQUIDE)			Oxygen ULTRAPLUS 6X (PRAXAIR)		
COMPONENT	x_i (mmol/mol)	u_i (mmol/mol)	COMPONENT	x_i (mmol/mol)	u_i (mmol/mol)	COMPONENT	x_i (mmol/mol)	u_i (mmol/mol)
Water	0.25	0.14	Oxygen	1	0.58	Carbon dioxide	0.05	0.029
Oxygen	0.25	0.14	Hydrocarbons	1	0.58	Carbon monoxide	0.10	0.058
Carbon monoxide	0.05	0.029	Hydrogen	0.25	0.14	Hydrocarbons	0.03	0.014
Carbon dioxide	0.05	0.029	Water	1.5	0.87	Argon	0.50	0.29
Hydrocarbons	0.05	0.029				Nitrogen	0.50	0.29
Nitrogen	999999.35	0.21	Carbon dioxide	999996.2	1.2	Water	0.25	0.14
						Oxygen	999998.58	0.44

Table 4: Purity tables of pure gases.

Carbon monoxide N47 (NIPPON GASES)			Propane N35 (AIR LIQUIDE)		
COMPONENT	x_i (mmol/mol)	u_i (mmol/mol)	COMPONENT	x_i (mmol/mol)	u_i (mmol/mol)
Argon	1	0.58	Propene	100	58
Carbon dioxide	0.5	0.29	Hydrocarbons	100	58
Oxygen	0.5	0.29	Nitrogen	20	12
Nitrogen	10	5.8	Hydrogen	20	12
Water	0.5	0.29	Oxygen	5	3
Hydrogen	10	5.8	Carbon dioxide	2	1
Hydrocarbons	0.5	0.29	Water	2	1
Carbon monoxide	999977	8	Propane	999750	83

Verification measures:

- The PSM were validated against CEM's calibration standards or against each other, using gas chromatography with thermal conductivity detector (micro-GC-TCD).
- The levels of water, methane, carbon monoxide and carbon dioxide in high purity nitrogen used as gas balance were checked by Cavity ring-down spectroscopy (CRDS). Results showed concentration of these compounds well below the specifications given by manufacturer.

B.3 Instrumentation

The analyses were performed in a micro gas chromatograph Agilent 3000A with thermal conductivity detector (micro-GC-TCD). Helium 5.6 (Nippon Gases) is used as carrier gas. The micro-GC is equipped with 2 channels:

- Channel A, with a PLOT-U (3 m x 0.30 mm x 30 μ m) pre-column and a Molsieve 5A column (10 m x 0.30 mm x 30 μ m) and a backflush injector type of 1 μ L.
- Channel B, with a OV-1 (length: 8 m; thick: 2.0 μ m) with fixed injector type.

A single analytical method is programmed to analyse oxygen in channel A and propane and carbon dioxide in channel B. A different method is used to analyse carbon monoxide using only channel A. The runtime of both methods is close to 60 s each.

After rolling the cylinders, the sample and the PSMs are connected to pressure regulators. A set of four 5-ways-valves are connected to the sampling rig and used for connecting the sample cylinder and the standards with the analyser. This rig is used to purge before the analysis of each cylinder.

B.4 Calibration method and value assignment

PSMs and sample cylinder are analysed individually in increasing order of concentration. A minimum of 40 injections per cylinder are performed meanwhile a minimum of 4 injections of pure nitrogen (6.0 LINDE) were done between cylinders. Only the 6 last analytical signals are used to calculations.

The analysis are performed under repeatability conditions during at least three days. Three results are selected for each component. The calibration curve, results of analysis and uncertainties associated were determined using the methodology of ISO 6143 [2]. The assigned values for components concentration is the average of the three individual values obtained.

Component	Measurement method	Type of calibration curve
Oxygen	GC/TCD	3 points, linear 3 points, linear 3 points, linear
Carbon dioxide	GC/TCD	3 points, linear 3 points, linear 3 points, linear
Propane	GC/TCD	3 points, linear 5 points, quadratic 3 points, linear
Carbon monoxide	GC/TCD	3 points, linear 3 points, linear 3 points, linear

Table 5: Measurement method and type of calibration curve used.

B.5 Uncertainty evaluation

The expression for combined standard uncertainty (u_c) includes the quadratic sum of individual standard uncertainties as obtained according to ISO 6143 from each measurement and the standard deviation of the mean from average the three results:

$$u_c = \sqrt{\frac{1}{3}(u_1^2 + u_2^2 + u_3^2) + \left(\frac{s}{\sqrt{3}}\right)^2}$$

The expanded uncertainty (U) was obtained by multiplying the combined uncertainty with a coverage factor of $k = 2$.

$$U = k \cdot u_c$$

References

- [1] International Organization for Standardization, "ISO 6142-1:2015 – Gas analysis -- Preparation of calibration gas mixtures – Part 1: Gravimetric method for Class I mixtures", ISO Geneva, 2015.
- [2] International Organization for Standardization, "ISO 6143 – Gas analysis – Comparison methods for determining and checking the composition of calibration gas mixtures", ISO Geneva, 2001.

Annex C Measurement report of LNE

C.1 Calibration standards

C.1.1 Method of preparation

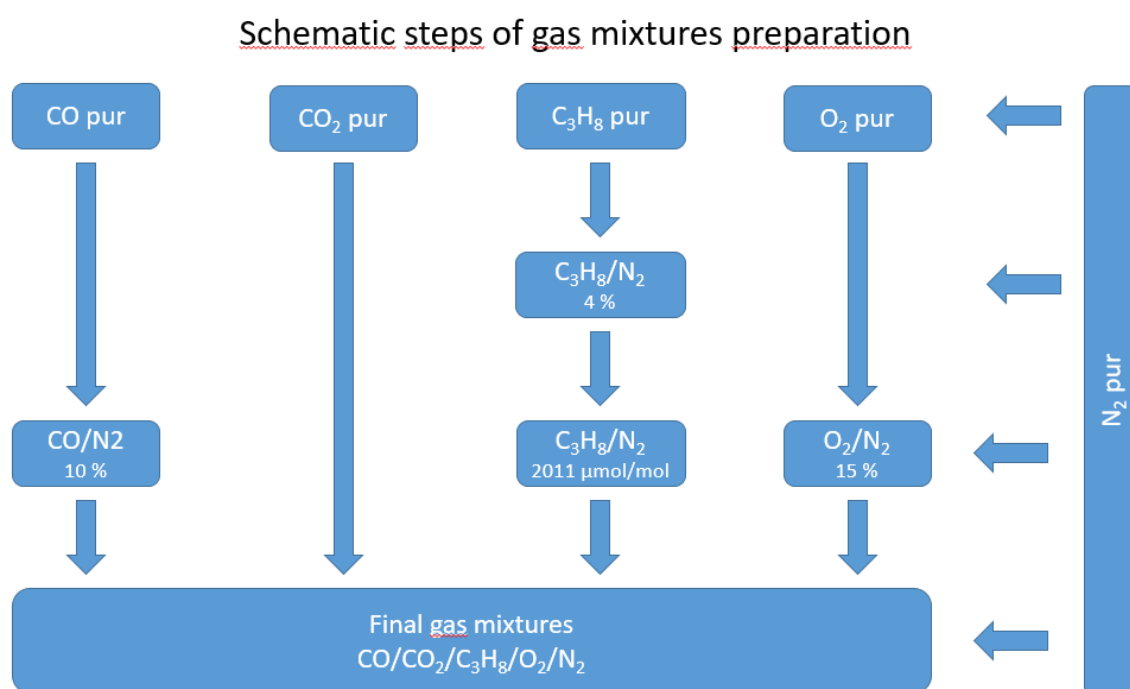
The standard gas mixtures were prepared according to the ISO 6142 standard.

Gas cylinders were evacuated using a turbo vacuum pump and weighed using a Mettler AX32004 mass comparator with a 0.1 mg resolution.

For propane, two premix gas mixtures in nitrogen were produced at amount fractions of 4 % and 0.2 %.

For CO and O₂, one premix gas mixture in nitrogen was produced respectively at amount fractions of 10% and 15%.

The final standard gas mixtures were produced by gravimetric dilution of the different premix gas mixtures and pure CO₂ in nitrogen as shown below.



C.1.2 Weighing data and amount fractions of Gazmot 0033

----- Gazmot 0033 -----			
Component	μmol/mol	uncertainty	% u/c

N ₂	939180.4701	7.63198477	0.001
O ₂	29922.71076	2.11616911	0.007
CO ₂	20413.95936	7.50705393	0.037
CO	10277.19254	1.41425695	0.014
propane	205.4284253	0.05279044	0.026
H ₂ O	0.06342333	0.02057347	32.438
Ar	0.05138667	0.02966840	57.736
methane	0.04538974	0.01477692	32.556
H ₂	0.03932743	0.01012582	25.747
C _n H _m	0.03082531	0.01326367	43.028
propene	0.02054798	0.01186222	57.729

INPUTS

=====

File	Mass (g)	u/c (g)

Pur\CO2pur009.tx	37.46552	0.01400
melanges\co_n2_o2	119.9104	0.01400
melanges\o2_n2_o2	237.8485	0.01500
melanges\c3h8_n2	119.4520	0.01600
G_d_f\N2_Bip_plus	672.2678	0.02100

----- Pur\CO2pur009.txt -----		
Component	μmol/mol	uncertainty

CO ₂	0.9999922500	0.0000026021
H ₂ O	0.0000015000	0.0000008660
O ₂	0.0000010000	0.0000005774
methane	0.0000010000	0.0000005774
H ₂	0.0000002500	0.0000001443
N ₂	0.0000040000	0.0000023094

----- Gas mixture\co_n2_0077.txt -----		
--	--	--

Component	$\mu\text{mol/mol}$	uncertainty
N ₂	0.8998755361	0.0000093232
CO	0.1001233929	0.0000093025
Ar	0.0000005006	0.0000002890
H ₂ O	0.0000001592	0.0000000869
O ₂	0.0000001547	0.0000000867
C _n H _m	0.0000001001	0.0000000578
H ₂	0.0000000726	0.0000000317
CO ₂	0.0000000613	0.0000000296
methane	0.0000000225	0.0000000130

----- Gas mixture\o2_n2_0030.txt -----

Component	$\mu\text{mol/mol}$	uncertainty
N ₂	0.8498931366	0.0000071596
O ₂	0.1501067912	0.0000071557
H ₂ O	0.0000000460	0.0000000223
methane	0.0000000288	0.0000000130
H ₂	0.0000000288	0.0000000130
CO ₂	0.0000000181	0.0000000075
CO	0.0000000111	0.0000000061

----- Gas mixture\c3h8_n2_0036.txt -----

Component	$\mu\text{mol/mol}$	uncertainty
N ₂	0.9979880776	0.0000005075
propane	0.0020113698	0.0000004550
propene	0.0000002012	0.0000001161
C _n H _m	0.0000002012	0.0000001161
H ₂	0.0000000652	0.0000000270
methane	0.0000000249	0.0000000137
CO ₂	0.0000000175	0.0000000074
O ₂	0.0000000150	0.0000000064
H ₂ O	0.0000000150	0.0000000062
CO	0.0000000125	0.0000000068

----- G_d_f\N2_Bip_plus.txt -----

Component	μmol/mol	uncertainty
N ₂	0.9999999100	0.0000000237
H ₂ O	0.0000000100	0.0000000058
O ₂	0.0000000050	0.0000000029
CO	0.0000000125	0.0000000072
CO ₂	0.0000000125	0.0000000072
methane	0.0000000250	0.0000000144
H ₂	0.0000000250	0.0000000144

C.1.3 Analytical validation of the preparation of the standard gas mixture Gazmot 0033

The standard gas mixture Gazmot 0033 was compared to other gas mixtures to validate the gravimetric preparation. The standard gas mixture was analysed ten times in repeatability conditions. The average amount fraction and the standard deviation of the five last injections were calculated.

The results are used to calculate the gravimetric uncertainties according to the ISO 6142-1 standard.

$$u_{prep} = \frac{1}{2} \sqrt{u_{gravi}^2 + u_{stab}^2 + u_{analy}^2 + (prep - analy)^2}$$

The gravimetric gas mixture is validated if:

$$|x_{analy} - x_{gravi}| < 2 \times \sqrt{u_{prep}^2 + u_{analy}^2}$$

Validation of the gravimetric preparation of the Gazmot 0033 standard gas mixture; all the data are given in μmol/mol

Compounds	X _{gravi}	u _{gravi}	u _{stab}	U _{prep} k=2	X _{analy}	u _{analy}	Validation
CO	10277.2	1.5	0.15%	41.6	10251	28.57	Yes
CO ₂	20414.0	7.5	0.05%	101.1	20496	57.69	Yes
C ₃ H ₈	205.428	0.053	0.05%	0.50	205.27	0.46	Yes
O ₂	29922.71	2.12	0.00%	184.3	29993.9	170.00	Yes

C.2 Instrumentation

The automotive exhaust gas mixtures were analysed by gas chromatography on a AGILENT instrument equipped with a 250 μl sample loop, a PDHID detector and a FID with methanizer.

Oxygen was eluted with a 50 m, 0.53 mm 5A Molsieve capillary column coupled to the PDHID detector. The oven was regulated at 30°C during 6 min and then the temperature rose to 100°C at 20°C/min during 5 min.

CO, CO₂ and C₃H₈ were eluted on a 30 m, 0.53 mm PLOTQ capillary column coupled to the FID detector with a methanizer. The oven was regulated at 40°C during 4.5 min and then the temperature rose to 130°C at 40°C/min during 2 min.

C.3 Calibration method and value assignment

The gas chromatograph was calibrated at a single point close to the amount fraction of the gas sample according to the sequence “standard gas mixture Gazmot 0033/gas sample N°8457E from VSL/ standard gas mixture Gazmot 0033” with 10 runs for each gas (only the last five runs were used for the calculation).

The amount fraction of the gas sample was determined as follows:

$$x = 2 \times \frac{S_{\text{sample}} \times x_{\text{gravi}}}{S_{\text{gravi1}} + S_{\text{gravi2}}} \quad (\text{Eq.1})$$

where

S_{sample} : Average of the last 5 chromatographic areas of the compound peaks for the gas sample N°8457E

x_{gravi} : Amount fraction of the standard gas mixture Gazmot 0033

S_{gravi1} : Average of the last 5 chromatographic areas of the compound peaks for the standard gas mixture Gazmot 0033 before the injection of the gas sample N°8457E

S_{gravi2} : Average of the last 5 chromatographic areas of the compound peaks for the standard gas mixture Gazmot 0033 after the injection of the gas sample N°8457E

The procedure is repeated three times on 3 different days.

The final amount fraction is the average of the three determinations obtained on the 3 different days.

C.4 Uncertainty evaluation

The uncertainty evaluation is established by combining the uncertainty on the amount fractions of the standard gas mixture and the repeatability and reproducibility standard deviations of the measurements; they are calculated according to the law of propagation of uncertainty (Eq.2).

$$u_c^2(y) = \sum_{i=1}^N \left(\frac{\partial f}{\partial x_i} \right)^2 \times u^2(x_i) + 2 \sum_{i=1}^{N-1} \sum_{j=i+1}^N \left(\frac{\partial f}{\partial x_i} \right) \left(\frac{\partial f}{\partial x_j} \right) u(x_i, x_j) \quad (\text{Eq. 1})$$

The table 4 shows an example for the determination of the CO amount fraction and the associated expanded uncertainty on the first day of the measurements (16/09/2021).

Uncertainty budget on the CO amount fraction for cylinder 8457E determined on the first day (16/09/2021)

Variable	Units	Value	u(Xi)	C(Xi)	C(Xi).u(Xi)	Weight%
X _{gravi}	cmol/mol	1.0277	0.00208	0.984807	0.002048	55.56%
S _{gravi1}		7160	10	-0.000071	-0.000670	5.95%
S _{sample}		7027	11	0.0001440	0.0015490	31.77%
S _{gravi2}		7111	10	-0.000071	-0.000712	6.71%

C_CO	cmol/mol	1.0121	0.0027	(k = 1)
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The final uncertainty is the quadratic sum of the maximal uncertainty calculated for the three determinations and the standard deviation of the three determinations.

$$u = \sqrt{u_c^2 + std_{repro}^2}$$

With:

u_c Maximal uncertainty calculated for the three measurements

Std_{repro} Standard deviation of the three determinations of the amount fractions under reproducibility conditions

Annex D Measurement report of BFKH

D.1 Calibration standards

D.1.1 Overview

Cylinder Identification Number	0153A		0157A		D521804	
Component	x ppm*	Expanded Measurement Uncertainty U (k=2) ppm	x ppm	Expanded Measurement Uncertainty U (k=2) ppm	x ppm	Expanded Measurement Uncertainty U (k=2) ppm
Oxygen	30123.47	6.0	30085.46	6.0	30061.31	6.0
Carbon dioxide	20003.28	3.6	19973.27	3.6	20002.18	3.6
Propane	200.81	0.10	200.98	0.10	200.90	0.10
Carbon monoxide	10073.27	1.9	10049.57	1.9	10026.77	1.9
Water	1.54	1.8	1.54	1.8	1.54	1.8
Methane	0.10	0.11	0.10	0.11	0.10	0.11
Argon	0.04	0.04	0.04	0.04	0.04	0.04

*1 ppm = 1 $\mu\text{mol mol}^{-1}$

D.1.2 Method of preparation of the standard

5 mixtures were prepared before:

	Cylinder No	Composition	Component	g	Component	g
1.	OMH231	30.091678 % O ₂ -N ₂	O ₂ 5.0	193.7311	N ₂ 5.0	394.0107
2.	OMH279	21.414597 % CO ₂ -N ₂	CO ₂ 4.8	192.8283	N ₂ 5.0	450.4207
3.	OMH222	4.07337 % C ₃ H ₈ -N ₂	C ₃ H ₈ 3.5	35.5720	N ₂ 5.0	539.8303
4.	OMH217	0.211177 % C ₃ H ₈ -N ₂	4.07337 % C ₃ H ₈ -N ₂	30.3193	N ₂ 5.0	534.1420
5.	A8629	9.986464 % CO-N ₂	CO 4.7	118.1035	N ₂ 5.0	1064.6446

D.1.3 Weighing data of preparation of the standard:

Standard 1.

Step	Component	g
1.	Mixture 1	123.4550
2.	Mixture 2	123.9795
3.	Mixture 4	112.5984
4.	Mixture 5	119.2918

5.	N ₂ 5.0	722.0410
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Standard 2.

Step	Component	g
1.	Mixture 1	123.1902
2.	Mixture 2	123.6840
3.	Mixture 4	112.5941
4.	Mixture 5	118.9059
5.	N ₂ 5.0	721.9026

Standard 3.

Step	Component	g
1.	Mixture 1	122.9387
2.	Mixture 2	123.7095
3.	Mixture 4	112.4083
4.	Mixture 5	118.4891
5.	N ₂ 5.0	721.2588

Parent gases:

Gas	Quality	Supplier
Nitrogen	99.999	Messer
Oxygen	99.999	Messer
Carbon dioxide	99.998	Siad
Propane	99.95	Messer
Carbon monoxide	99.997	Siad

Purity table of pure Oxygen:

Component	Method	Mole fraction (mol/mol)	Uncertainty (mol/mol)
O ₂	specifications	0.99998675	0.0000153
N ₂	specifications	0.00001000	0.0000115
H ₂ O	specifications	0.00000250	0.0000029
CH ₄	specifications	0.00000025	0.00000029
CO	specifications	0.00000025	0.00000029
CO ₂	specifications	0.00000025	0.00000029

Purity table of pure CO₂: (88434)

Component	Method	Mole fraction (mol/mol)	Uncertainty (mol/mol)
CO ₂	specifications	0.999990	0.000012
N ₂	specifications	0.0000050	0.0000058
H ₂ O	specifications	0.0000025	0.0000029
CH ₄	specifications	0.0000010	0.0000012
O ₂	specifications	0.0000010	0.0000012
CO	specifications	0.0000005	0.00000058

Purity table of pure C₃H₈:

Component	Method	Mole fraction (mol/mol)	Uncertainty (mol/mol)
C ₃ H ₈	measured	0.9997825	0.000100

N ₂	measured	0.000030	0.000010
H ₂ O	specifications	0.0000025	0.0000029
CH ₄	measured	0.000060	0.000010
O ₂	specifications	0.000005	0.0000058
CO ₂	measured	0.000120	0.000010

Purity table of pure CO:

Component	Method	Mole fraction (mol/mol)	Uncertainty (mol/mol)
CO	specifications	0.9999895	0.000012
N ₂	specifications	0.0000025	0.0000029
H ₂ O	specifications	0.0000015	0.0000017
H ₂	specifications	0.0000005	0.00000058
CH ₄	specifications	0.000001	0.0000012
O ₂	specifications	0.0000015	0.0000017
Ar	specifications	0.0000035	0.0000041

Purity table of pure nitrogen:

Component	Method	Mole fraction (mol/mol)	Uncertainty (mol/mol)
N ₂	specifications	0.99999745	0.0000029
O ₂	specifications	0.000001	0.00000115
H ₂ O	specifications	0.0000015	0.00000173
CH ₄	specifications	0.00000005	0.000000058

Verification measures confirmed the gravimetric method validity, their results were within the stated uncertainty of the standard in case of each components.

D.2 Instrumentation

D.2.1 Measurement of Carbon dioxide and Propane

HP 6890 GC-TCD/FID with two parallel columns:

to TCD: Restek Haysep A 8.8 m x 0.75 mm Sulfinert SST, Pressure in 10 bar He,

to FID: Restek Porapak PS 4.4 m x 0.75 mm Sulfinert SST, Pressure in 4.5 bar He.

Isotherm method at 180 °C.

D.2.2 Measurement of Carbon monoxide and Oxygen

HP 7890 GC-TCD/FID with two parallel columns:

to TCD: Restek PCK 5A Mole Sieve 80/100 2m 1.00 mm ID

to FID: Restek PCK PP-PS 100/120 4.4 m 0.75 MMID 0.95

Isotherm method at 70 °C.

D.3 Calibration method and value assignment

The results came from two independent measurements.

1. Measurement of Carbon dioxide and Propane with HP 6890 GC-TCD/FID with two parallel columns.

The sample injections were in case of the standard and the sample during the same run with 3.5 min different, and sample exchange were every 30 min. After the excluding the outliers, the area data were used for the calculation.

2. Measurement of Carbon monoxide and Oxygen (and Propane) with HP 7890 GC-TCD/FID with two parallel columns.

The sample injections were in case of the standard and the sample during the same run with 4 min different, and sample exchange were every 60 min. After the excluding the outliers, the area data were used for the calculation.

D.4 Uncertainty evaluation

The standard uncertainty of x_i component can be expressed as

$$u^2(x_i) = u_{x_i, meas}^2 + u_{sti}^2 + u_{sti, meas}^2 \quad (1)$$

$$U(x_i) = k \times u(x_i) \quad (2)$$

x_i	amount-of-substance fraction of gas i component
$u_{x_i, meas}$	uncertainty of measurements of gas i component
u_{sti}	uncertainty of calibration standard value of i component
$u_{sti, meas}$	uncertainty of measurements of calibration standard i component
$U_{i, x}$	stated uncertainty of laboratory, at 95 % level of confidence (cmol/mol)
k	assigned coverage factor for degree of equivalence

D.5 Results

The table shows the detailed results in case of measurement.

Component	x_i	$u_{x_i, meas}$	u_{sti}	$u_{sti, meas}$	k	$U(x_i)$	$U(x_i)$
	%	rel	rel	rel		rel	%
Oxygen	3.0032	0.0001 9	0.00010	0.00036	2	0.00084	0.0025
Carbon dioxide	1.9832	0.0003	0.00009	0.0009	2	0.0019	0.0038
Propane	0.020060	0.0014	0.00025	0.0016	2	0.0043	0.00008 6
Carbon monoxide	1.0021	0.0012 1	0.00010	0.00090	2	0.00302	0.0030

Component	x_i	$u_{x_i, meas}$	u_{sti}	$u_{sti, meas}$	k	$U(x_i)$	$U(x_i)$
	%	rel	rel	rel		rel	%
Oxygen	3.0015	0.0009 5	0.00010	0.00048	2	0.00214	0.0064

Component	x_i	$u_{x_i, \text{meas}}$	u_{sti}	$u_{\text{sti, meas}}$	k	$U(x_i)$	$U(x_i)$
	%	rel	rel	rel		rel	%
Carbon dioxide	1.9781	0.0008	0.00009	0.0005	2	0.0019	0.0037
Propane	0.020037	0.0007	0.00025	0.0013	2	0.0030	0.000060
Carbon monoxide	1.0024	0.0013 5	0.00010	0.0008	2	0.00314	0.0032

Component	x_i	$u_{x_i, \text{meas}}$	u_{sti}	$u_{\text{sti, meas}}$	k	$U(x_i)$	$U(x_i)$
	%	rel	rel	rel		rel	%
Oxygen	3.0032	0.0003 0	0.00010	0.00065	2	0.00145	0.0043
Carbon dioxide	1.9852	0.0007	0.00009	0.0013	2	0.00296	0.0059
Propane	0.020097	0.0010	0.00025	0.0010	2	0.00287	0.000058
Carbon monoxide	1.0012	0.0019 3	0.00010	0.00110	2	0.00445	0.0045

Component	x_i	$u_{x_i, \text{meas}}$	u_{sti}	$u_{\text{sti, meas}}$	k	$U(x_i)$	$U(x_i)$
	%	rel	rel	rel		rel	%
Oxygen	3.0022	0.0013 5	0.00010	0.00126	2	0.00370	0.0111
Carbon dioxide	1.9862	0.0005	0.00009	0.0005	2	0.00143	0.0028
Propane	0.020038	0.0006	0.00025	0.0013	2	0.00291	0.000058
Carbon monoxide	1.0015	0.0014 8	0.00010	0.00237	2	0.00559	0.0056

Component	x_i	$u_{x_i, \text{meas}}$	u_{sti}	$u_{\text{sti, meas}}$	k	$U(x_i)$	$U(x_i)$
	%	rel	rel	rel		rel	%
Oxygen	3.0052	0.0002 4	0.00010	0.00027	2	0.00075	0.0023
Carbon dioxide	1.9828	0.0005	0.00009	0.0008	2	0.00190	0.0038
Propane	0.020026	0.0007	0.00025	0.0006	2	0.00191	0.000038
Carbon monoxide	1.0033	0.0006 3	0.00010	0.00060	2	0.00175	0.0018

Summary of all measurements

Component	Result	k	Expanded uncertainty $U(x_i)$
	%		%
Oxygen	3.0031	2	0.0028

Carbon dioxide	1.9831	2	0.0018
Propane	0.020052	2	0.000028
Carbon monoxide	1.0021	2	0.0017

The submitted value of the component came from the average of the five measurements. The uncertainty of this value calculated from the root of sum of the independent uncertainty's square by divided 5.

References

CCQM key comparison CCQM-K3 of measurements of CO, CO₂, and C₃H₈ in N₂, É. Deák - **A van der Veen and all**, [Metrologia](#), **Volume 39**, **Number 1**, 2002 *Metrologia* 39 121

Annex E Measurement report of GUM

E.1 Calibration standards

Composition of calibration standards:

Cylinder number	Component	Assigned value mol/mol	Standard uncertainty mol/mol
D752078_2	CO ₂	0.020135	0.000033
	C ₃ H ₈	0.0001001	0.0000002
	CO	0.005054	0.000006
	O ₂	0.010107	0.000007
	N ₂	balance	

Cylinder number	Component	Assigned value mol/mol	Standard uncertainty mol/mol
D752091_2	CO ₂	0.034476	0.000034
	C ₃ H ₈	0.0001997	0.0000003
	CO	0.012639	0.000011
	O ₂	0.027914	0.000046
	N ₂	balance	

Cylinder number	Component	Assigned value mol/mol	Standard uncertainty mol/mol
D752094_2	CO ₂	0.050020	0.000035
	C ₃ H ₈	0.0003014	0.0000006
	CO	0.020104	0.000012
	O ₂	0.042508	0.000047
	N ₂	balance	

Standards were prepared (by Central Office of Measures) by gravimetric method according to ISO 6142 from separate premixtures. The premixtures were prepared by using: propane 3.5, carbon dioxide 4.5, carbon monoxide 4.7, oxygen 5.5 and nitrogen 6.0 The minimal weighed samples are 20 g. The minimal weighed sample of nitrogen is 150g. The cylinders were evacuated on turbo molecular pump, filled up and weighted on the verification balance. All the standards were prepared in aluminum (with coated layers) cylinders. The standards were (and still are) under metrological control.

E.2 Instrumentation

The measurements were repeated 9 times for the sample and the standards by gas chromatograph Thermo Trace 1300 with TCD detector and FID detector with capillary column. The cylinders (standards and sample) were in the same room for the whole time also during the measurements (temperature stabilization) and the mixtures were mixed up before the measurements. Samples were transferred to the instrument via the reducing valve and the automatic input pressure stabilization system.

E.3 Calibration method and value assignment

The measurements were made by using standards: D752078_2, D752091_2, D752094_2 by multi point calibration according to ISO 6143.

E.4 Uncertainty evaluation

The final uncertainty, consists of the following components:

- the uncertainty of standard preparation calculated according to ISO 6142
- the standard deviation of the measurement

Resolution of the chromatograph is negligible.

Annex F Measurement report of SMU

Cylinder number: **8449E**

F.1 Calibration standards

All calibration standards were made gravimetrically according ISO 6142-1:2015 and verified against SMU Primary standard gas mixtures in accordance to ISO 6143:2001. Impurities (hydrocarbons, oxygen, nitrogen, CO, CO₂) in parent gases were analysed on GC and FTIR. Each parent gas has its purity table with composition from purity measurements and manufacturer specifications.

The result values for parent gases and premixtures are given in following purity tables:

Table 1 CO Messer

Component	mole fraction (mol/mol)	uncertainty k=1 (mol/mol)
Ar	0.0000035	0.0000010
CO ₂	0.00000052	0.00000015
N ₂	0.000045	0.000013
O ₂	0.0000025	0.0000007
H ₂	0.00000050	0.00000014
H ₂ O	0.00000387	0.00000045
Methane	0.00000010	0.00000003
CO	0.999944	0.000013

Table 2 CO₂ Messer

Component	mole fraction (mol/mol)	uncertainty k=1 (mol/mol)
CO	0.00000028	0.000000010
H ₂ O	0.00000100	0.00000029
N ₂	0.000151	0.000041
O ₂	0.0000297	0.0000085
Methane	0.00000025	0.00000007
CO₂	0.999817	0.000042

Table 3 O₂ BIP

Component	mole fraction (mol/mol)	uncertainty k=1 (mol/mol)
CO	0.00000024	0.000000013
CO ₂	0.000000050	0.000000010
N ₂	0.00001499	0.00000081
H ₂ O	0.00000100	0.00000029
Methane	0.00000010	0.000000029
O₂	0.99998312	0.00000087

Table 4 N₂ BIP Plus

Component	mole fraction (mol/mol)	uncertainty k=1 (mol/mol)
CO	0.0000000050	0.00000000025
CO ₂	0.00000050	0.000000087
H ₂	0.000000025	0.0000000072
O ₂	0.000000011	0.0000000042
H ₂ O	0.000000010	0.0000000029
Methane	0.000000090	0.000000026
N₂	0.999999355	0.000000092

Table 5 Propane Messer

Component	mole fraction (mol/mol)	uncertainty k=1 (mol/mol)
O ₂	0.000138	0.000010
CO ₂	0.0000025	0.0000007
N ₂	0.0002385	0.0000301
H ₂ O	0.0000050	0.0000014
Methane	0.0000022	0.0000010
Ethylene	0.0000093	0.0000023
Ethane	0.0000537	0.0000026
iso-butane	0.0000518	0.0000019
Butane	0.00001057	0.00000043
1-butene	0.00000608	0.00000028
Hexane	0.00000103	0.00000014

Propane	0.999481	0.000032
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Table 6 CO premixture 0015F_12

Component	mole fraction (mol/mol)	uncertainty k=1 (mol/mol)
Argon	0.000000070	0.000000020
Methane	0.000000092	0.000000022
CO ₂	0.000000615	0.000000089
H ₂	0.000000121	0.000000030
H ₂ O	0.000000786	0.000000090
N ₂	0.7986653	0.0000039
O ₂	0.00000051	0.00000015
CO	0.2013319	0.0000053

Table 7 Propane premixture 0722E_7

Component	mole fraction (mol/mol)	uncertainty k=1 (mol/mol)
Methane	0.000000101	0.000000024
CO	0.0000000044	0.00000000002
CO ₂	0.000000648	0.000000095
Hydrogen	0.000000025	0.000000007
Water	0.000000035	0.000000008
Nitrogen	0.99496838	0.00000033
Oxygen	0.000000702	0.000000051
Ethane	0.000000270	0.000000013
iso-butane	0.000000261	0.000000010
Butane	0.000000053	0.000000002
Hexane	0.000000005	0.000000001
Ethylene	0.000000047	0.000000012
1-butene	0.000000031	0.000000001
Propane	0.00502944	0.00000025

Table 8 Propane premixture 0721E_6

Component	mole fraction (mol/mol)	uncertainty k=1 (mol/mol)
Methane	0.000000106	0.000000025
CO	0.0000000044	0.00000000002
CO ₂	0.000000653	0.000000095
Hydrogen	0.000000025	0.000000007
Water	0.000000047	0.000000011
Nitrogen	0.99250248	0.00000048
Oxygen	0.000001043	0.000000076
Ethane	0.000000403	0.000000020
iso-butane	0.000000389	0.000000014
Butane	0.000000079	0.000000003
Hexane	0.000000008	0.000000001
Ethylene	0.000000069	0.000000017

1-butene	0.000000046	0.000000002
Propane	0.00749465	0.00000037

Table 9 Propane premixture 0012F_4

Component	mole fraction (mol/mol)	uncertainty k=1 (mol/mol)
Methane	0.000000111	0.000000028
CO	0.0000000049	0.00000000002
CO ₂	0.000000524	0.000000087
Hydrogen	0.000000025	0.000000007
Water	0.000000060	0.000000015
Nitrogen	0.9900301	0.0000013
Oxygen	0.00000139	0.00000010
Ethane	0.000000535	0.000000026
iso-butane	0.000000517	0.000000019
Butane	0.000000105	0.000000004
Hexane	0.000000010	0.000000001
Ethylene	0.000000092	0.000000023
1-butene	0.000000061	0.000000003
Propane	0.0099665	0.0000013

Preparation of 3 Primary standard gas mixtures containing automotive gases with oxygen was carried out gravimetrically in 5 steps:

1. Propane in N₂, one of the three premixtures 0722E_7, 0721E_6, 0012F_4,
2. CO in N₂, premixture 00015F_12, or pure CO,
3. pure CO₂,
4. pure O₂,
5. pure N₂.

All mixtures were prepared in aluminium cylinders, V= 5 dm³. Inner surface of the cylinder was Aculife IV. Before preparation, the cylinder was evacuated at least 15 hours using dry evacuation system. The final value of vacuum was approximately 6x10⁻⁵ Pa.

The mass of added amount of parent mixture was determined by the difference of the cylinder masses before and after filling (line **m** in following tables). Weighting of evacuated or filled cylinder were executed on automatic SMU balance system. Filled cylinder mass was not determined absolutely, but as a difference (6 repetitions) between filled cylinder mass and reference cylinder mass. Mass scale reading for loaded cylinder (line **r** in following tables) is value obtained from loaded built-in weights (line **M** in following tables) and from arithmetic mean of the differences.

For the calculations of the gravimetric composition with associated uncertainties, validated spreadsheet xliISO6142 v1.11 was used based on the models described in ISO 6142-1:2015 and ISO 19229.

Weighing data of 3 Primary standard mixtures are shown below. The values of the added mass in the tables are in [g].

Table 10 0024F_6 16.03.2021

		0012F_4	CO	O ₂	CO ₂	N ₂
	Vacuum	SMU	Messer	Bip Plus	Messer	Bip Plus
Difference [1]	53.652	8.004	17.965	42.737	-0.324	26.134
Difference [2]	53.653	8.004	17.965	42.736	-0.324	26.135
Difference [3]	53.653	8.004	17.964	42.736	-0.325	26.135
Difference [4]	53.653	8.003	17.964	42.736	-0.325	26.136
Difference [5]	53.653	8.003	17.965	42.736	-0.324	26.136
Difference [6]	53.653	8.003	17.965	42.737	-0.324	26.136
M	0	53.5482	53.5482	53.5482	114.5305	571.9389

r	53.6526	61.5518	71.5128	96.2848	114.2069	598.0859
u(rept)	0.0004	0.0003	0.0002	0.0003	0.0002	0.0006
u(res)	0.0003	0.0003	0.0003	0.0003	0.0003	0.0003
u(M)	0.0000	0.0003	0.0003	0.0003	0.0003	0.0008
u(r)	0.0005	0.0005	0.0005	0.0005	0.0005	0.0036

m	-	7.8992	9.9609	24.7720	17.9221	483.8791
u(m)	-	0.0007	0.0007	0.0007	0.0007	0.0051
	-	0.0089%	0.0069%	0.0028%	0.0041%	0.0011%

Table 11 0059F_7 01.06.2021

		0722E_7	0015F_12	CO ₂	O ₂	N ₂
	Vacuum	SMU	SMU	Messer	Bip Plus	Bip Plus
Difference [1]	3.303	26.584	1.960	19.052	39.497	12.226
Difference [2]	3.301	26.585	1.961	19.052	39.497	12.227
Difference [3]	3.303	26.585	1.961	19.052	39.498	12.227
Difference [4]	3.305	26.585	1.961	19.053	39.498	12.227
Difference [5]	3.305	26.585	1.960	19.053	39.498	12.227
Difference [6]	3.305	26.585	1.961	19.053	39.498	12.226
M	0	0	53.5482	53.5482	53.5482	571.9389

r	3.3035	26.5854	55.5094	72.6007	93.0462	584.1773
u(rept)	0.0010	0.0003	0.0002	0.0005	0.0003	0.0002
u(res)	0.0003	0.0003	0.0003	0.0003	0.0003	0.0003
u(M)	0.0000	0.0000	0.0003	0.0003	0.0003	0.0008
u(r)	0.0010	0.0004	0.0005	0.0006	0.0005	0.0037

m	-	23.2819	28.9240	17.0913	20.4455	491.1311
u(m)	-	0.0011	0.0007	0.0008	0.0008	0.0052
	-	0.0047%	0.0024%	0.0048%	0.0041%	0.0011%

Table 12 0726E_7 24.06.2021

		0721E_6	CO	CO ₂	O ₂	N ₂
	Vacuum	SMU	Messer	Messer	Bip Plus	Bip Plus
Difference [1]	-16.113	3.249	14.727	-1.790	-29.255	26.556
Difference [2]	-16.112	3.250	14.727	-1.789	-29.255	26.556
Difference [3]	-16.110	3.251	14.728	-1.788	-29.254	26.557
Difference [4]	-16.109	3.251	14.729	-1.788	-29.254	26.558
Difference [5]	-16.108	3.252	14.729	-1.788	-29.254	26.559
Difference [6]	-16.108	3.253	14.729	-1.788	-29.254	26.560
M	0	0	0	53.5482	114.5305	571.9389

r	-					
u(rept)	16.1099	3.2511	14.7283	51.7604	85.2770	598.5088
u(res)	0.0013	0.0008	0.0006	0.0006	0.0005	0.0009
u(M)	0.0003	0.0003	0.0003	0.0003	0.0003	0.0003
u(r)	0.0000	0.0000	0.0000	0.0003	0.0003	0.0008
	0.0013	0.0009	0.0007	0.0007	0.0007	0.0040

m	-	19.3610	11.4771	37.0321	33.5166	513.2318
u(m)	-	0.0016	0.0011	0.0010	0.0010	0.0055
	-	0.0083%	0.0095%	0.0027%	0.0031%	0.0011%

For the verification of Primary standard mixtures GC method in accordance to ISO 6143:2001 was used calibration with SMU standard. All of the target components corresponded to the validation criteria. Neither stability changes were assumed for automotive gas mixtures. To the validated PRM's were assigned values of amount fraction of target component k (derived from the process of gravimetric preparation) and associated combined standard uncertainty calculated in accordance to the following formula (ISO 6142-1:2015):

$$u_{cert,x,k} = \frac{1}{2} \sqrt{u^2(x_{k,grav}) + u^2(x_{k,ver}) + (x_{k,grav} - x_{k,ver})^2}$$

Composition of calibration mixtures used in this comparison is shown in following table.

Table 13 SMU Primary calibration standards

Cylinder number	Component	x (mol/mol)	U _{cert} (k=2) (mol/mol)	U _{cert.rel} (k=2) (%)
0059F_7	Carbon monoxide	0.010180	0.000010	0.10
	Carbon dioxide	0.019016	0.000020	0.11
	Oxygen	0.031290	0.000084	0.27
	Propane	0.00020410	0.00000026	0.13
0024F_6	Carbon monoxide	0.018627	0.000036	0.19
	Carbon dioxide	0.021330	0.000030	0.14
	Oxygen	0.040552	0.000060	0.15
	Propane	0.00014637	0.00000032	0.22
0726E_7	Carbon monoxide	0.019229	0.000020	0.10
	Carbon dioxide	0.039487	0.000040	0.10
	Oxygen	0.049158	0.000080	0.16
	Propane	0.00024205	0.00000030	0.12

F.2 Instrumentation

GC method

For this key comparison, following equipment of Slovak national standard of amount fraction in gaseous phase was used for the verification of calibration standards and for analytical measurement of unknown sample:

Table 14 Equipment for automotive gas analysis

Equipment	Specifications
GC Agilent 7890A	
Columns set	Plot Q, Molsieve
Detectors	TCD, FID
electric or pneumatic valves	for dosing, shut-off for measured gas mixture
PC	control software for measurement on GC which records chromatograms to the PC
valve for gas mixture selection	1 output and min. 10 inputs, controlled from PC
distribution of gas (pipes, connections)	stainless steel, dimensions and threads 1/4', 1/8' Swagelok
regulations of outlet pressure	outlet pressure 2×10^2 kPa, stainless steel membrane input DIN-1, output 1/8' Swagelok
lines of pipes	stainless steel or FEP
measuring system of temperature, pressure and relative humidity of ambient air during measurement	resolution of: temperature: less than 0.05°C, pressure: less than 0.1 kPa, humidity: less than 0.2%

Analytical method GC settings

GC method parameters for automotive gas analysis are shown in the following tables:

Table 15 GC Settings

GC	L; cm ³	Oven temperature; °C	Total time; min	Inlet flow cm ³ /min
Agilent 7890A	1.0	10-220	13.83	20.4

Table 16 Gas flows

Gas	Flow ml/min
Carrier gas He	20.4
Make up N ₂	10
Make up He	5
Hydrogen	40
Air	300

Detectors:

TCD O₂, CO₂, CO

FID Propane

F.3 Calibration method and value assignment

GC measurement method with several automated runs was used. All runs in first, third, fifth measurement sequence had rising amount fraction. Second, fourth and sixth were processed in reverse order. From each run was made one calibration curve. 3 point linear model of analytical curve was used. No corrections were used.

Data were subjected to the B_least program (weighted least square regression). Inputs to the B_least were: values of measured signals (peak areas) with their standard uncertainties and amount fractions certified values of calibration gas mixtures with their standard uncertainties.

Uncertainty of instrument response consisted from figure characterized roughly immediate repeatability and from signal drift estimated. B-least program for each run produced sample mole fraction with its standard uncertainty. From all runs results = average of amount fractions in one sequence were standard deviation found $u_1(\bar{x}_i)$ and from runs results uncertainties the mean (through squares) was found $u_2(\bar{x}_i)$. These 2 figures were combined to give standard uncertainty $u_{an}(\bar{x}_i)$ for one sequence. Following formulas were used for the uncertainty calculations:

$$u_1(\bar{x}_i) = 1.3 * \sqrt{\frac{\sum_{j=1}^n (x_j - \bar{x}_i)^2}{n * (n-1)}}$$

$$u_2(\bar{x}_i) = \sqrt{\frac{\sum_{j=1}^n u(x_j)^2}{n^2}}$$

$$u_{an}(\bar{x}_i) = \sqrt{u_1(\bar{x}_i)^2 + u_2(\bar{x}_i)^2}$$

$$\bar{x}_i = \frac{\sum_{j=1}^n x_j}{n}$$

n - number of measuring runs

Final result of amount fraction was average value from 5 day results. Associated type A standard uncertainties were calculated as a standard deviation. The results are shown in the table 17.

Table 17 Measurement results with associated standard uncertainty type-A

Comp.	1. meas. (mol/mol)	2. meas. (mol/mol)	3. meas. (mol/mol)	4. meas. (mol/mol)	5. meas. (mol/mol)	\bar{x} (mol/mol)	u_A (mol/mol)
CO	0.0100097	0.0100091	0.0100066	0.0100134	0.0100032	0.0100084	0.0000017
CO ₂	0.0200365	0.0200286	0.0200278	0.0200270	0.0200157	0.0200271	0.0000033
Propane	0.000199213	0.000199048	0.000199010	0.000199047	0.000198960	0.000199056	0.000000043
Oxygen	0.0301203	0.0301172	0.0301168	0.0301080	0.0300935	0.0301112	0.0000049

Standard uncertainty type B was calculated as a combined value from maximum value from 5 day analysis uncertainties and uncertainty of primary standard calibration gas (0059F_7). Results are shown in the table 18.

Table 18 Type-B uncertainty evaluation results

Component	u_{an} 1. meas. (mol/mol)	u_{an} 2.meas. (mol/mol)	u_{an} 3. meas. (mol/mol)	u_{an} 4. meas. (mol/mol)	u_{an} 5. meas. (mol/mol)
CO	0.000014	0.000013	0.0000059	0.0000095	0.000010
CO ₂	0.000021	0.000028	0.0000080	0.000017	0.000015
Propane	0.00000027	0.00000018	0.00000011	0.00000013	0.00000017
Oxygen	0.000042	0.000051	0.000023	0.000031	0.000029

Component	$u_{an,max}$ (mol/mol)	u_{cert} (mol/mol)	u_B (mol/mol)
CO	0.000014	0.000005	0.000015
CO ₂	0.000028	0.000010	0.000030
Propane	0.00000027	0.00000013	0.00000030
Oxygen	0.000051	0.000042	0.000066

Total standard uncertainty was evaluated by following formula:

$$u(\bar{x}) = \sqrt{u_A^2 + u_B^2}$$

Expanded uncertainty (k=2) of final result:

$$U(\bar{x}) = 2 * u(\bar{x})$$

Table 19 Results

Component	Fraction (cmol/mol)	Expanded uncertainty (cmol/mol)	Coverage factor
Oxygen	3.011	0.013	2
CO ₂	2.0027	0.0060	2
Propane	0.019906	0.000061	2
CO	1.0008	0.0030	2

References

- ISO 6142-1:2015 Preparation of calibration gas mixtures – Part 1: Gravimetric method for Class I mixtures.
- ISO 6143:2001 Gas analysis– Comparison methods for determining and checking the composition of calibration gas mixtures.
- ISO 14912:2003 Gas analysis – Conversion of gas mixture composition data

Annex G Measurement report of METAS**G.1 C₃H₈:****G.1.1 Calibration standards**

Reference 1: Cylinder No. NPL 1420	METAS value (100.19 ± 0.40) µmol/mol
Reference 2: Cylinder No. Carbagas A1W24E9	METAS value (99.52 ± 0.40) µmol/mol
Reference 3: Cylinder No. Messer D358312	METAS value (100.40 ± 0.40) µmol/mol
Reference 4: Cylinder No. PanGas 75631100061351	METAS value (300.2 ± 1.2) µmol/mol
Reference 5: Cylinder No. Messer 80786	METAS value (299.5 ± 1.2) µmol/mol
Reference 6: Cylinder No. PanGas 75631110845534	METAS value (601.1 ± 2.4) µmol/mol
Reference 7: Cylinder No. Messer 6728F	METAS value (597.6 ± 2.4) µmol/mol

G.1.2 Instrumentation

A fully automatic pressure controlled Gas-Chromatograph with FID (Orthodyne S.A., Belgium) was used with an autosampler (Swagelok IGC-III, all gas lines are electropolished and pneumatically controlled).

Sample Handling: The sample flow through the sample loop of the injector is controlled at 60 ml/min @ 0 °C/1013 hPa, the pressure of the sample flow after the sample loop is also controlled at 1000 hPa absolute.

G.1.3 Calibration method and value assignment

The transfer standard has been compared against seven standard mixtures. These mixtures were cylinders out of the set of national reference gas mixtures for C₃H₈ in nitrogen in the range between 100 µmol/mol to 600 µmol/mol C₃H₈.

The area results (responses) of known mixtures and the unknown mixture have been evaluated using the bracketing technique with a linear regression according to ISO standard 6143 using the B-Least software.

G.2 CO:

G.2.1 Calibration standards

Reference 1: Cylinder No. Messer 80507	METAS value (0.29927 ± 0.0012) cmol/mol
Reference 2: Cylinder No. PanGas 220117	METAS value (0.29941 ± 0.0012) cmol/mol
Reference 3: Cylinder No. Messer 680	METAS value (0.8004 ± 0.0032) cmol/mol
Reference 4: Cylinder No. PanGas 101357	METAS value (0.8005 ± 0.0032) cmol/mol
Reference 5: Cylinder No. VSL D248683	METAS value (0.8003 ± 0.0032) cmol/mol
Reference 6: Cylinder No. Messer 24889	METAS value (1.5039 ± 0.0060) cmol/mol
Reference 7: Cylinder No. Carbagas A1EFUN2	METAS value (1.5003 ± 0.0060) cmol/mol
Reference 8: Cylinder No. Carbagas A1DNG69	METAS value (3.005 ± 0.012) cmol/mol
Reference 9: Cylinder No. PanGas 221510	METAS value (3.008 ± 0.012) cmol/mol

G.2.2 Instrumentation

A fully automatic pressure controlled Gas-Chromatograph with methaniser and FID (Orthodyne S.A., Belgium) was used with an autosampler (Swagelok IGC-III, all gas lines are electropolished and pneumatically controlled).

Sample Handling: The sample flow through the sample loop of the injector is controlled at 60 ml/min @ 0 °C/1013 hPa, the pressure of the sample flow after the sample loop is also controlled at 1000 hPa absolute.

G.2.3 Calibration method and value assignment

The transfer standard has been compared against nine standard mixtures. These mixtures were cylinders out of the set of national reference gas mixtures for CO in nitrogen in the range between 0.3 cmol/mol to 3.0 cmol/mol CO.

The area results (responses) of known mixtures and the unknown mixture have been evaluated using the bracketing technique with a linear regression according to ISO standard 6143 using the B-Least software.

G.3 O₂:

G.3.1 Calibration standards

Reference 1: Cylinder No. Messer D053973	METAS value (2.500 ± 0.010) cmol/mol
Reference 2: Cylinder No. PanGas 175920	METAS value (2.497 ± 0.010) cmol/mol
Reference 3: Cylinder No. VSL D247747	METAS value (2.503 ± 0.010) cmol/mol
Reference 4: Cylinder No. Messer 81263	METAS value (5.006 ± 0.020) cmol/mol
Reference 5: Cylinder No. Messer D492909	METAS value (4.992 ± 0.020) cmol/mol

G.3.2 Instrumentation

A paramagnetic measurement system (SIEMENS Oxymat 6E) was used with an autosampler (Swagelok IGC-III, all gas lines are electropolished and pneumatically controlled).

Sample Handling: The sample flow through the sample cell is controlled by a massflow controller at 400 ml/min @ 0 °C/1013 hPa.

G.3.3 Calibration method and value assignment

The transfer standard has been compared against five standard mixtures. These mixtures were cylinders out of the set of national reference gas mixtures for O₂ in nitrogen in the range between 2.5 cmol/mol to 5.0 cmol/mol O₂.

The signals (responses) of known mixtures and the unknown mixture have been evaluated using the bracketing technique with a linear regression according to ISO standard 6143 using the B-Least software.

G.4 CO₂:

G.4.1 Calibration standards

Reference 1: Cylinder No. Messer D055005	METAS value (9.013 ± 0.032) cmol/mol
Reference 2: Cylinder No. NPL 1485	METAS value (9.984 ± 0.035) cmol/mol
Reference 3: Cylinder No. PanGas 104777	METAS value (11.008 ± 0.039) cmol/mol

The premixtures were diluted with purified nitrogen (liquid nitrogen quality 99.999%, Purifier: SAES-Getter MonoTorr, CO₂ < 10 nmol/mol), using a METAS-calibrated massflow controller system for the flow measurements.

G.4.2 Instrumentation

A fully automatic pressure controlled Gas-Chromatograph with Methaniser and FID (Orthodyne S.A., Belgium) was used with an autosampler (Swagelok IGC-III, all gas lines are electropolished and pneumatically controlled).

Sample Handling: The sample flow through the sample loop of the injector is controlled at 80 ml/min @ 0 °C/1013 hPa, the pressure of the sample flow after the sample loop is also controlled at 1000 hPa absolute.

G.4.3 Calibration method and value assignment

The transfer standard has been compared against six mixtures by means of a fully automatic pressure controlled GC with methaniser and FID. The mixtures have been prepared by dilution of three gravimetric premixtures with nitrogen. The binary gravimetric premixtures were cylinders out of the set of national reference gas mixtures for CO₂ in nitrogen in the range between 9 cmol/mol to 11 cmol/mol.

A preliminary analysis of the transfer standard revealed an approximate amount of substance fraction of 2 cmol/mol. The calibration points for referencing the transfer standard readings have therefore been adapted. Two different dilutions have been produced for each cylinder. The area results (responses) of known calculated mixtures and the unknown mixture have been evaluated.

Detailed uncertainty budget:

The uncertainty estimation has been done using GUM Workbench Pro V. 2.4.1.406.

G.5 Example uncertainty evaluation

List of Quantities:

Quantity	Unit	Definition
AnzA1	au	first reference gas mixture: indicated measuring signal
AnzA2	au	second reference gas mixture: indicated measuring signal
AnzA3	au	third reference gas mixture: indicated measuring signal
AnzA4	au	fourth reference gas mixture: indicated measuring signal
AnzA5	au	fifth reference gas mixture: indicated measuring signal
AnzA6	au	sixth reference gas mixture: indicated measuring signal
AnzRes	au	DUT: indicated measuring signal
a	au	Axis intercept of the calibration line
b	au/(cmol/mol)	Slope of the calibration line

Quantity	Unit	Definition
p	au*cmol/mol	numerator for slope
q	(cmol/mol) ²	denominator for slope
Q _{MFC1n1}	ml/min@STP	first reference gas mixture: flow of the first gas mixture standard
Q _{MFC1n2}	ml/min@STP	second reference gas mixture: flow of the first gas mixture standard
Q _{MFC1n3}	ml/min@STP	third reference gas mixture: flow of the second gas mixture standard
Q _{MFC1n4}	ml/min@STP	fourth reference gas mixture: flow of the second gas mixture standard
Q _{MFC1n5}	ml/min@STP	fifth reference gas mixture: flow of the third gas mixture standard
Q _{MFC1n6}	ml/min@STP	sixth reference gas mixture: flow of the third gas mixture standard
Q _{MFC2n1}	ml/min@STP	first reference gas mixture: nitrogen flow
Q _{MFC2n2}	ml/min@STP	second reference gas mixture: nitrogen flow
Q _{MFC2n3}	ml/min@STP	third reference gas mixture: nitrogen flow
Q _{MFC2n4}	ml/min@STP	fourth reference gas mixture: nitrogen flow
Q _{MFC2n5}	ml/min@STP	fifth reference gas mixture: nitrogen flow
Q _{MFC2n6}	ml/min@STP	sixth reference gas mixture: nitrogen flow
X _{Ref1}	cmol/mol	first gas mixture standard: amount fraction
X _{Ref2}	cmol/mol	second gas mixture standard: amount fraction
X _{Ref3}	cmol/mol	third gas mixture standard: amount fraction
X _N	cmol/mol	nitrogen: residual CO ₂ (amount fraction)
X _{A1}	cmol/mol	first reference gas mixture: amount fraction in final mixture
X _{A2}	cmol/mol	second reference gas mixture: amount fraction in final mixture
X _{A3}	cmol/mol	third reference gas mixture: amount fraction in final mixture
X _{A4}	cmol/mol	fourth reference gas mixture: amount fraction in final mixture
X _{A5}	cmol/mol	fifth reference gas mixture: amount fraction in final mixture
X _{A6}	cmol/mol	sixth reference gas mixture: amount fraction in final mixture
X _{meanA}	cmol/mol	mean CO ₂ amount fraction of the reference gas mixtures
Anz _{meanA}	au	mean indicated measuring signal of the reference gas mixtures
XC02Res	cmol/mol	DUT: amount fraction

Uncertainty Budget:

XC02Res: DUT: amount fraction

Quantity	Value	Standard Uncert. (rel.)	df	Sens.-Coeff.	Uncert.-Contrib.	Corr.-Coeff.	Index
Anz _{A1}	13.03382·10 ⁶ au	0.033 %	4	-22·10 ⁻⁹	-93·10 ⁻⁶ cmol/mol	-0.0266	0.0 %
Anz _{A2}	14.41982·10 ⁶ au	0.025 %	4	-26·10 ⁻⁹	-95·10 ⁻⁶ cmol/mol	-0.0272	0.0 %
Anz _{A3}	12.98667·10 ⁶ au	0.016 %	4	-22·10 ⁻⁹	-45·10 ⁻⁶ cmol/mol	-0.0128	0.0 %
Anz _{A4}	14.38665·10 ⁶ au	0.009 %	4	-26·10 ⁻⁹	-34·10 ⁻⁶ cmol/mol	-9.7·10 ⁻³	0.0 %
Anz _{A5}	13.03161·10 ⁶ au	0.022 %	4	-22·10 ⁻⁹	-61·10 ⁻⁶ cmol/mol	-0.0174	0.0 %

Quantity	Value	Standard Uncert. (rel.)	df	Sens.- Coeff.	Uncert.- Contrib.	Corr.- Coeff.	Index
Anz _{A6}	14.40962·10 ⁶ au	0.030 %	4	-26·10 ⁻⁹	-110·10 ⁻⁶ cmol/mol	-0.0322	0.1 %
AnzRes	13.77692·10 ⁶ au	0.019 %	4	140·10 ⁻⁹	380·10 ⁻⁶ cmol/mol	0.1083	1.2 %
a	-194·10 ³ au	68 %	130				
b	6.9689·10 ⁶ au/(cmol/mol)	0.97 %	140				
p	414.93·10 ³ au*cmol/mol	0.97 %	140				
q	0.05954 (cmol/mol) ²	1.9 %	130				
Q _{MFC1n1}	16.8890 ml/min@STP	0.25 %	100	0.013	570·10 ⁻⁶ cmol/mol	0.2146	3.5 %
Q _{MFC1n2}	18.6670 ml/min@STP	0.25 %	100	0.016	740·10 ⁻⁶ cmol/mol	0.2243	4.7 %
Q _{MFC1n3}	15.2000 ml/min@STP	0.25 %	100	0.015	580·10 ⁻⁶ cmol/mol	0.2153	3.5 %
Q _{MFC1n4}	16.8000 ml/min@STP	0.25 %	100	0.018	750·10 ⁻⁶ cmol/mol	0.2252	4.8 %
Q _{MFC1n5}	13.8180 ml/min@STP	0.25 %	100	0.017	590·10 ⁻⁶ cmol/mol	0.2162	3.7 %
Q _{MFC1n6}	15.2730 ml/min@STP	0.25 %	100	0.020	780·10 ⁻⁶ cmol/mol	0.2266	5.0 %
Q _{MFC2n1}	63.328 ml/min@STP	0.20 %	100	-3.6·10 ⁻³	-450·10 ⁻⁶ cmol/mol	0.1565	-2.0 %
Q _{MFC2n2}	61.566 ml/min@STP	0.20 %	100	-4.8·10 ⁻³	-590·10 ⁻⁶ cmol/mol	0.1488	-2.5 %
Q _{MFC2n3}	65.001 ml/min@STP	0.20 %	100	-3.6·10 ⁻³	-460·10 ⁻⁶ cmol/mol	0.1560	-2.1 %
Q _{MFC2n4}	63.416 ml/min@STP	0.20 %	100	-4.7·10 ⁻³	-600·10 ⁻⁶ cmol/mol	0.1481	-2.5 %
Q _{MFC2n5}	66.370 ml/min@STP	0.20 %	100	-3.6·10 ⁻³	-470·10 ⁻⁶ cmol/mol	0.1553	-2.1 %
Q _{MFC2n6}	64.929 ml/min@STP	0.20 %	100	-4.8·10 ⁻³	-620·10 ⁻⁶ cmol/mol	0.1470	-2.6 %
X _{Ref1}	9.0130 cmol/mol	0.20 %	100	0.074	1.3·10 ⁻³ cmol/mol	0.7626	29.1 %
X _{Ref2}	9.9840 cmol/mol	0.20 %	100	0.067	1.3·10 ⁻³ cmol/mol	0.7610	28.8 %
X _{Ref3}	11.0080 cmol/mol	0.20 %	100	0.061	1.3·10 ⁻³ cmol/mol	0.7625	29.1 %
X _N	200.0·10 ⁻⁶ cmol/mol	41 %	∞	0.80	65·10 ⁻⁶ cmol/mol	0.0186	0.0 %
X _{A1}	1.89777 cmol/mol	0.23 %	160				
X _{A2}	2.09712 cmol/mol	0.23 %	160				
X _{A3}	1.89237 cmol/mol	0.23 %	170				
X _{A4}	2.09115 cmol/mol	0.23 %	160				

Quantity	Value	Standard Uncert. (rel.)	df	Sens.-Coeff.	Uncert.-Contrib.	Corr.-Coeff.	Index
X _{A5}	1.89706 cmol/mol	0.24 %	170				
X _{A6}	2.09643 cmol/mol	0.23 %	170				
X _{meanA}	1.99532 cmol/mol	0.17 %	190				
Anz _{meanA}	13.71137·10 ⁶ au	0.01 %	17				
XCO2Res	2.00472 cmol/mol	0.17 %	190				

Result:

Quantity	Value	Expanded Uncertainty	Coverage factor	Coverage
XCO2Res	2.0047 cmol/mol	0.35 % (relative)	2.00	95% (t-table 95.45%)

Annex H Measurement report of UME

Cylinder code : D751979

H.1 Calibration standards

Primary reference gas mixtures of automotive exhaust gases mixtures, used in calibrations are given in Table 1. They were prepared individually in accordance with ISO 6142-1:2015 [1] at UME and were verified. Several pre-mixtures were individually prepared from pure components; these pre-mixtures were diluted to prepare three reference gas standards for automotive exhaust gases mixtures. The pure gases used for the preparation of gas mixtures are given in Table 2. The uncertainties of reference gas mixtures were determined by combining the standard uncertainties of weighing, purity and molecular masses for all components.

Table 1. List of primary reference gas mixtures

Item	Prepared By	Cylinder Number	Component	Mole Fraction (cmol/mol)	Uncertainty (k=1) (cmol/mol)
1	UME	PSM499770	Oxygen	2.8146	0.0007
			Carbon dioxide	1.8777	0.0005
			Propane	0.019243	0.000005
			Carbon monoxide	0.8944	0.0001
			Nitrogen	Balance	
2	UME	PSM499763	Oxygen	2.9696	0.0007
			Carbon dioxide	2.0209	0.0005
			Propane	0.019980	0.000005
			Carbon monoxide	0.9997	0.0001
			Nitrogen	Balance	
3	UME	PSM298280	Oxygen	3.2498	0.0007
			Carbon dioxide	2.0636	0.0005
			Propane	0.021053	0.000005
			Carbon monoxide	1.1001	0.0001
			Nitrogen	Balance	

Table 2. List of pure gases

Component	Suppliers	Grade	Origin
Nitrogen	Linde	6.0	Turkey
Oxygen	Linde	5.0	Germany
Carbon dioxide	Linde	5.5	Turkey
Propane	Air Liquide	3.5	Germany
Carbon monoxide	Hat Group	4.7	Germany

H.2 Instrumentation

The automotive exhaust gases mixtures were analyzed on an Agilent 6890N gas chromatography instrument equipped with FID and TCD, split/splitless injector, gas injection valve, including GC ChemStation software (Rev. B. 04.03-SP2 [108]) to collect and process data. Carbon monoxide, carbon dioxide and propane in the mixtures were analyzed on Agilent 6890N gas chromatography with FID using Helium as carrier gas. Oxygen in the mixtures was also analyzed on Agilent 6890N gas chromatography with TCD using Argon as carrier gas. The conditions for the analyses are given below:

a) Conditions for Agilent 6890N (for CO, CO₂ and C₃H₈)

Oven

Equilibration Time : 0.30 min
Initial Temperature : On 50°C
Max Temperature : 300°C
Oven Program : On
#1 then 100 °C/min to 100 °C for 2.5 min
Run Time : 13 min

Front SS Inlet He

Mode : Split
Heater : On 150 °C
Pressure : On 18.77 psi
Total Flow : On 210.0 mL/min
Split Ratio : 25:1
Split Flow : 199.6 mL/min
Gas Type : Helium

Column #1

Agilent PLOTQ+ Molesieve + Agilent 19095P-Q04+PerkinNR210040

Max Temperature : 270°C
Nominal length : 60.0 m
Nominal diameter : 530.00 um
Nominal film thickness : 50.00 um
Mode : Constant flow
Initial flow : 8.0 mL/min
Nominal init pressure : 18.78 psi
Average velocity : 58 cm/sec
Inlet : Front Inlet
Outlet : Back Detector

Outlet pressure : Ambient

Front Detector FID

Heater : On 250 °C
H₂ Flow : On 40.0 mL/min
Air Flow : On 400 mL/min
Makeup Flow : Off 45.0 mL/min
Makeup Gas Type : Helium
Flame : On
Electrometer : On
Data rate : 20 Hz

Back Detector TCD

Heater : Off 250 °C
Reference Flow : Off 15 mL/min
Makeup Flow : Off 8 mL/min
Makeup Gas Type : Helium
Filament : Off
Negative Polarity : Off

b) Conditions for Agilent 6890N (for O₂)

Oven

Equilibration Time : 0.30 min
Initial Temperature : On 50°C
Max Temperature : 300°C
Oven Program : On
Initial Time : 5.5 min
#1 then 25 °C/min to 150 °C for 1.0 min
Run Time : 10.5 min

Front SS Inlet He

Mode : Split
Heater : On 150 °C
Pressure : On 20.8 psi
Total Flow : On 17.8 mL/min
Split Ratio : 1:1
Split Flow : 7.9 mL/min
Gas Type : Argon methane 5%

Column #1

Agilent PLOTQ+ Molesieve + Agilent 19095P-Q04+PerkinNR210040

Max Temperature : 270°C
Nominal length : 60.0 m
Nominal diameter : 530.00 um
Nominal film thickness : 50.00 um
Mode : Constant flow
Initial flow : 7.9 mL/min
Nominal init pressure : 20.81 psi
Average velocity : 55 cm/sec
Inlet : Front Inlet
Outlet : Back Detector
Outlet pressure : Ambient

Front Detector FID

Heater	: Off 250 °C
H ₂ Flow	: Off 40.0 mL/min
Air Flow	: Off 400 mL/min
Makeup Flow	: Off 45.0 mL/min
Makeup Gas Type	: Helium
Flame	: Off
Electrometer	: Off

Back Detector TCD

Heater	: On 250 °C
Reference Flow	: On 15 mL/min
Makeup Flow	: On 8 mL/min
Makeup Gas Type	: Helium
Filament	: On
Negative Polarity	: On
Data rate	: 5Hz

H.3 Calibration method and value assignment

After the arrival of the cylinder from coordinator, it was stored in the laboratory at least 24 hours prior to analyses. Reference gas mixtures were also stored in the same laboratory during all measurements. Sample and the calibration standards were equipped with pressure reducers and connected to computer programmed multiposition valve gas sampling box. They were flushed at least 3 times before the first measurement. The pressures of sample and reference gases were controlled by an electronic pressure controller at 150 mbar.

The data was collected using ChemStation software. Each sample in the sequence was injected 9 times, and the first two injections in each case were discarded as they were considered as flushing of sample loop. The responses were averaged. The software "CurveFit" was utilized to determine the fitting data for the calibrations. The value for goodness of fit in each measurement was found to be less than 2 in each case for linear function.

The assigned value was calculated by averaging the results of four independent sets of measurements.

H.4 Uncertainty evaluation

The measurement uncertainty of sample was determined according to ISO 6143:2001 [2], using the CurveFit software.

The largest uncertainty of the four sets of measurements for each component was taken as the standard uncertainty in the sample for the corresponding component.

The expanded uncertainty was determined by multiplying the standard uncertainty by a coverage factor of 2 with a confidence interval of 95%.

References

- [1] International Organization for Standardization, ISO 6142-1:2015 "Gas analysis - Preparation of calibration gas mixtures - Part 1: Gravimetric method for Class I mixtures"
- [2] International Organization for Standardization, ISO 6143:2001 "Gas analysis - Comparison methods for determining and checking the composition of calibration gas mixtures"

Annex I Measurement report of CMI

I.1 Calibration standards

All mixtures were prepared in aluminium cylinders, $V = 5 \text{ dm}^3$. Before preparation, the cylinder was evacuated at least 12 hours using dry evacuation system. The mass of added amount was determined by the difference of the cylinder masses before and after filling. Weighting of evacuated or filled cylinder were executed on automatic balance system "Kalipro 15". During automatic weighting, the filled cylinder mass was not determined absolutely, but as a difference between filled cylinder mass and reference cylinder mass.

RM number:		D036624
compounds:	x_i	$U_{(xi)}$
	$(10^{-2} \text{ mol/mol})$	
propane	0.73	1.80
CO	0.079	0.001
CO ₂	0.018	0.001
Oxygen	0.03	0.001

RM number:		D036625
compounds:	x_i	$U_{(xi)}$
	$(10^{-2} \text{ mol/mol})$	
propane	390.51	1.83
CO	2.0046	0.001
CO ₂	5.0653	0.001
Oxygen	4.0038	0.001

RM number:		D036641
compounds:	x_i	$U_{(xi)}$
	$(10^{-2} \text{ mol/mol})$	
propane	102.76	1.76
CO	0.5003	0.001
CO ₂	2.0815	0.001
Oxygen	1.0044	0.001

RM number:		20427
compounds:	x_i	$U_{(xi)}$
	$(10^{-2} \text{ mol/mol})$	
propane	178.12	1.81
CO	1.007	0.001
CO ₂	3.0214	0.001
Oxygen	2.0108	0.001

I.2 Instrumentation

Dani Master (gas chromatograph with 2x TCD and FID)

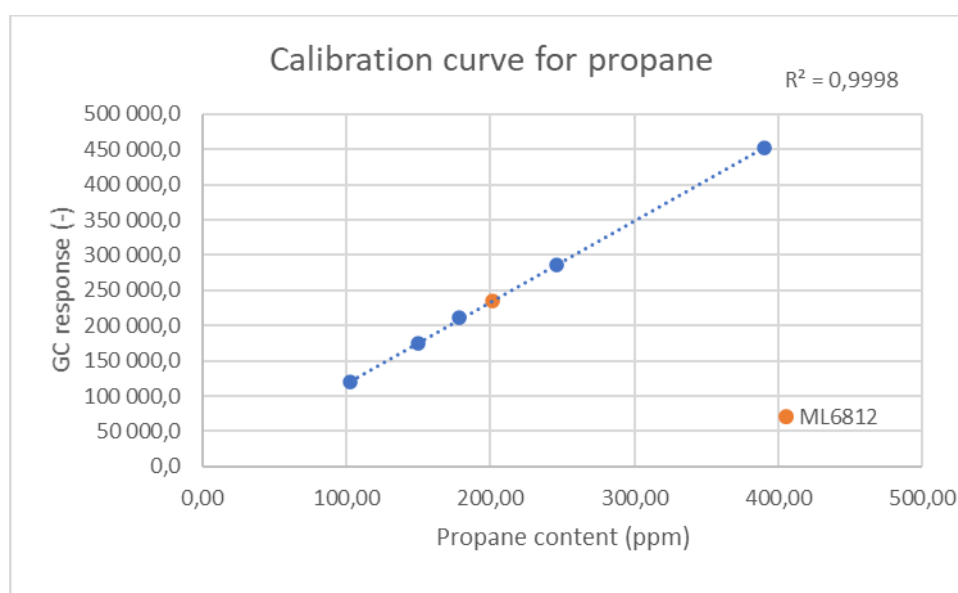
- Owen temperature: 60°C, isothermal,

Carrier gas:	sample loop:	Detector:	
hydrogen	1ml	FID	MTX-1 60m x 0,53mm
helium	1ml	TCD	HS-Q 80/100 3m x 1/16" x 1,0 mm
			MS5A 80/100 3m x 1/16" x 1,0 mm

I.3 Calibration method and value assignment

GC measurement method with automated runs was used. Four-point linear model of analytical curve was used.

Four calibration mixtures were prepared for building of the linear calibration curve. For the verification the GC method was used. All of target components corresponded to the validation criteria. The values of mole fraction of target component were assigned to the validated PRM's (derived from the process of gravimetric preparation).



I.4 Uncertainty evaluation:

$$u_a = \text{Rel. std. dev (\%)} = \sqrt{\frac{\sum_{j=1}^n (x_j - x_i)^2}{N \cdot (N - 1)}} * 100$$

$$u_{RM}(x_i) = \sqrt{\sum_{j=1}^m u_{RMj}^2}$$

$$u_c(x_i) = \sqrt{u_a(x_i)^2 + u_b(x_i)^2}$$

$$U_{c(xi)} = k \cdot u_{c(xi)}$$

Where:

- u_a - uncertainty from measurement
- $u_{RM(xi)}$ - uncertainty reference materials
- u_{xi} - standard uncertainty compounds x_i
- U_{xi} - extended standard uncertainty compounds x_i
- k - coverage factor,
- x_i - molar fraction of component i
- N - number of measurements

References:

ISO 6142-1:2015 Preparation of calibration gas mixtures – Part 1: Gravimetric method for Class I mixtures.