

Guide to the Realization of the ITS-90

Platinum Resistance Thermometry

APPENDIX 1: Alternative interpolating functions for special applications



Consultative Committee for Thermometry
under the auspices of the
International Committee for Weights and Measures

Appendix 1

Alternative interpolating functions for special applications

The following subsections list alternative interpolating functions for each of the ITS-90 subranges covered by this chapter on platinum resistance thermometry. Each subsection presents the ITS-90 equation, formulae for the interpolating functions, and a graph showing the interpolating functions as a function of temperature. All of the graphs are expressed in temperature rather than resistance ratio; both axes are scaled so that the graphs plot the propagated temperature error divided by the temperature error at the specified fixed point.

Water-Gallium Subrange: 0.01 °C to 29.7646 °C

The ITS-90 equation for this subrange is

$$W_r(W) = W - a(W - 1), \quad (\text{A1.1})$$

where the a coefficient is determined by calibration of the SPRT at the melting point of gallium. The interpolation is rewritten as

$$W_r(W) = \frac{(W - W_{\text{Ga}})}{(1 - W_{\text{Ga}})} + W_{r,\text{Ga}} \frac{(W - 1)}{(W_{\text{Ga}} - 1)}, \quad (\text{A1.2})$$

from which the sensitivity coefficients are identified as

$$f_{\text{H}_2\text{O}}(W) = \frac{(W - W_{\text{Ga}})}{(1 - W_{\text{Ga}})} \quad (\text{A1.3a})$$

and

$$f_{\text{Ga}}(W) = \frac{(W - 1)}{(W_{\text{Ga}} - 1)}. \quad (\text{A1.3b})$$

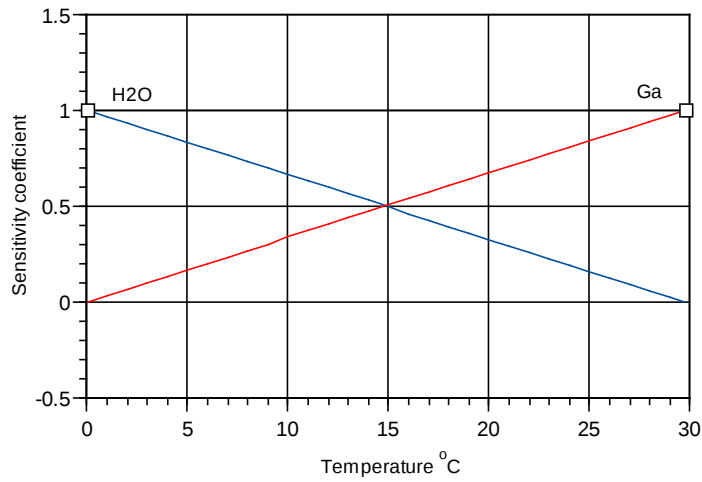


Figure A1.1: Sensitivity coefficients for the water-gallium subrange. The curves show the propagated error in millikelvin for a 1 mK error at each of the fixed points.

Water-Indium Subrange: 0.01 °C to 156.5985 °C

The ITS-90 interpolation equation is

$$W_r(W) = W - a(W - 1), \quad (\text{A1.4})$$

where the a coefficient is determined by calibration of the SPRT at the freezing point of indium. The interpolation equation can be written in the form

$$W_r(W) = \frac{(W - W_{\text{In}})}{(1 - W_{\text{In}})} + W_{\text{r,In}} \frac{(W - 1)}{(W_{\text{In}} - 1)}, \quad (\text{A1.5})$$

from which the sensitivity coefficients are identified as

$$f_{\text{H}_2\text{O}}(W) = \frac{(W - W_{\text{In}})}{(1 - W_{\text{In}})} \quad (\text{A1.6a})$$

and

$$f_{\text{In}}(W) = \frac{(W - 1)}{(W_{\text{In}} - 1)}. \quad (\text{A1.6b})$$

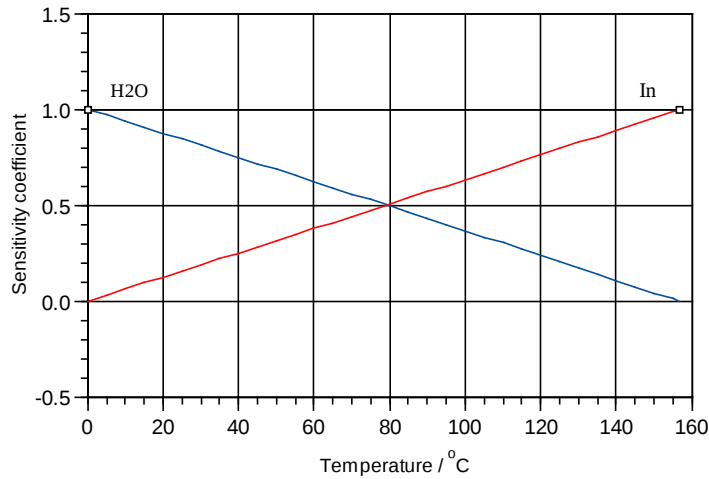


Figure A1.2: The sensitivity coefficients for the water-indium subrange. The curves show the propagated error in millikelvin for a 1 mK error at each of the fixed points.

Mercury-Gallium Subrange: –38.8344 °C to 29.7646 °C

The ITS-90 interpolation equation is

$$W_r(W) = W - a(W - 1) - b(W - 1)^2, \quad (\text{A1.7})$$

where the a and b coefficients are determined by calibration of the SPRT at the triple point of mercury and the melting point of gallium. The interpolation equation may be written in the form

$$W_r(W) = W_{r,\text{Hg}} \frac{(W - 1)(W - W_{\text{Ga}})}{(W_{\text{Hg}} - 1)(W_{\text{Hg}} - W_{\text{Ga}})} + \frac{(W - W_{\text{Hg}})(W - W_{\text{Ga}})}{(1 - W_{\text{Hg}})(1 - W_{\text{Ga}})} + W_{r,\text{Ga}} \frac{(W - W_{\text{Hg}})(W - 1)}{(W_{\text{Ga}} - W_{\text{Hg}})(W_{\text{Ga}} - 1)}, \quad (\text{A1.8})$$

from which the sensitivity coefficients are identified as

$$f_{\text{Hg}} = \frac{(W - 1)(W - W_{\text{Ga}})}{(W_{\text{Hg}} - 1)(W_{\text{Hg}} - W_{\text{Ga}})}, \quad (\text{A1.9a})$$

$$f_{\text{H}_2\text{O}} = \frac{(W - W_{\text{Hg}})(W - W_{\text{Ga}})}{(1 - W_{\text{Hg}})(1 - W_{\text{Ga}})}, \quad (\text{A1.9b})$$

and

$$f_{\text{Ga}} = \frac{(W - W_{\text{Hg}})(W - 1)}{(W_{\text{Ga}} - W_{\text{Hg}})(W_{\text{Ga}} - 1)}. \quad (\text{A1.9c})$$

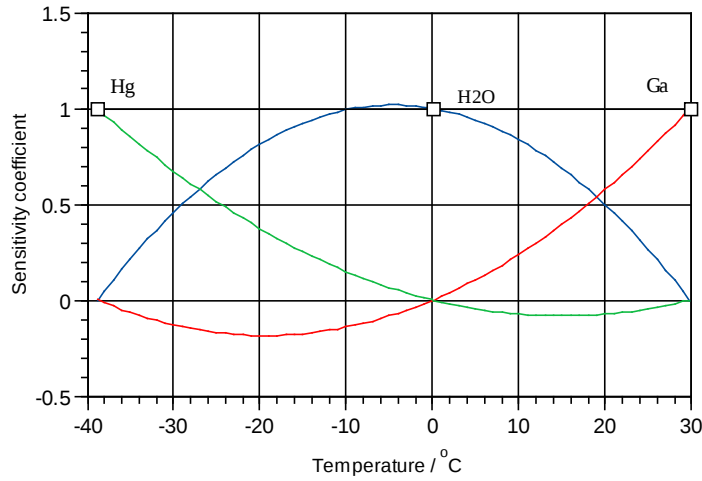


Figure A1.3: The sensitivity coefficients for the mercury-gallium subrange. The curves show the propagated error in millikelvin for a 1 mK error at each of the fixed points.

Water-Tin Subrange: 0.01 °C to 231.928 °C

The ITS-90 interpolation equation is

$$W_r(W) = W - a(W - 1) - b(W - 1)^2, \quad (\text{A1.10})$$

where the a and b coefficients are determined by calibration of the SPRT at the freezing points of indium and tin. The interpolation equation may be written in the form

$$W_r(W) = \frac{(W - W_{\text{In}})(W - W_{\text{Sn}})}{(1 - W_{\text{In}})(1 - W_{\text{Sn}})} + W_{\text{r,In}} \frac{(W - 1)(W - W_{\text{Sn}})}{(W_{\text{In}} - 1)(W_{\text{In}} - W_{\text{Sn}})} + W_{\text{r,Sn}} \frac{(W - 1)(W - W_{\text{In}})}{(W_{\text{Sn}} - 1)(W_{\text{Sn}} - W_{\text{In}})} \quad (\text{A1.11})$$

from which the sensitivity coefficients are identified as

$$f_{\text{H2O}} = \frac{(W - W_{\text{In}})(W - W_{\text{Sn}})}{(1 - W_{\text{In}})(1 - W_{\text{Sn}})}, \quad (\text{A1.12a})$$

$$f_{\text{In}} = \frac{(W - 1)(W - W_{\text{Sn}})}{(W_{\text{In}} - 1)(W_{\text{In}} - W_{\text{Sn}})}, \quad (\text{A1.12b})$$

and

$$f_{\text{Sn}} = \frac{(W - 1)(W - W_{\text{In}})}{(W_{\text{Sn}} - 1)(W_{\text{Sn}} - W_{\text{In}})}. \quad (\text{A1.12c})$$

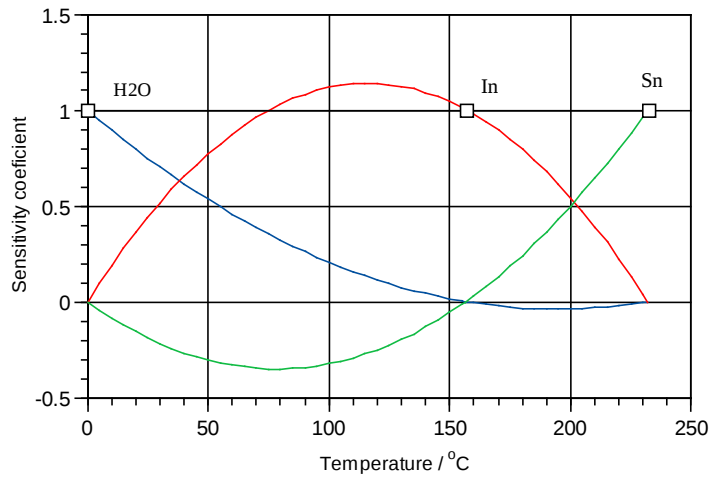


Figure A1.4: The sensitivity coefficients for the water-tin subrange. The curves show the propagated error in millikelvin for a 1 mK uncertainty at each of the fixed points.

Water-Zinc Subrange: 0.01 °C to 419.527 °C

The ITS-90 interpolation equation is

$$W_r(W) = W - a(W - 1) - b(W - 1)^2, \quad (\text{A1.13})$$

where the a and b coefficients are determined by calibration of the SPRT at the freezing points of tin and zinc. The interpolation equation may be written in the form

$$W_r(W) = \frac{(W - W_{\text{Sn}})(W - W_{\text{Zn}})}{(1 - W_{\text{Sn}})(1 - W_{\text{Zn}})} + W_{r,\text{Sn}} \frac{(W - 1)(W - W_{\text{Zn}})}{(W_{\text{Sn}} - 1)(W_{\text{Sn}} - W_{\text{Zn}})} + W_{r,\text{Zn}} \frac{(W - 1)(W - W_{\text{Sn}})}{(W_{\text{Zn}} - 1)(W_{\text{Zn}} - W_{\text{Sn}})}, \quad (\text{A1.14})$$

from which the sensitivity coefficients are identified as

$$f_{\text{H2O}} = \frac{(W - W_{\text{Sn}})(W - W_{\text{Zn}})}{(1 - W_{\text{Sn}})(1 - W_{\text{Zn}})}, \quad (\text{A1.15a})$$

$$f_{\text{Sn}} = \frac{(W - 1)(W - W_{\text{Zn}})}{(W_{\text{Sn}} - 1)(W_{\text{Sn}} - W_{\text{Zn}})}, \quad (\text{A1.15b})$$

and

$$f_{\text{Zn}} = \frac{(W - 1)(W - W_{\text{Sn}})}{(W_{\text{Zn}} - 1)(W_{\text{Zn}} - W_{\text{Sn}})}. \quad (\text{A1.15c})$$

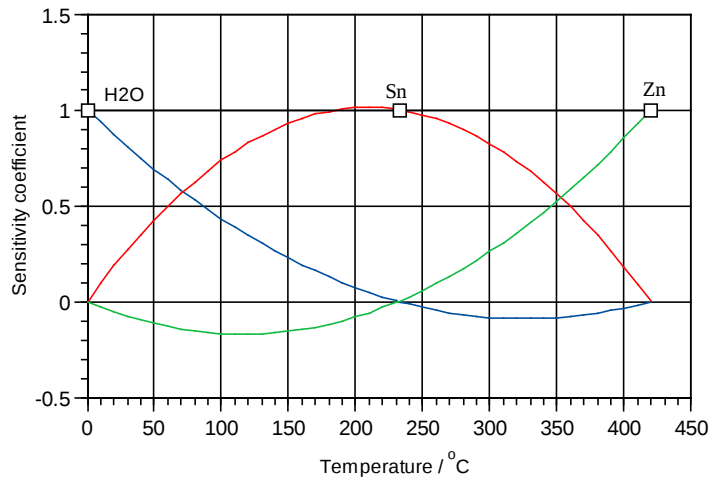


Figure A1.5: The sensitivity coefficients for the water-zinc subrange. The curves show the propagated error in millikelvin for a 1 mK error at each of the fixed points.

Water-Aluminium Subrange: 0.01 °C to 660.323 °C

The ITS-90 interpolation equation is

$$W_r(W) = W - a(W - 1) - b(W - 1)^2 - c(W - 1)^3, \quad (\text{A1.16})$$

where the a , b and c coefficients are determined by calibration of the SPRT at the freezing points of tin, zinc and aluminium. The interpolation equation may be written in the form

$$W_r(W) = f_{\text{H}_2\text{O}} + W_{r,\text{Sn}} f_{\text{Sn}} + W_{r,\text{Zn}} f_{\text{Zn}} + W_{r,\text{Al}} f_{\text{Al}}, \quad (\text{A1.17})$$

where the sensitivity coefficients are

$$f_{\text{H}_2\text{O}} = \frac{(W - W_{\text{Sn}})(W - W_{\text{Zn}})(W - W_{\text{Al}})}{(1 - W_{\text{Sn}})(1 - W_{\text{Zn}})(1 - W_{\text{Al}})}, \quad (\text{A1.18a})$$

$$f_{\text{Sn}} = \frac{(W - 1)(W - W_{\text{Zn}})(W - W_{\text{Al}})}{(W_{\text{Sn}} - 1)(W_{\text{Sn}} - W_{\text{Zn}})(W_{\text{Sn}} - W_{\text{Al}})}, \quad (\text{A1.18b})$$

$$f_{\text{Zn}} = \frac{(W - 1)(W - W_{\text{Sn}})(W - W_{\text{Al}})}{(W_{\text{Zn}} - 1)(W_{\text{Zn}} - W_{\text{Sn}})(W_{\text{Zn}} - W_{\text{Al}})}, \quad (\text{A1.18c})$$

$$f_{\text{Al}} = \frac{(W - 1)(W - W_{\text{Sn}})(W - W_{\text{Zn}})}{(W_{\text{Al}} - 1)(W_{\text{Al}} - W_{\text{Sn}})(W_{\text{Al}} - W_{\text{Zn}})}. \quad (\text{A1.18d})$$

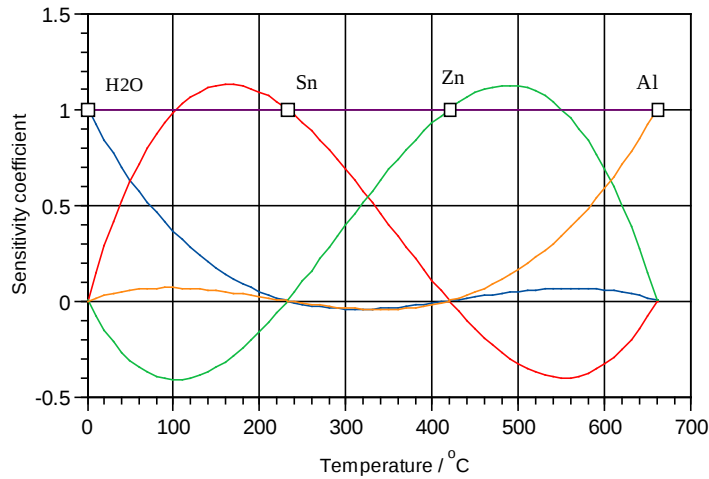


Figure A1.6: The sensitivity coefficients for the water-aluminium subrange. The curves show the propagated error in millikelvin for a 1 mK error at each of the fixed points.

Water-Silver Subrange: 0.01 °C to 961.78 °C

The ITS-90 interpolation equation takes the form

$$W_r(W) = W - a(W - 1) - b(W - 1)^2 - c(W - 1)^3 - d(W - W_{Al})^2, \quad (\text{A1.19})$$

where the a , b and c coefficients are the same as determined from the water-aluminium subrange, and the d coefficient is determined by calibration of the SPRT at the freezing point of silver. The equation is applicable only over the temperature range between the aluminium and silver points (660.323 °C to 961.78 °C). The interpolation equation may be written in the form

$$W_r(W) = f_{H_2O} + W_{r,Sn} f_{Sn} + W_{r,Zn} f_{Zn} + W_{r,Al} f_{Al} + W_{r,Ag} f_{Ag}, \quad (\text{A1.20})$$

where the sensitivity coefficients (applicable only above 660.323 °C) are

$$f_{H_2O} = f_{H_2O}^{Al}(W) - f_{H_2O}^{Al}(W_{Ag}) \left(\frac{W - W_{Al}}{W_{Ag} - W_{Al}} \right)^2, \quad (\text{A1.21a})$$

$$f_{Sn} = f_{Sn}^{Al}(W) - f_{Sn}^{Al}(W_{Ag}) \left(\frac{W - W_{Al}}{W_{Ag} - W_{Al}} \right)^2, \quad (\text{A1.21b})$$

$$f_{Zn} = f_{Zn}^{Al}(W) - f_{Zn}^{Al}(W_{Ag}) \left(\frac{W - W_{Al}}{W_{Ag} - W_{Al}} \right)^2, \quad (\text{A1.21c})$$

$$f_{Al} = f_{Al}^{Al}(W) - f_{Al}^{Al}(W_{Ag}) \left(\frac{W - W_{Al}}{W_{Ag} - W_{Al}} \right)^2, \quad (\text{A1.21d})$$

$$f_{Ag} = \left(\frac{W - W_{Al}}{W_{Ag} - W_{Al}} \right)^2, \quad (\text{A1.21e})$$

where the f_i^{Al} are the sensitivity coefficients for the water-aluminium subrange, as given by (A1.18a-d). For temperatures below 660.323 °C, Equations (A1.18a-d) apply.

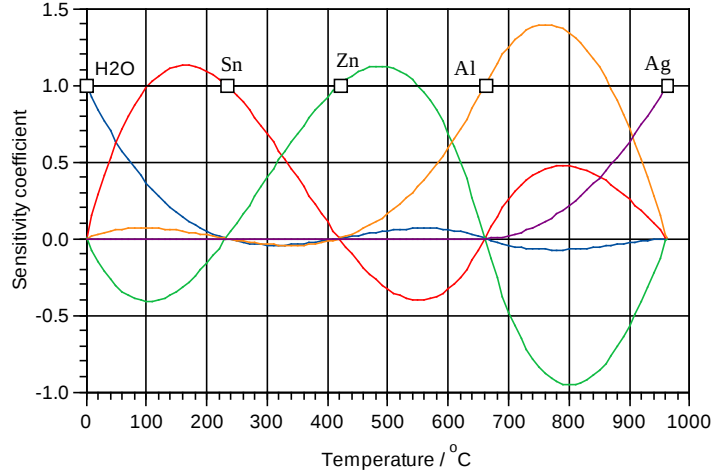


Figure A1.7: The sensitivity coefficients for the water-silver subrange. The curves plot the propagated error in millikelvin for a 1 mK error at each of the fixed points. Below 660.323 °C (the aluminium point) the curves are identical to those in Figure A1.6.

Argon-Water Subrange: 83.8058 K to 273.16 K

The ITS-90 interpolation equation is

$$W_r = W - a(W - 1) - b(W - 1)\ln(W), \quad (\text{A1.22})$$

where the a and b coefficients are determined by calibration of the SPRT at the triple points of argon and mercury. The interpolation may be written in the form

$$W_r(W) = f_{\text{H}_2\text{O}} + W_{\text{r,Hg}} f_{\text{Hg}} + W_{\text{r,Ar}} f_{\text{Ar}}, \quad (\text{A1.23})$$

where the sensitivity coefficients are

$$f_{\text{H}_2\text{O}} = \frac{(W - 1)(W_{\text{Hg}} - W_{\text{Ar}})\ln W + (W_{\text{Hg}} - 1)(W_{\text{Ar}} - W)\ln W_{\text{Hg}} - (W_{\text{Hg}} - W)(W_{\text{Ar}} - 1)\ln W_{\text{Ar}}}{(W_{\text{Hg}} - 1)(W_{\text{Ar}} - 1)(\ln W_{\text{Hg}} - \ln W_{\text{Ar}})}, \quad (\text{A1.24a})$$

$$f_{\text{Hg}} = \frac{(W - 1)(\ln W - \ln W_{\text{Ar}})}{(W_{\text{Hg}} - 1)(\ln W_{\text{Hg}} - \ln W_{\text{Ar}})}, \quad (\text{A1.24b})$$

$$f_{\text{Ar}} = \frac{(W - 1)(\ln W - \ln W_{\text{Hg}})}{(W_{\text{Ar}} - 1)(\ln W_{\text{Ar}} - \ln W_{\text{Hg}})}. \quad (\text{A1.24c})$$

The sensitivity coefficient for the triple point of water is most easily calculated as

$$f_{\text{H}_2\text{O}} = 1 - f_{\text{Hg}} - f_{\text{Ar}}.$$

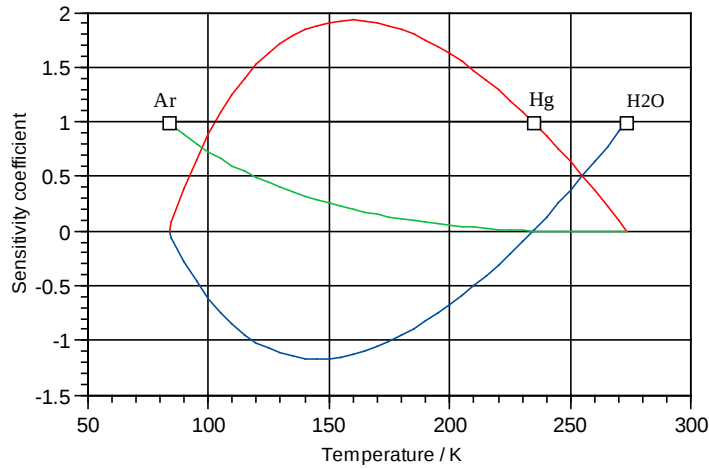


Figure A1.8: The sensitivity coefficients for the argon-water subrange. The curves show the propagated error in millikelvin for a 1 mK error at each of the fixed points.

Oxygen-Water Subrange: 54.3584 K to 273.16 K

The ITS-90 interpolation equation takes the form

$$W_r(W) = W - a(W - 1) - b(W - 1)^2 - c(\ln W)^2, \quad (\text{A1.25})$$

where the a , b and c coefficients are determined by calibration of the SPRT at the triple points of oxygen, argon and mercury. The interpolation may be written in the form

$$W_r = f_{\text{H2O}} + W_{r,\text{Hg}} f_{\text{Hg}} + W_{r,\text{Ar}} f_{\text{Ar}} + W_{r,\text{O2}} f_{\text{O2}}, \quad (\text{A1.26})$$

where the sensitivity coefficient for the oxygen point is

$$f_{\text{O2}}(W) = \frac{(\ln W)^2 (W_{\text{Hg}} - 1)(W_{\text{Ar}} - 1)(W_{\text{Ar}} - W_{\text{Hg}}) - (\ln W_{\text{Ar}})^2 (W_{\text{Hg}} - 1)(W - 1)(W - W_{\text{Hg}}) + (\ln W_{\text{Hg}})^2 (W - 1)(W_{\text{Ar}} - 1)(W - W_{\text{Ar}})}{(\ln W_{\text{O2}})^2 (W_{\text{Hg}} - 1)(W_{\text{Ar}} - 1)(W_{\text{Ar}} - W_{\text{Hg}}) - (\ln W_{\text{Ar}})^2 (W_{\text{Hg}} - 1)(W_{\text{O2}} - 1)(W_{\text{O2}} - W_{\text{Hg}}) + (\ln W_{\text{Hg}})^2 (W_{\text{O2}} - 1)(W_{\text{Ar}} - 1)(W_{\text{O2}} - W_{\text{Ar}})} \quad (\text{A1.27})$$

The equations for the argon and mercury functions can be found by permuting the indices (e.g., swap the Ar and O2 subscripts to obtain the $f_{\text{Ar}}(W)$ function), and the water function found from $f_{\text{H2O}} = 1 - f_{\text{O2}} - f_{\text{Ar}} - f_{\text{Hg}}$. Alternatively, the following numerical approximations may be used:

$$f_{\text{O}_2} = -1.598425679 + 3.403991489W - 1.805565809W^2 + 0.4032254894(\ln W)^2, \quad (\text{A1.28a})$$

$$f_{\text{Ar}} = +3.930746333 - 8.465193581W + 4.534447249W^2 - 0.5593488293(\ln W)^2, \quad \text{to}$$

$$f_{\text{Hg}} = -4.044936086 + 16.33613127W - 12.29119519W^2 + 0.4643043265(\ln W)^2, \quad (\text{A1.28d})$$

$$f_{\text{H}_2\text{O}} = +2.712615432 - 11.27492918W + 9.562313751W^2 - 0.3081809866(\ln W)^2.$$

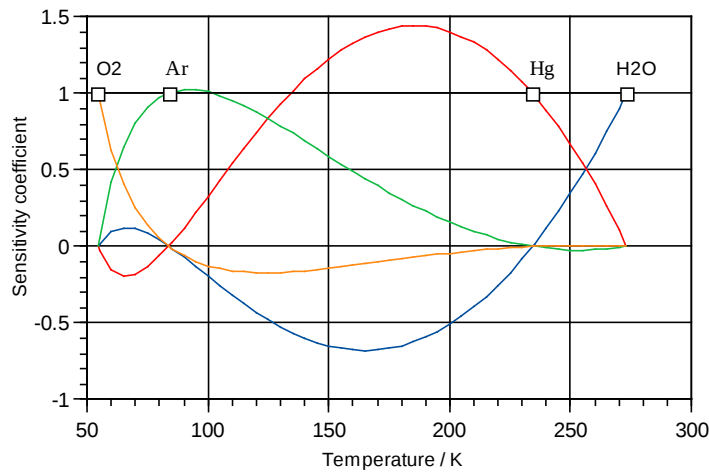


Figure A1.9: The sensitivity coefficients for the oxygen-water subrange. The curves show the propagated error in millikelvin for a 1 mK error at each of the fixed points.

Neon-Water Subrange: 24.5561 K to 273.16 K

The ITS-90 interpolation equation is

$$W_r = W - a(W - 1) - b(W - 1)^2 - c \ln W - d(\ln W)^2 - e(\ln W)^3, \quad (\text{A1.29})$$

where the a , b , c , d , and e coefficients are determined by calibration of the SPRT at the triple points of equilibrium hydrogen, neon, oxygen, argon and mercury. The interpolation may be written in the form

$$W_r = f_{\text{H}_2\text{O}} + W_{r,\text{Hg}} f_{\text{Hg}} + W_{r,\text{Ar}} f_{\text{Ar}} + W_{r,\text{O}_2} f_{\text{O}_2} + W_{r,\text{Ne}} f_{\text{Ne}} + W_{r,\text{e-H}_2} f_{\text{e-H}_2}. \quad (\text{A1.30})$$

The sensitivity coefficients in this case are too complex to write down in algebraic form. The following numerical approximations may be used:

$$f_{\text{Hg}} = -24.30265740 + 42.99793720W - 18.69528095W^2 - 14.20610018\ln(W) - 2.778961721\ln^2(W) - 0.1788181624\ln^3(W), \quad (\text{A1.31a})$$

$$f_{\text{Ar}} = +33.02979668 - 46.64306946W + 13.61327293W^2 + 20.51522598\ln(W) + 4.195086350\ln^2(W) + 0.2785479955\ln^3(W),$$

$$f_{\text{O}_2} = -29.45087407 + 39.79347654W - 10.34260259W^2 - 19.77485453\ln(W) - 4.269113231\ln^2(W) - 0.2941747479\ln^3(W),$$

$$f_{\text{Ne}} = +7.413452875 - 9.508661062W + 2.095208241W^2 + 5.423905534\ln(W) + 1.384488984\ln^2(W) + 0.1102290468\ln^3(W),$$

$$f_{\text{H}} = -1.715380724 + 2.160321157W - 0.4449404100W^2 - 1.291145588\ln(W) - 0.3520630280\ln^2(W) - 0.03269335415\ln^3(W),$$

$$f_{\text{H}_2\text{O}} = +16.02566264 - 28.80000438W + 13.77434278W^2 + 9.33296879\ln(W) + 1.820562646\ln^2(W) + 0.1169092222\ln^3(W). \quad (\text{A1.31f})$$

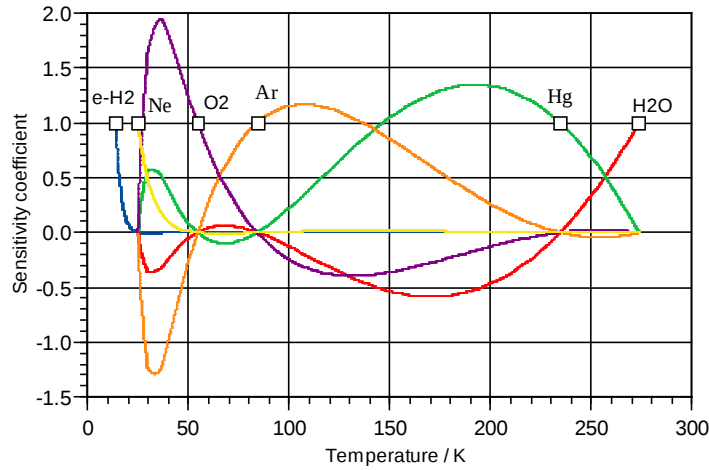


Figure A1.10: The sensitivity coefficients for the neon-water subrange. The curves plot the propagated error in millikelvin for a 1 mK error at each of the fixed points. Note that the subrange extends only down to the Neon point (24.5561 K), although the hydrogen point is below this temperature.

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