

# SI-traceable carbon isotope ratios

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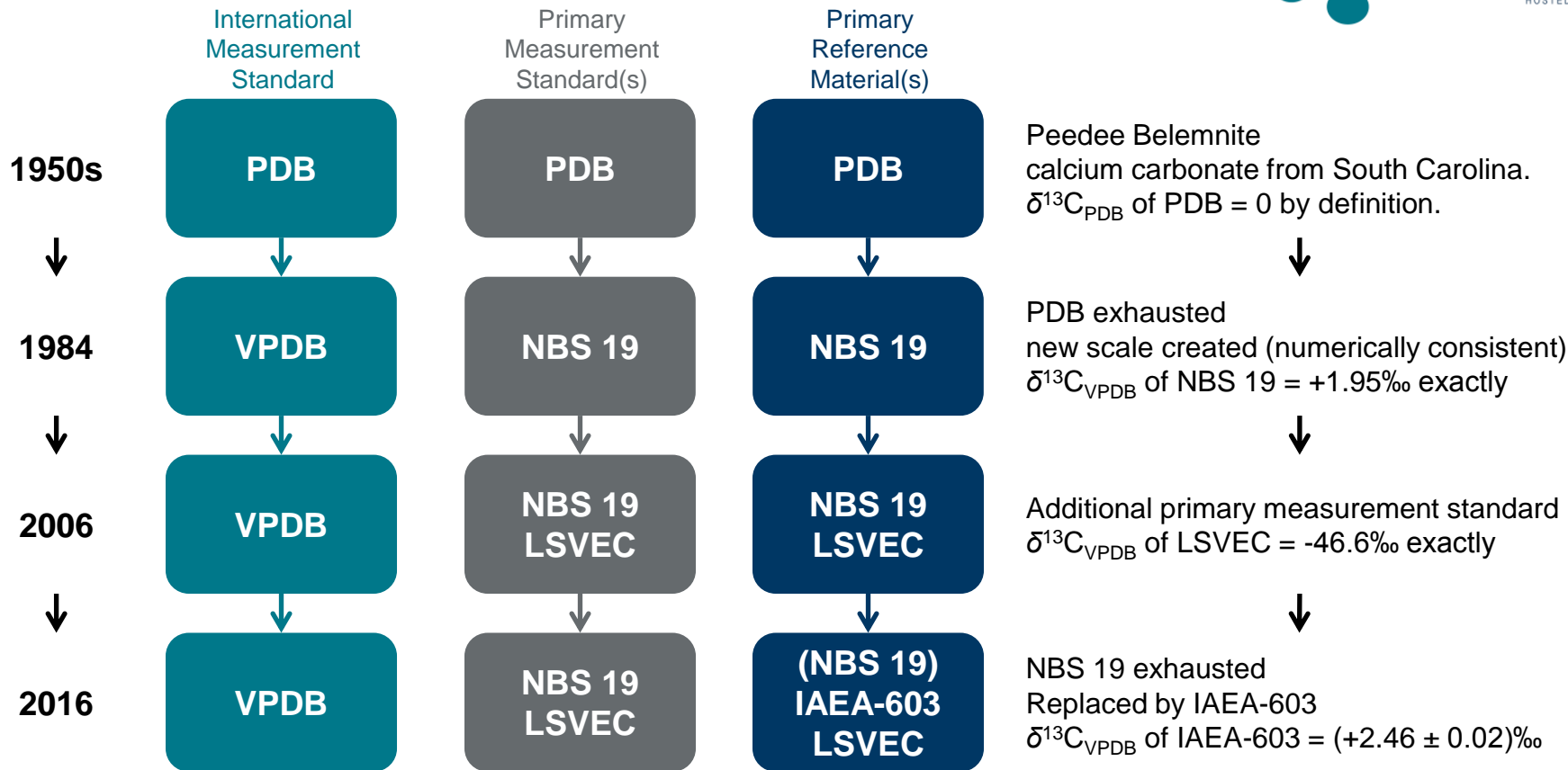
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# Measurement standards for isotope delta



- International Measurement Standards
  - Adopted by convention
  - Zero-point of scale
  - Exact isotope delta
  - Can be a reference material, homogenous natural reservoir or virtual material
  - Can be more than one for each element (e.g. oxygen) but only one per scale
  - E.g. VSMOW, NIST SRM 987, NBS 28, atmospheric nitrogen, SMOB, etc.
- Primary measurement standards
  - Adopted by convention
  - Exact isotope delta (can be non-zero)
  - Can be more than one per scale
  - E.g. SLAP, IAEA-S-1, USGS32, etc.
- Primary reference materials
  - Adopted by convention
  - Exact zero or non-zero isotope delta (“replacement” materials bear uncertainty)
  - Realisations of primary and international measurement standards
  - E.g. VSMOW2, SLAP2, IRMM-014, etc.

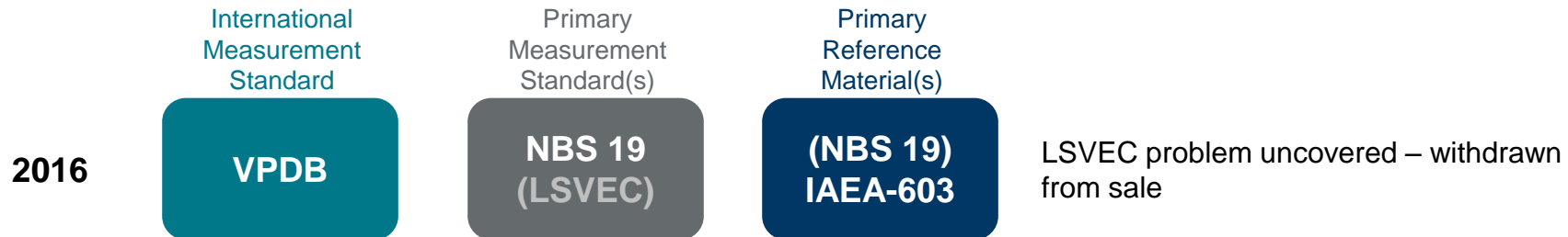
# Measurement standards for carbon isotope delta



## Problem with LSVEC uncovered in 2015-16



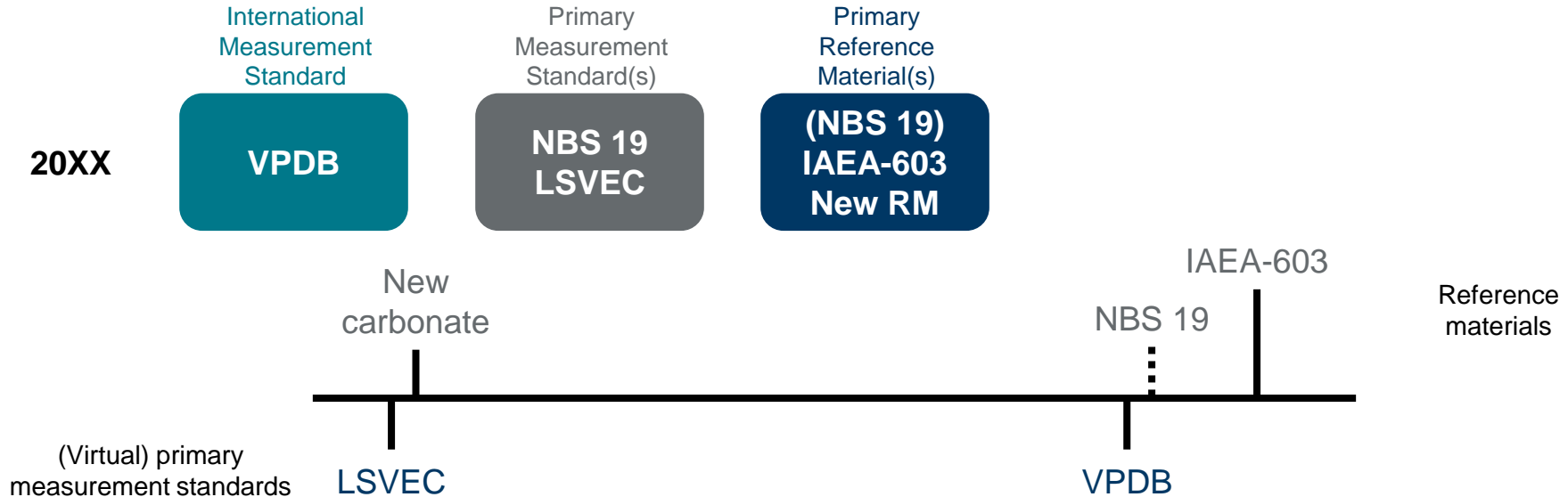
- IAEA noticed that carbon isotope ratio of LSVEC varied between different storage containers.
- Result of water present in original material (as well as that adsorbed from the atmosphere during use).
  - Atmospheric CO<sub>2</sub> ( $\delta^{13}\text{C}_{\text{VPDB}} \sim -8 \text{ ‰}$ ) absorbed by water and then exchanges with carbonate.
- Withdrawn from sale (for C) – but remains within traceability chain



# Possible options for addressing the LSVEC problem

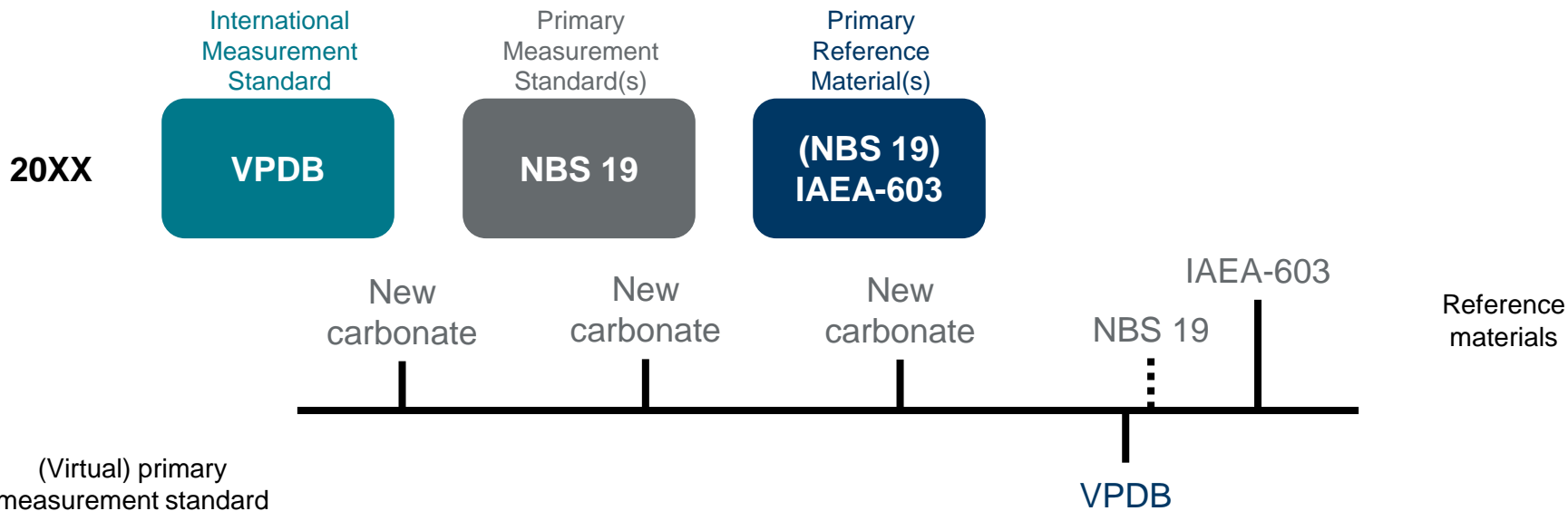


- Replace it with another carbonate and maintain the current (2006) scale
  - Effectively defines a “virtual LSVEC” at  $\delta^{13}\text{C}_{\text{VPDB}} = -46.6 \text{ ‰}$
  - All other existing RMs retain their currently-assigned values.
  - LSVEC physical material still not for C isotope analysis – potentially very confusing.



## Possible options for addressing the LSVEC problem

- Replace it with other carbonate(s) and change the scale
  - New range of carbonate RMs calibrated to IAEA-603 (secondary RMs)
  - Need to re-assess all other RMs to provide assigned values on new scale.
  - Resultant need to change terminology for scales – e.g. include year.



## Artefact-based scales vs the SI



- Scales traceable to an artefact
  - Can develop independently (e.g. oxygen isotope-delta scales).
  - Need to maintain the scale (drift in reference artefacts, difficult/impossible to regenerate).
  - Provide very consistent and precise results.

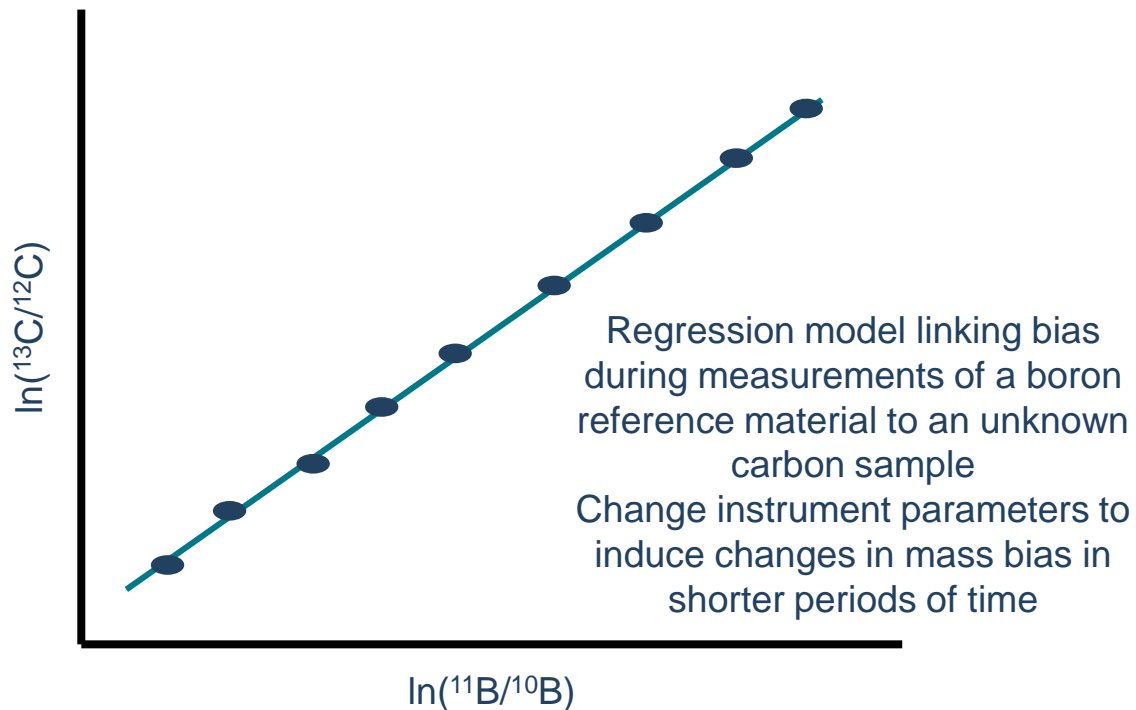
- Scales traceable the SI (i.e. a measurement process).
  - Provide accurate results.
  - Can have more than one source.
  - Enhanced metrological traceability and/or comparability

$^{13}R_{sample}$

$$\delta^{13}C = \frac{^{13}R_{sample}}{^{13}R_{standard}} - 1$$

## Alternative to relative (i.e. delta) measurements

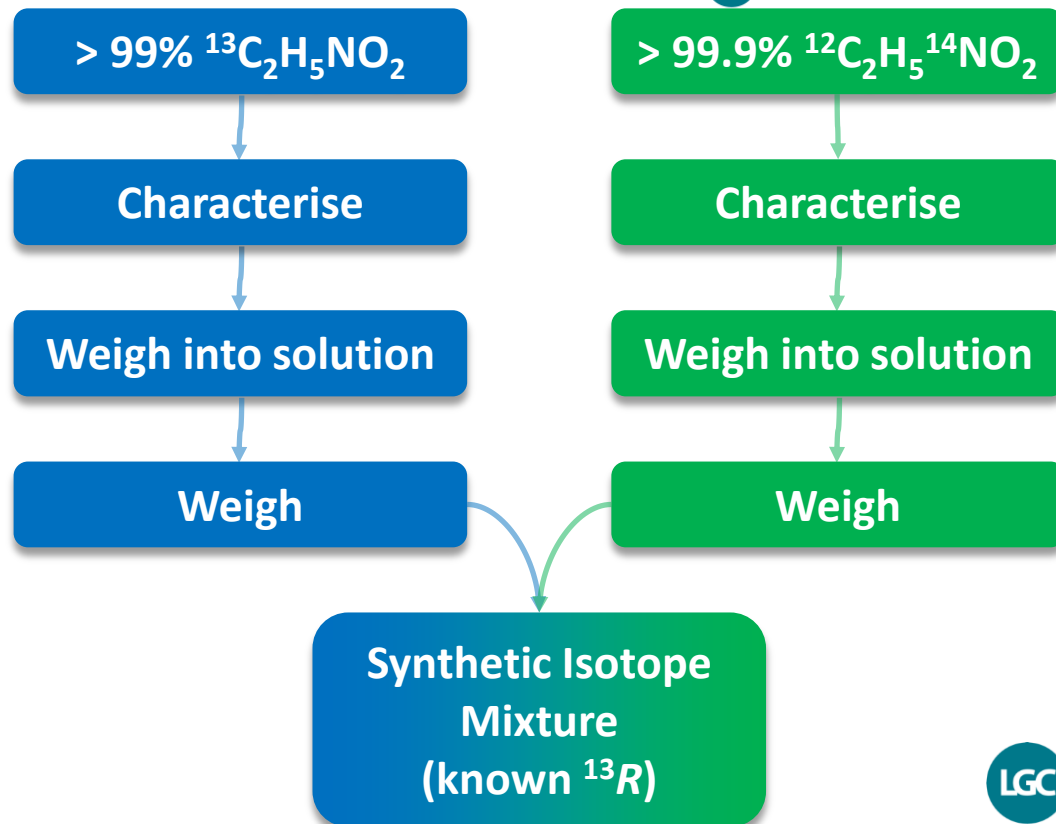
- Mass spectrometers are precise but not accurate
- Need materials with known isotope ratio for calibration
  - Reference material(s) with known absolute isotope ratio
  - Reference material of a different element with known isotope ratio





## Alternative to relative (i.e. delta) measurements

- Mass spectrometers are precise but not accurate
- Need materials with known isotope ratio for calibration
  - Reference material(s) with known absolute isotope ratio
  - Reference material of a different element with known isotope ratio
  - Gravimetric preparation of mixtures of isotopes (or isotopologues) with known purity



# Absolute carbon isotope ratio measurements at LGC



2008

*Anal. Chem.* 2008, 80, 5963-5969

## Precise and Traceable $^{13}\text{C}/^{12}\text{C}$ Isotope Amount Ratios by Multicollector ICPMS

Rebeca Santamaria-Fernandez,\* David Carter, and Ruth Hearn

LGC, Queens Road, Teddington, Middlesex, TW11 0LY, U.K.

MC-ICP-MS used to determine absolute carbon isotope ratios for the first time. Mass bias corrected using NIST Boron CRM as internal standard

2013

JAAS

RSCPublishing

PAPER

[View Article Online](#)  
[View Journal](#) | [View Issue](#)

Cite this: *J. Anal. At. Spectrom.*, 2013, 28, 1760

## Determination of absolute $^{13}\text{C}/^{12}\text{C}$ isotope amount ratios by MC-ICPMS using calibration with synthetic isotope mixtures

Dmitry Malinovsky,\* Philip J. H. Dunn and Heidi Goenaga-Infante

Use of synthetic isotopologue mixtures of glycine for calibration. RM produced under accreditation to ISO 17025 and 17034

2015

*Anal. Bioanal. Chem.* (2015) 407:3169-3180  
DOI 10.1007/s00216-014-7926-1

RESEARCH PAPER

## Calibration strategies for the determination of stable carbon absolute isotope ratios in a glycine candidate reference material by elemental analyser-isotope ratio mass spectrometry

Philip J. H. Dunn · Dmitry Malinovsky · Heidi Goenaga-Infante

Confirmation using EA-IRMS and redetermination of absolute isotope ratio of VPDB international measurement standard.

## Limitations of the work



- Parent glycines enriched in  $^{13}\text{C}$  and  $^{12}\text{C}$  were not investigated for possible co-enrichment of H, N and O isotopes
  - Potential for molecular weights of the parent glycines to be outside of uncertainty bounds used in calculations (so uncertainty in final ratios may be underestimated)
  - Parent glycines sent to the USGS Stable Isotope Laboratory in Reston, Virginia for analysis
  - No significant enrichment found which would cause the molecular weights we used for the parent glycines to be invalid.
- With one reference material, calibration is not possible as it is preferable to use two or more RMs to allow scale span to be determined.
  - Suite of three glycines with isotopic compositions spanning the natural abundance range prepared as reference materials.

### Acknowledgement:

Tyler Coplen and Haiping Qi at the USGS Reston SIL



Fully calibrated?	✓
Detailed uncertainty budget?	✓
RM available?	✓
CIAAW concerns addressed	✓



JAAS

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[View Journal](#) | [View Issue](#)



Cite this: *J. Anal. At. Spectrom.*, 2019, 34, 147

## Development and characterisation of new glycine certified reference materials for SI-traceable $^{13}\text{C}/^{12}\text{C}$ isotope amount ratio measurements

D. Malinovsky, \* P. J. H. Dunn, \* G. Holcombe, S. Cowen and H. Goenaga-Infante

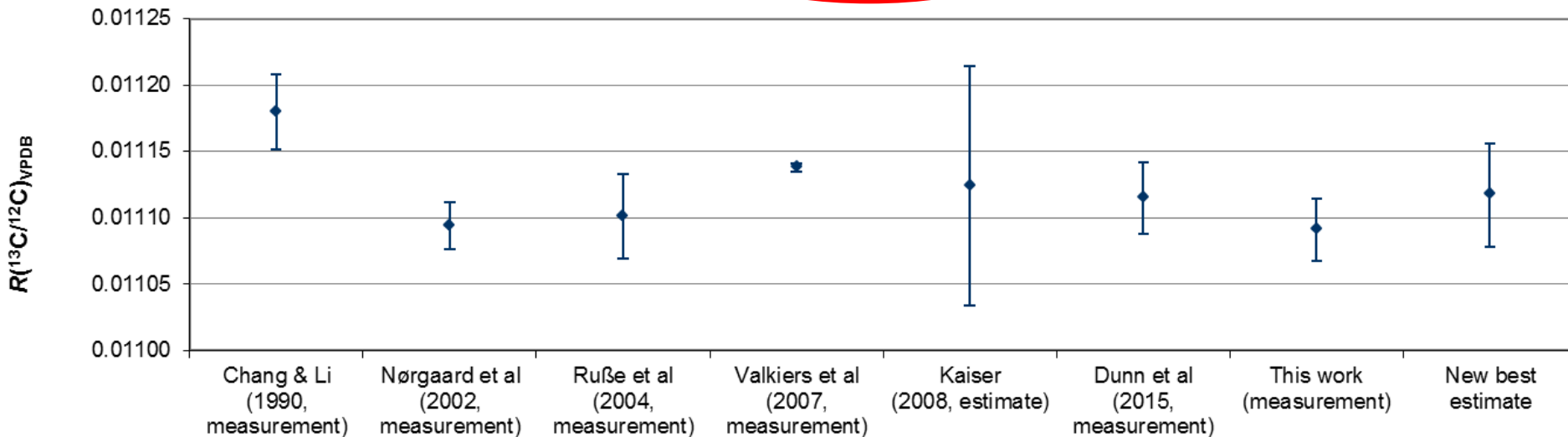


- CIAAW has recognised this work as a new best measurement of the isotopic abundance of carbon



## What about VPDB?

$$\delta^{13}\text{C} = \frac{^{13}\text{R}_{\text{sample}}}{^{13}\text{R}_{\text{standard}}} - 1$$



- Similar approach now adopted by CIAAW but using a slightly different calculation approach to come up with their own best estimate (NIST Consensus Builder – available online)

## Impacts of change to recommended absolute isotope ratio of VPDB



- Change to recommended  $^{17}\text{O}$  correction algorithm
  - Impact on measurement of clumped isotopes
- Optical instrumentation
- Conversion of isotope delta to other quantities (absolute ratios, atom fractions, atomic weight)
  
- Submitted STELLAR project (EURAMET) is hoping to supply more independent measurements of this ratio to reduce the associated uncertainty significantly.

## Summary



- LGC has worked towards measuring absolute isotope ratios traceable to the SI for a number of years
  - Achieved ISO/IEC 17025 accreditation for measurement by MC-ICP-MS
  - Produced four RMs under ISO/IEC 17034 accreditation
  - Re-determined the absolute isotope ratio of VPDB (the virtual international standard)
- This body of work has recently been recognised as the best measurement of the isotopic abundance of carbon by IUPAC CIAAW

## Acknowledgements



- Gill Holcombe and the RM production team
- Simon Cowen, Steve Ellison and the stats team
- Cailean Clarkson and the purity team
- Ty Coplen and Haiping Qi at the USGS Reston Stable Isotope Laboratory