

# Candidates for cooled and trapped optical standards

– key features for the perfect atom/ion

Stephen Lea

*National Physical Laboratory, Teddington, Middlesex TW11 0LW U.K.*

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# Outline of talk

## Desirable features of the ideal atom/ion

- why has caesium been so successful (~ 50 years)?

## Allowed and forbidden transitions

## Hyperfine structure and choice of isotope

## Trapped ions

- stabilisation: quantum jumps
- alkali-like ions:  $\text{Ca}^+$ ,  $\text{Sr}^+$ ,  $\text{Hg}^+$ ,  $\text{Yb}^+$ , ...

## Cold atoms

- stabilisation: atom interferometer
- alkaline-earth atoms: Mg, Ca, Sr, Yb, ...

## Other candidates

## Stability

## Some conclusions

**N.B. (mainly) not a review of current status – see: Proc. 6<sup>th</sup> Symp. Freq. Stds. Metrol., ed P. Gill (World Scientific 2002)**

# Desirable features of a primary frequency standard

It must ...

- show significant gain in accuracy and stability ( $> \times 10$ ) over previous standard
- provide useful output signal – easily compared with existing standards
- be reproducible by (at least) several standards laboratories

- femtosecond comb

- wavelengths accessible with existing laser technology

Atoms/ions

$$Q > 10^{14}$$

laser linewidth

$$\Delta\nu < 1 \text{ Hz}$$

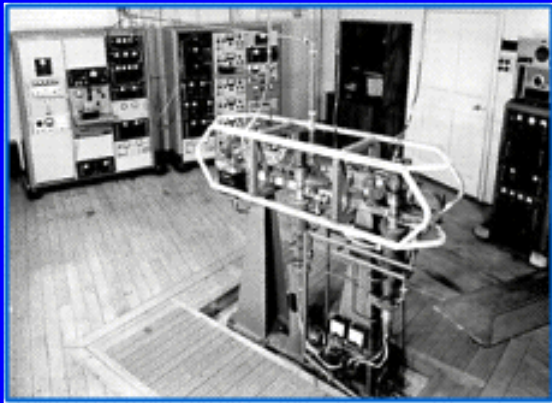
Would like ...

- to make commercial version
- to relate directly to fundamental constants (quantum standard)

- probably challenging...

- theoretical understanding of atomic structure

# Caesium – an ideal atom?



1950s: thermal beam standard

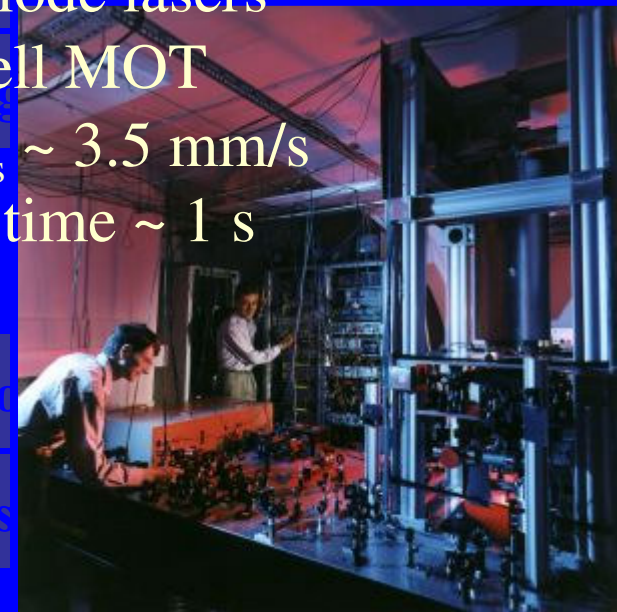
- Easy to make caesium thermal beam
- HFS transition at 9.2 GHz – X-band  $\mu\text{W}$
- Simple enough to make commercial product

1990s: laser cooling – fountain standard

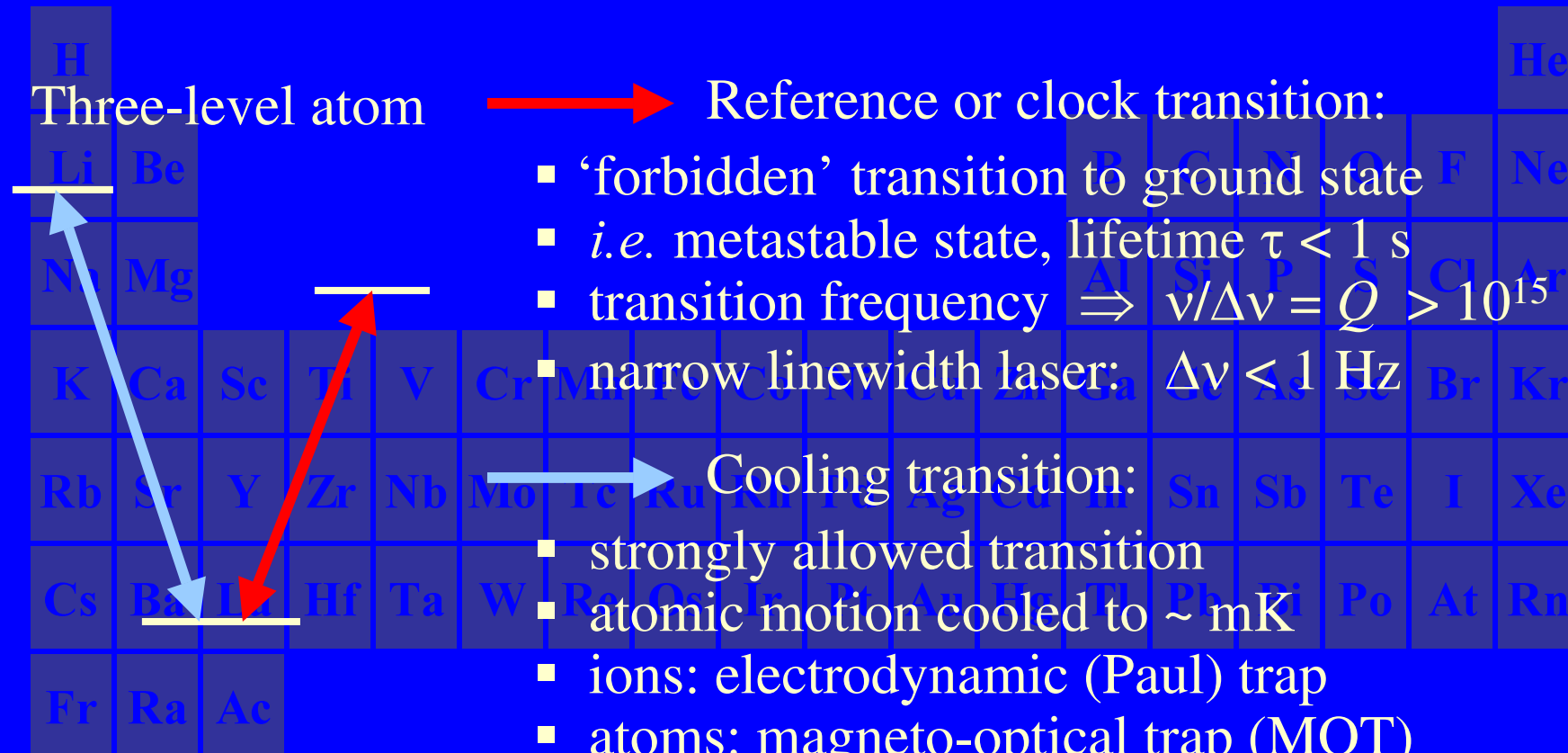
- cooling transition at 852 nm – diode lasers
- low vapour pressure – vapour cell MOT
- HFS  $\rightarrow$  Sisyphus cooling to  $v_{\text{rms}} \sim 3.5 \text{ mm/s}$
- fountain geometry – interaction time  $\sim 1 \text{ s}$

2000s: present status

- accuracy  $\Delta\nu/\nu = 1.0 \times 10^{-15}$
- stability  $\sigma_y = 5 \times 10^{-14} \tau^{-1/2}$
- limits: collisions; LO stability; QPN



# Laser-cooled optical frequency standard



Suitable atoms/ions:

- S ground state – zero orbital angular momentum
- one valence electron outside closed shells – spin  $S = 1/2 \rightarrow {}^2S_{1/2}$
- two valence electrons outside closed shells – spin  $S = 0 \rightarrow {}^1S_0$

# Cooling transition: allowed electric dipole ( $E1$ )

H	P				He											
Li	Be				F	Ne										
Na	Mg				Cl	Ar										
K	Ca	Sc	Ti	V	Br	Kr										
Rb	Sr	S	Zr	Nb	I	Xe										
Cs	Ba	La	Hf	Ta	W	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn

Rate:  $A = \frac{4}{3} \alpha \omega^3 \langle S | \mathbf{r} | P \rangle^2 / c^2$

Angular frequency – Rydberg constant:

$$\omega \sim 2\pi c R_\infty \rightarrow \omega \approx c R_\infty \Leftrightarrow 572 \text{ nm}$$

Matrix element:  $\langle S | \mathbf{r} | P \rangle \sim a_0 \rightarrow 5a_0$

$$\Rightarrow A \approx 25 \alpha c R_\infty^3 a_0^2 = 2 \times 10^8 \text{ s}^{-1}$$

$$\Rightarrow \tau \sim 5 \text{ ns}, \quad \Gamma \approx 30 \text{ MHz}$$

Selection rules:

$$\Delta L = \pm 1 \text{ (states of opposite parity)}$$

$$\Delta S = 0 \text{ (no change of spin)}$$

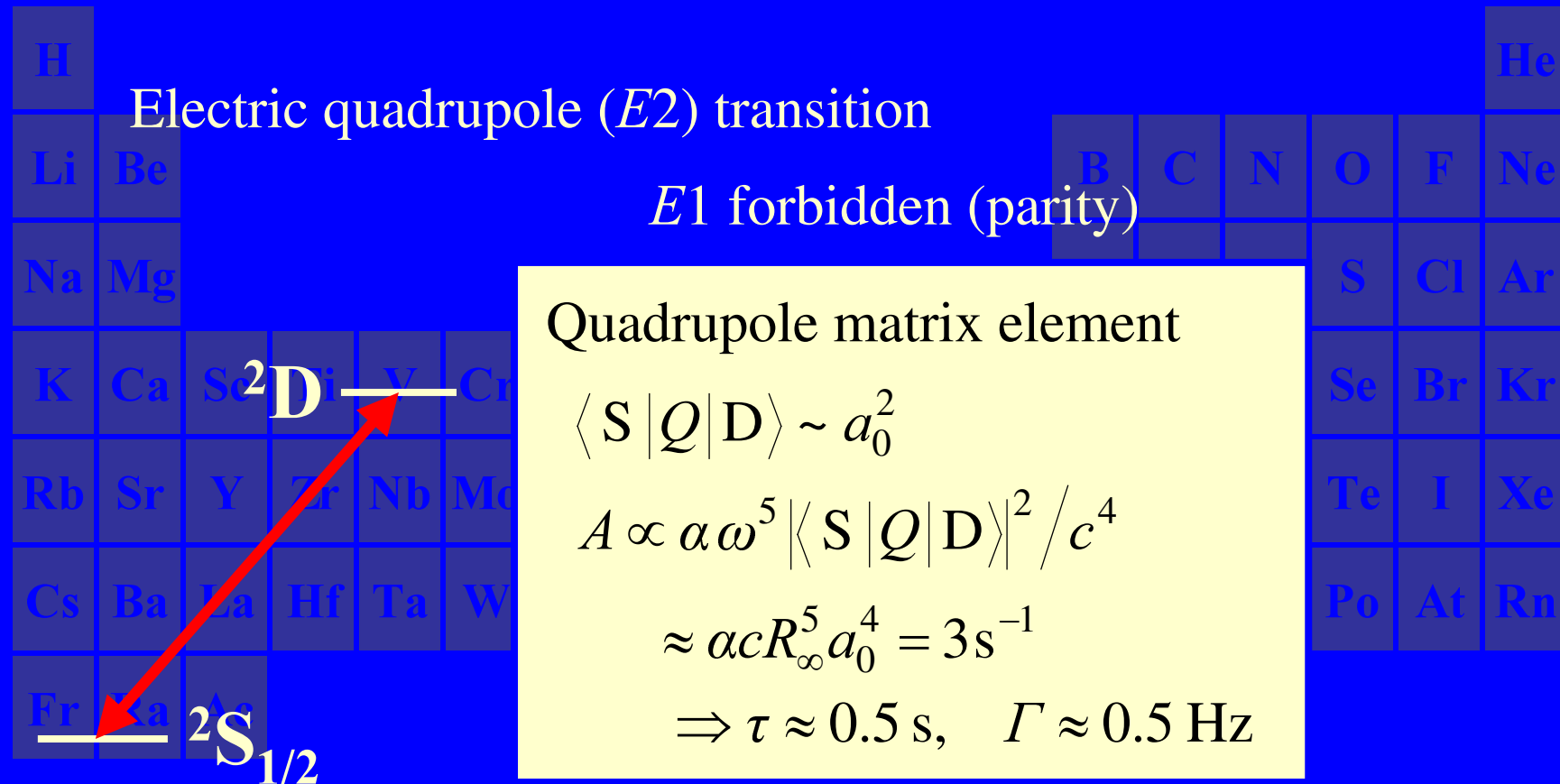
Doppler cooling:

$$\Gamma \sim 30 \text{ MHz} \Rightarrow T_D = \hbar \Gamma / 2k_B \sim 1 \text{ mK}, \quad v_{\text{rms}} \sim 0.3 \text{ ms}^{-1}$$

Photon recoil limit – momentum exchange with cooling light:

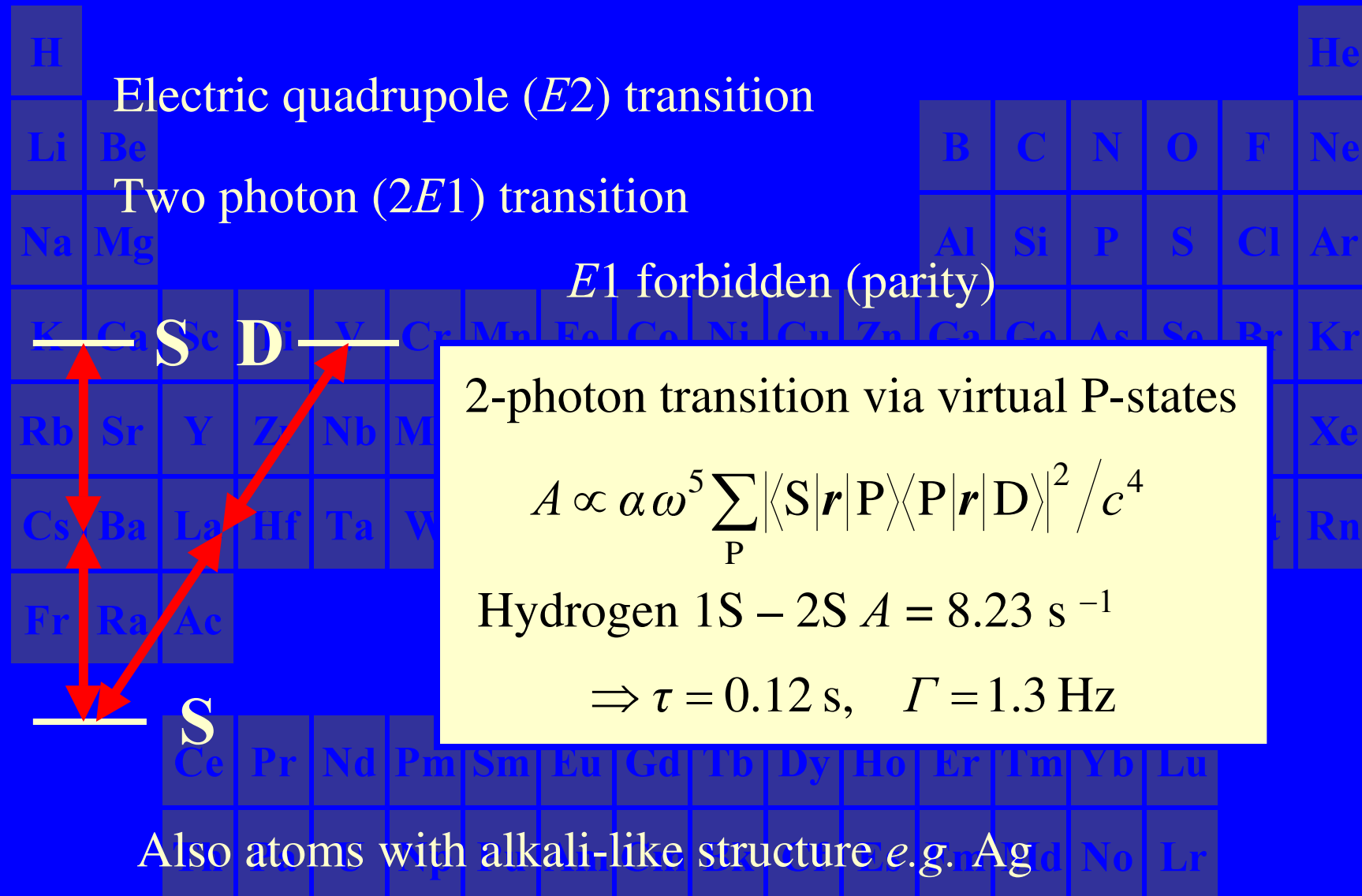
$$T_{\text{recoil}} = (h/\lambda)^2 / k_B M_A \sim 1 \mu\text{K}, \quad v_{\text{rms}} \sim 0.01 \text{ ms}^{-1}$$

# Clock transition: $E1$ forbidden



e.g. alkali-like ions:  $\text{Sr}^+$ ,  $\text{Yb}^+$ ,  $\text{Hg}^+$ ...

# Clock transition: $E1$ forbidden



Electric quadrupole ( $E2$ ) transition

Two photon ( $2E1$ ) transition

$E1$  forbidden (parity)

2-photon transition via virtual P-states

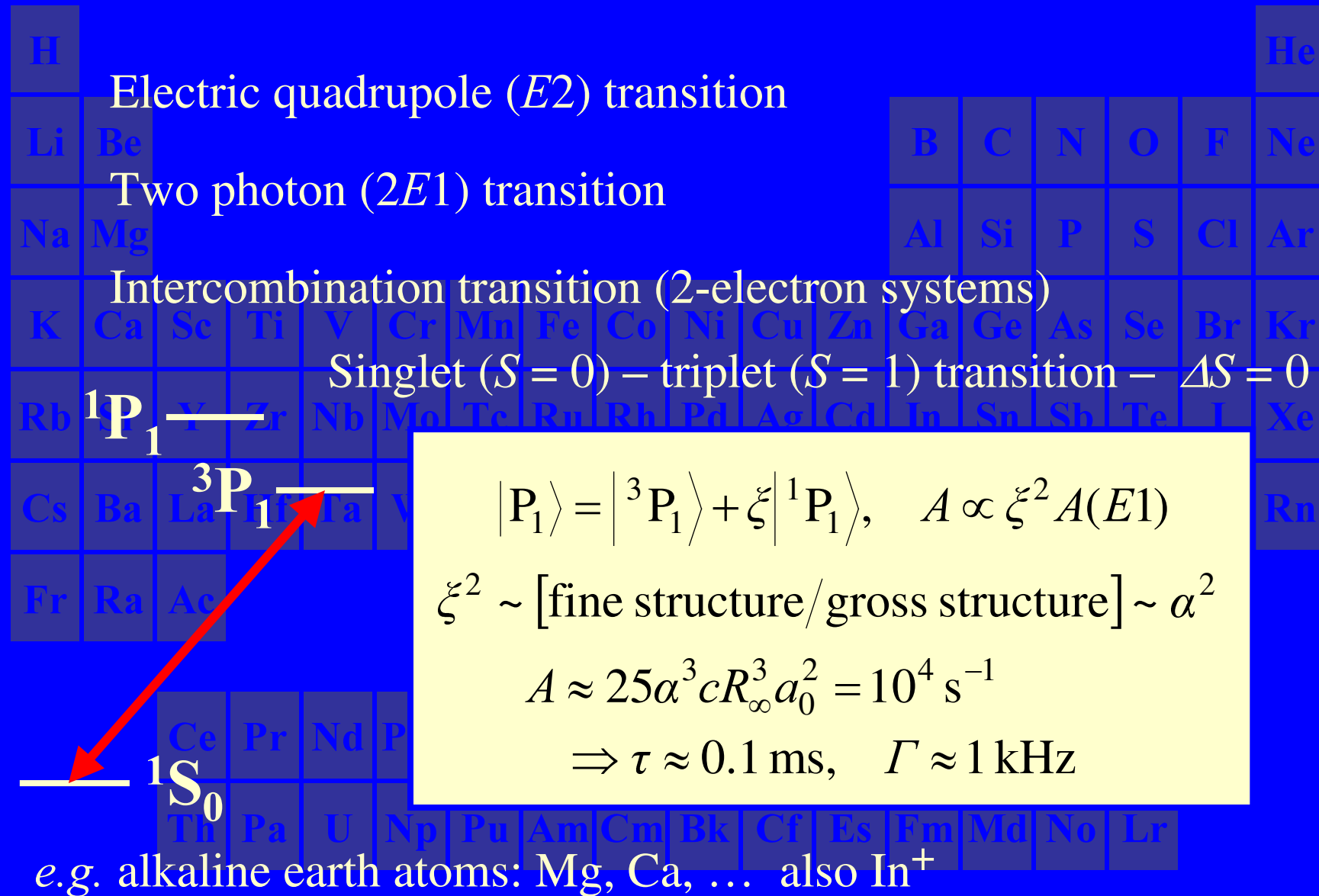
$$A \propto \alpha \omega^5 \sum_{\text{P}} |\langle \text{S} | \mathbf{r} | \text{P} \rangle \langle \text{P} | \mathbf{r} | \text{D} \rangle|^2 / c^4$$

Hydrogen  $1\text{S} - 2\text{S}$   $A = 8.23 \text{ s}^{-1}$

$$\Rightarrow \tau = 0.12 \text{ s}, \quad \Gamma = 1.3 \text{ Hz}$$

Also atoms with alkali-like structure *e.g.* Ag, Nd, No, Lr

# Clock transition: $E1$ forbidden



# The Zeeman effect and hyperfine structure

## Linear (first order) Zeeman shift

$\Delta E = m_J g_J \mu_B B \Rightarrow$  No shift for  $m_J = 0 \rightarrow m_J' = 0$

1 electron *e.g.*  $^2D_{5/2}$  level  $\Rightarrow m_J = \pm 1/2, \pm 3/2, \pm 5/2$

2 electrons *e.g.*  $^3P_1$  level  $\Rightarrow m_J = 0, \pm 1$

## Hyperfine splitting:

interaction of total electron angular momentum  $J$  with nuclear spin  $I$

Mass number  $A$  even  $\Rightarrow I = 0$ ,  $A$  odd  $\Rightarrow I = 1/2 \dots$

$\Delta E \propto AIJ \Rightarrow$  Splitting into levels described by  $F$

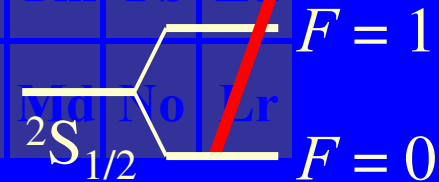
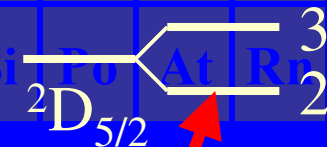
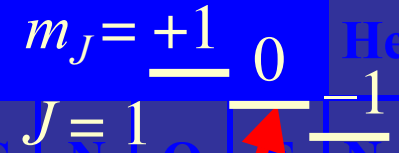
*e.g.* for  $I = 1/2$ ,  $F = J \pm 1/2$

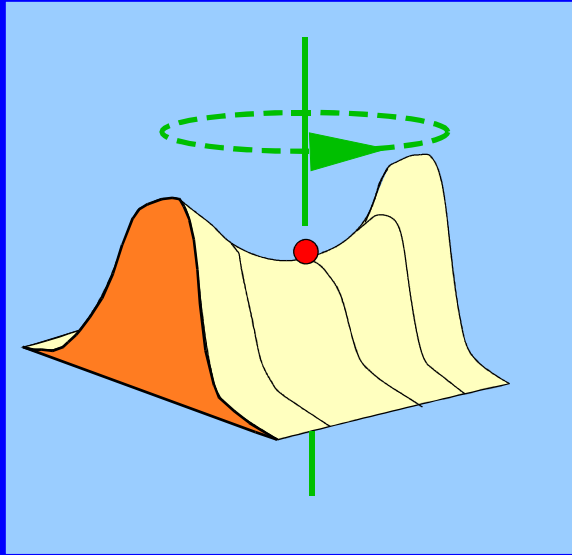
1 electron *e.g.*  $^2D_{5/2}$   $F = 2$  level  $\Rightarrow m_F = 0, \pm 1, \pm 2$

2 electrons *e.g.*  $^3P_1$   $F = 1/2$  level  $\Rightarrow m_F = \pm 1/2$

$\Rightarrow$  1-e systems  $\Rightarrow$  odd isotope

2-e systems  $\Rightarrow$  even isotope





# Trapped single ion

Paul (electrodynamic) trap, drive frequency  $\Omega_{\text{RF}}$ :

Ion's motion has two components:

*Secular* motion,  $\omega_{r,z} = \frac{1}{2} \beta_{r,z} \Omega_{\text{RF}}$

where  $\beta$  is a function of the trap geometry,

*Micromotion* at the drive frequency  $\Omega_{\text{RF}}$ .

Doppler cooling:

$$\Gamma \sim 30 \text{ MHz} \Rightarrow T_D = \hbar\Gamma/2k_B \sim 1 \text{ mK}$$

Harmonic oscillator potential:

$$E_{\text{total}} = \frac{1}{2}T + \frac{1}{2}V = \frac{1}{2}M_A\omega^2 x_0^2$$

$$x_0 = \frac{1}{\omega} \sqrt{\frac{\hbar\Gamma}{M_A}} \sim 50 \text{ nm} \quad \langle n \rangle = \frac{1}{2} \left( \frac{\Gamma}{\omega} - 1 \right) \sim 10$$

Typical values:

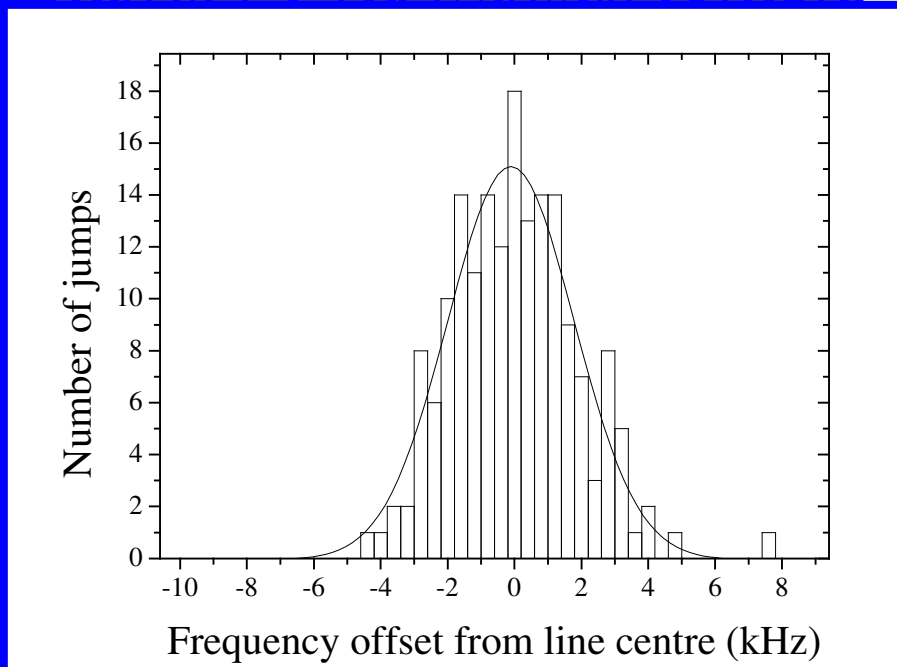
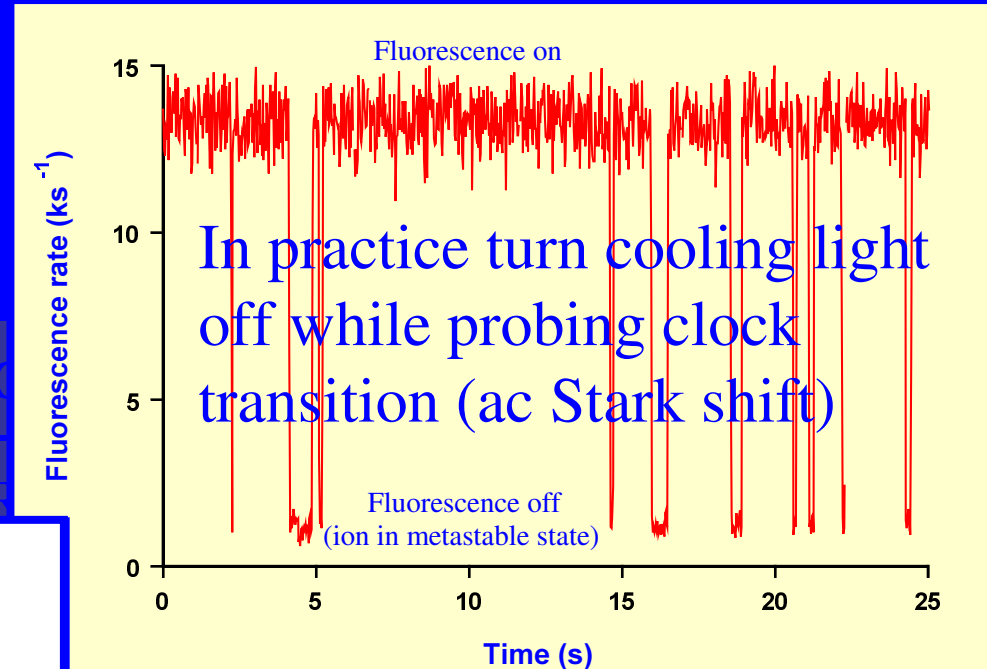
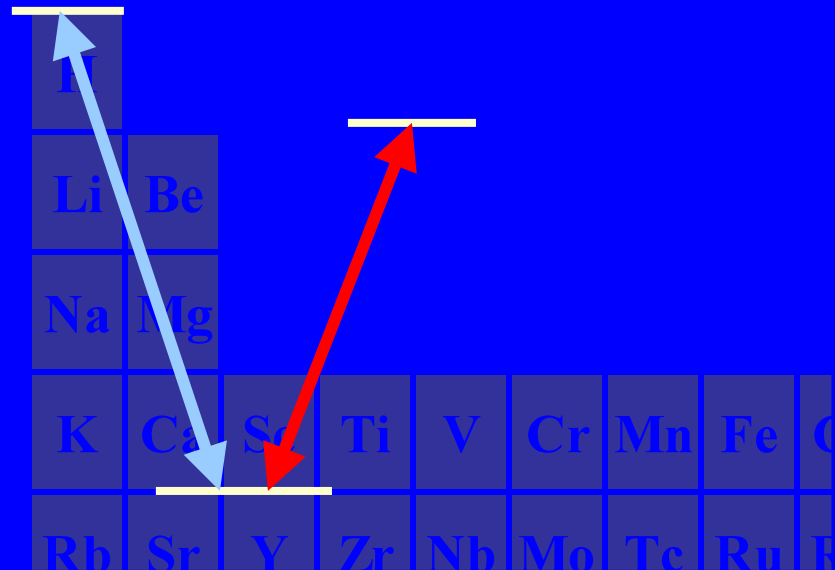
$$\Omega_{\text{RF}} \sim 10 \text{ MHz}$$

$$\beta_{r,z} \sim 0.2$$

$$\omega_{r,z} \sim 1 \text{ MHz}$$

$x_0 < \lambda \Rightarrow$  Lamb-Dicke regime, Doppler shift eliminated

# Stabilisation: quantum jumps



## Stabilisation:

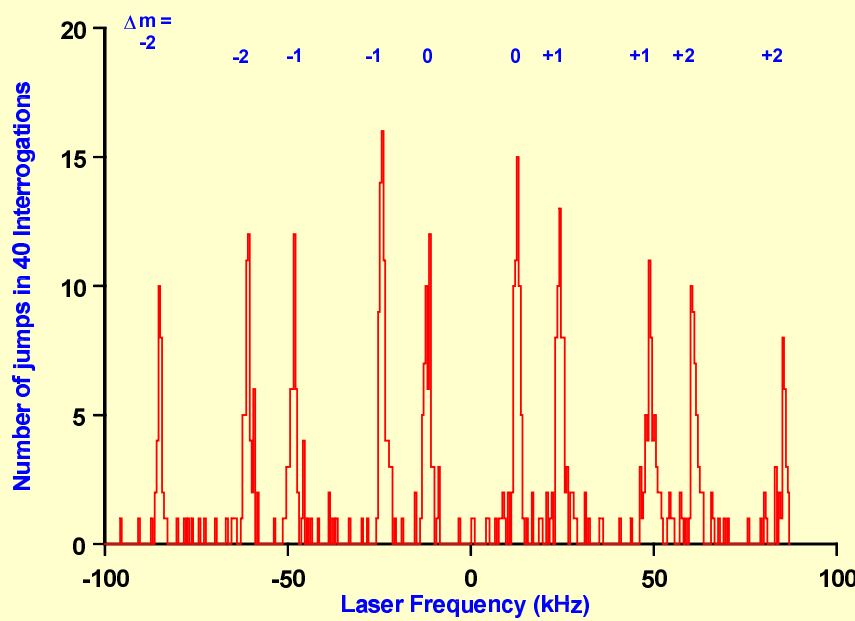
Count number of quantum jumps in time  $\tau > \tau_D$  at half-height on each side of line centre

# Alkali-like ions – quadrupole transitions

H	$^2P_{3/2}$ —	
Li	Be	$^2P_{1/2}$ —
Na	Mg	
K	<b>Ca</b>	Sc
Rb	<b>Sr</b>	Y
Cs	<b>Ba</b>	La
Fr	Ra	Ac

$^{40}\text{Ca}^+$  —  $\lambda$ 's ~

Zeeman components of the  $^2S_{1/2} - ^2D_{5/2}$  transition at 674 nm in  $^{88}\text{Sr}^+$



nm	clock Hz
729	0.15
674	0.46
1761	0.003
282	1.6
411	20
435	3.1

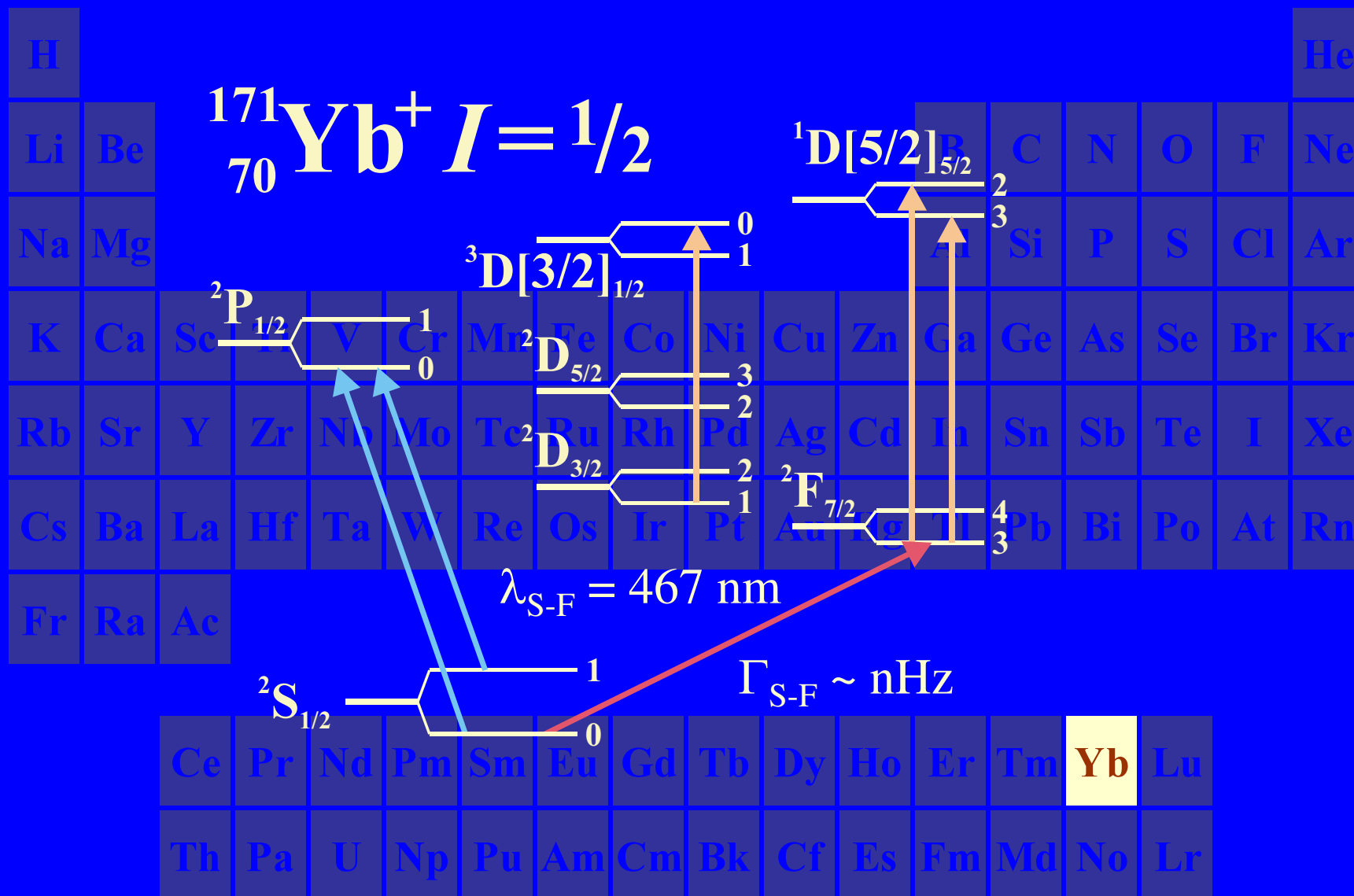
$^{88}\text{Sr}^+$  —  $\lambda$ 's ~ OK; no HFS  $\Rightarrow m_J \neq 0 \rightarrow 0$   
 $^{87}\text{Sr}^+$  —  $m_F = 0 \rightarrow 0$  but  $I = 9/2 \dots$

NRC, NPL  
 LNPL

$^{137/138}\text{Ba}^+$  — clock  $\lambda$  in the infrared

$^{199}\text{Hg}^+$  —  $\lambda$ 's hard, cryotrap — but NIST currently world-leading

# Alkali-like ions – ytterbium



# $^{171}\text{Yb}^+ \ ^2\text{S}_{1/2} (F=0) - ^2\text{F}_{7/2} (F=3, m_F=0)$ : projected shifts

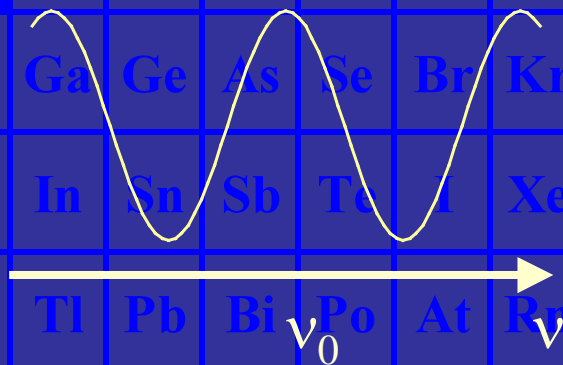
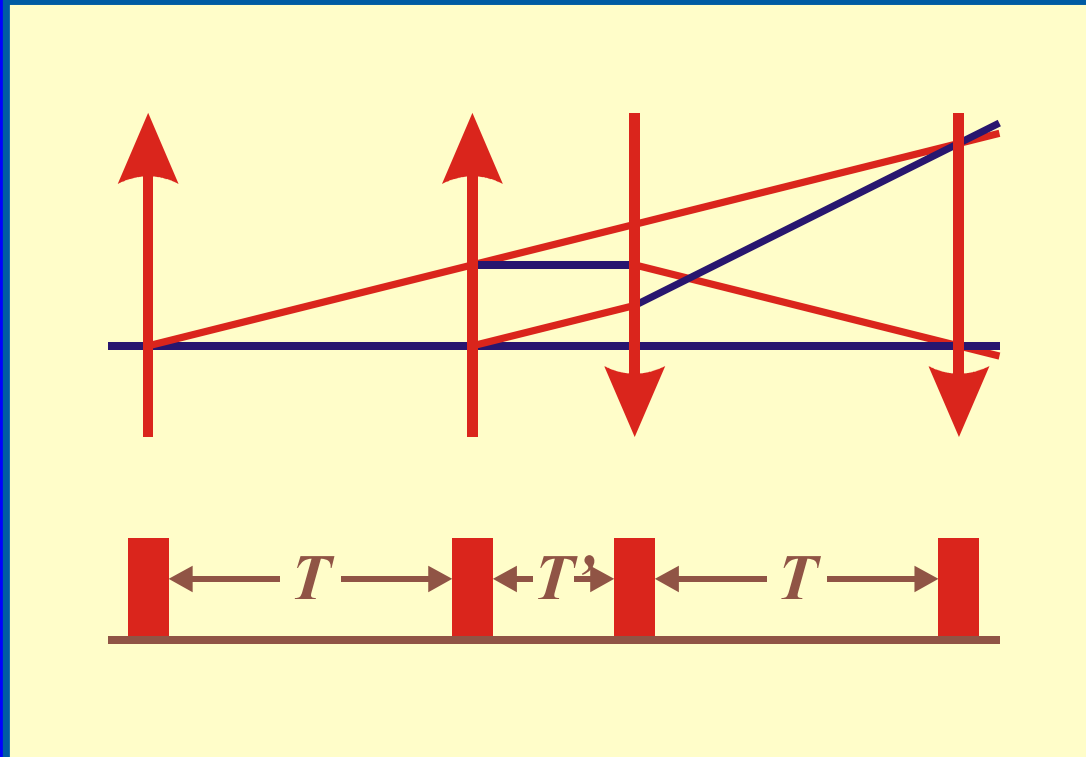
H adapted from Blythe et al., <i>J. Phys. B</i> <b>36</b> (2003) 981 – NPL He																	
Li	Be											B	C	N	O	F	Ne
Systematic shift											$\Delta(\text{mHz})$		$u(\text{mHz})$				
Na	Mg											Al	Si	P	S	Cl	Ar
2nd-order Doppler											0.0		0.3				
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Gravitational potential ( $u_h = 1 \text{ m}$ )													64				
Rb	Sr	Y	Zr	Nb	Mo	Tc	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
Quadrupole ( $u_{V_{dc}=0} = 0.2\text{V}$ )											$\pm 12$		10				
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
dc Stark											-0.3		0.1				
ac Stark (probe)											10		0.2				
Fr	Ra	Ac															
ac Stark (blackbody)											-90		30				
2nd-order Zeeman											-0.0024		0.0003				
2nd-order Zeeman (blackbody)											-13		4				
Total											-93		72				

# Cold atom interferometry

$\sim 10^7$  atoms cooled to  $\sim$  few mK in MOT;  
then free expansion of atom cloud

Ramsey-Bordé interferometer

$\Rightarrow$  Ramsey fringes



Stabilise to half-height  
either side of central  
Ramsey Fringe

Phase uniformity of laser wavefronts  $\Rightarrow$  Doppler shifts



# $^{40}\text{Ca } ^1\text{S}_0 - ^3\text{P}_1$ : projected shifts

from Wilpers et al., *PRL* **89** (2002) 230801  
and Helmke (private communication) – PTB

	Systematic shift		T = 3 mK (Hz)		6 $\mu\text{K}$ (Hz)	
	residual 1st-order Doppler		2.6		0.15	
	2nd-order Doppler		0.005		0.000025	
	lineshape asymmetry		0.05		0.05	
	other phase contribution		4		0.10	
	magnetic field [60 Hz/(mT <sup>2</sup> )]		0.1		0.08	
	quadratic Stark		0.06		0.02	
	blackbody radiation		4.3		0.05	
	servo electronics		3.2		0.10	
	cold atom collisions		1.8		0.26	
	Total		7.4		0.35	



# Stability

## Allan variance

fractional stability as function of averaging time  $\tau$

$$\sigma_y(\tau) = \left\langle \frac{\Delta f_{\text{rms}}}{f_0} \right\rangle_{\tau} \approx \frac{\Delta f}{\pi f_0} \sqrt{\frac{T}{N\tau}}$$

$\Delta f_{\text{rms}}$  – measured frequency fluctuation

$N$  – number of atoms       $T$  – cycle time, with  $\tau > T$

Quantum projection noise: statistical fluctuations in  
– number of detected atoms (atom interferometer)

– number of detected quantum jumps (ions)  
in cycle time  $T$ .

e.g. single ion,  $N = 1$ ,  $T \approx 1/\Gamma \sim 1$  s,  $Q \sim 10^{15} \Rightarrow 10^{-16} \tau^{-1/2}$

(calcium) atoms,  $N \sim 10^7$ ,  $T \sim 10^{-3}$ ,  $Q \sim 10^{12} \Rightarrow 10^{-18} \tau^{-1/2}$

# Hydrogen – the simplest atom?

**H**

Possibility of a true quantum standard? Second  $\Leftrightarrow R_\infty$

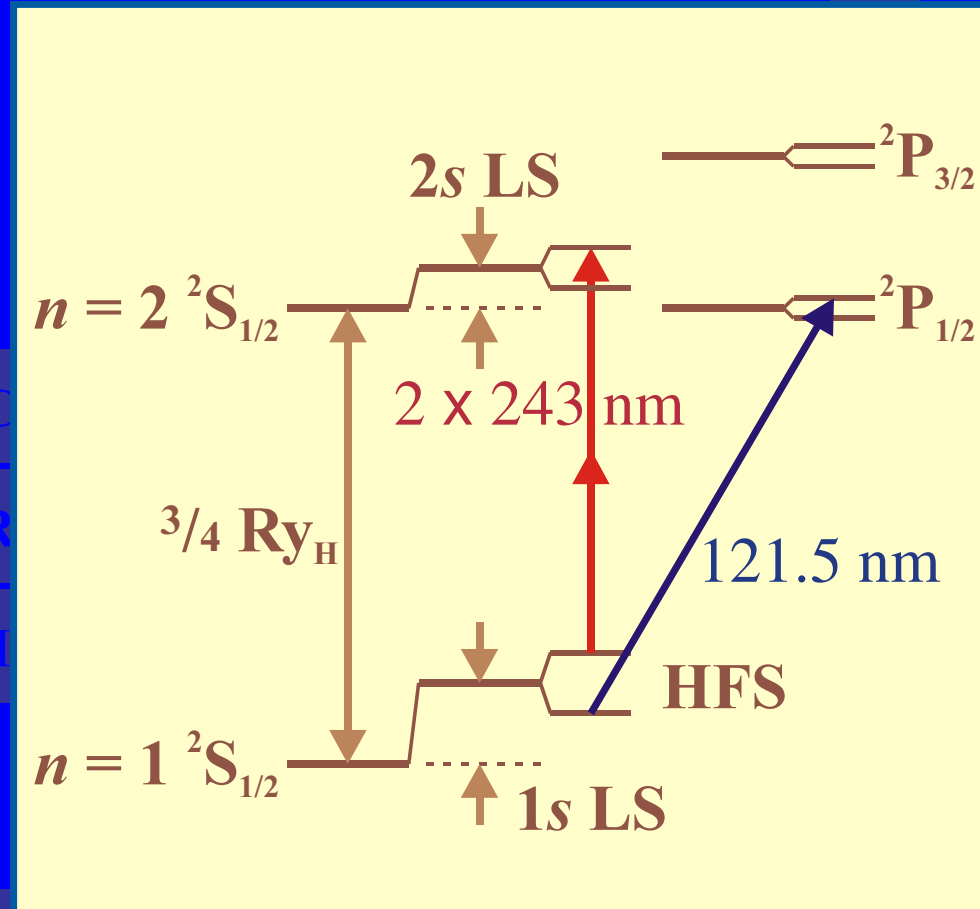
But:

Reduced mass  $m_e/m_p$

Lamb shift – QED

HFS

Proton charge radius



Laser cooling at 121.5 nm – hard, but considerable effort underway for hydrogen/antihydrogen comparison

Other 2-photon transitions – e.g. Ag – atomic fountain?

# Summary

- alkali-like ions:

H																		He
Li	Be	<sup>88</sup> Sr <sup>+</sup>										B	C	N	O	F		Ne
Na	Mg	<sup>199</sup> Hg <sup>+</sup>																
		<sup>171</sup> Yb <sup>+</sup>																
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br		Kr

- alkaline earth atoms:

Rb	Sr	<sup>20</sup> Mg																
		<sup>40</sup> Ca																
Cs	Ba	<sup>88</sup> Sr																
		<sup>174</sup> Yb																

- alkaline earth-like ions:

		<sup>115</sup> In <sup>+</sup>																

- alkali-like atoms:

		<sup>105</sup> Ag																
		<sup>1</sup> H																

NRC, NPL

NPL

NIST

PTB (S<sub>1/2</sub>-D<sub>3/2</sub>)

NPL (S<sub>1/2</sub>-F<sub>7/2</sub>)

U. Hannover

NIST, PTB, U. Hamburg ...

JILA, Tokyo, BNM-SYRTE, PTB...

various

MPQ, Washington State

MPQ, BNM-INM

beam: MPQ, U. Paris VI

cooling: MIT

# The ideal atom/ion?

H																	He
Li	Be										B	C	N	O	F		Ne
Na	Mg										Al	Si	P	S	Cl		Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac															

About 10 – 12 systems currently studied

- all have some advantages
- all have some disadvantages

⇒ The ideal atom/ion has yet to be created!

But fs combs may enable one to have the advantages of two or more systems

e.g. use atom standard for stability, ion standard for accuracy

⇒ More research required ...

Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr